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ABSTRACT

Far-infrared spectra of a germanium-antimony-tellurium alloy in the glassy and crystalline states have been measured and analyzed in the frequency range 20-400 cm⁻¹ at room temperature. The absorption in this range is due to the phonon modes of the structural units of crystalline and correlated librational vibrations (boson peak) of glassy alloy, which precede the appearance of relaxation dynamics. The vibrational assignments of various absorption bands and the differences revealed in the spectra will make it possible to more confidently elucidate the possible molecular mechanism of the crystal-to-amorphous transition in other chalcogenide alloys. New details of the crystal-toamorphous transition scenario are suggested.

Far-Infrared Spectra of the Alloy of

Germanium-Antimony-Tellurium

in the Glassy and Crystalline State

Original Research Article

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Keywords: far-infrared spectra, phonons, boson peak, chalcogenide, phase transition.

1. INTRODUCTION 12

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14 Phase-change materials (PCMs) based on chalcogenide alloys are of great technological importance 15 due to their ability to undergo on being heated a fast and reversible transition between the amorphous 16 and crystalline phases [1, 2]. This property is exploited in rewritable optical media and electronic 17 nonvolatile memory, which are based on the strong optical and electronic contrast between the two 18 phases [3].

Te-containing chalcogenides such as the Ge-Sb-Te ternary alloy system usually abbreviated as GST 19 20 has been known to satisfy material requirements for PCMs. The system is already in use for the 21 rewritable compact discs and digital versatile discs and their commercial application in electronic non-22 volatile memory devices is expected in the near future as well. However, neither the local structure of 23 material nor the change in the structure during the phase transition is well established. Several details 24 of the GST electronic structure, supposed to control the transport properties and the switching 25 mechanism, are still a matter of debate. In some reports, the strong optical and electronic contrast 26 between the amorphous and crystalline phases GST system is attributed to a change in the 27 coordination numbers [4], and in others, to a change of bonding upon crystallization [5] or to a 28

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30 high concentration of defects [6] and Peierls-distortion in an octahedral environment of Ge atoms [7].

31 At present, the most advanced techniques, such as the extended x-ray absorption fine structure 32 (EXAFS) spectroscopy and spectroscopy of neutron scattering and Raman scattering (RS) are 33 employedin studies of the local atomic structure and dynamics of GST. Interestingly, the infrared (IR) 34 spectroscopy is rarely used [8], and no IR absorption spectra of GST system are available for the 35 frequency range of RS-spectrum, on which conclusions about the molecular mechanism of the phase 36

The IR absorption in solids can provide useful information about the lattice vibration density and 37 38 structure of solids. The vibrations that contribute to IR absorption yield the most direct structural 39 information because their frequencies are determined primarily by the nearest interactions, and 40 thus,the relative IR activity are governed by local molecular symmetry. IR spectroscopy is one of the 41 most frequently used spectroscopic tools for the study of the crystals and amorphous materials 42 (including chalcogenide) [9]. Its application for studying the structure of short- range order and the 43 dynamics of individual structural units of GST would be guite natural. This is especially so at low 44 frequencies, where manifest multiphonon absorption in groups of the short- range order, lattice absorption in the crystal, and disorder-induced absorption in glasses ("boson peak" by terminology of 45 46 RS-spectroscopy [10]) is manifested.

47 However, it is only known about the optical properties of the GST system that these materials have 48 high transparency in the infrared range, from the electronicabsorption edge up to the multiphonon 49 edge near 20 micrometers. However, spectrophotometry has been recently used in the mid-IR region

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50 to measure the transmittance of selected materials from this system [11]. The experimental data were 51 successfully interpreted in terms of Drude-type plasma of free carriers and it was found that the main increase of optical constants occurs in the far-infrared domain. Consequently, the far-infrared 52 53 spectroscopy appears to be convenient experimental method to study directly the structure and 54 dynamics of the GSTsystem. This is also evidenced by the results of the comparative study and 55 analysis of the terahertz (THz) time-domain spectra of four different GST compounds, in whose 56 spectra a phonon mode and a contribution of free charge carriers at range of 10-80 cm⁻¹ [12] are 57 observed.

58 The aim of the present study was to obtain IR transmission spectra of polycrystalline and glassy 59 material of a GST system containing 15 at % germanium, 15 at % antimony, and 70 at % tellurium (henceforth Ge₁₅Sb₁₅Te₇₀) in the region 20-400 cm⁻¹, to analyze these spectra on the basis of 60 61 calculation and published data, and to find the relationship between the parameters of the far-IR 62 spectra and the molecular characteristics of Ge₁₅Sb₁₅Te₇₀. The results are interpreted in terms of 63 vibrations of isolated molecular units of the alloy under study. A detailed comparison of the spectra demonstrated that glassy Ge15Sb15Te70 exhibits characteristic extreme far-IR absorption as result of 64 phonon coupling to modes which are not active in the corresponding crystalline material. The 65 66 vibrational assignments of various absorption bands and the differences revealed in the spectra will 67 make it possible tomore confidently specify the possible molecular mechanism of the crystal-to-68 amorphous transition in GST alloys. 69

70 2. MATERIAL AND METHODS / EXPERIMENTAL DETAILS /

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72 The melt-quench technique with elemental Ge, Sb and Te of 5N purity was used to 73 prepareGe₁₅Sb₁₅Te₇₀ glassy samples [13]. The materials were weighed according to their atomic 74 weight percentages and sealed in evacuated (~10⁻³ Pa) quartz ampoules. The ampoules were kept in 75 a furnace the temperature of which was raised up to 950 °C at a heating rate of 3-4 °C min⁻¹. The 76 ampoules were frequently rocked for 20 h to make the melt homogeneous. Then the melts were 77 rapidly guenched outside the furnace by dropping the ampoules in liquid nitrogen for fast cooling and 78 to ensure the amorphization of the alloy. The alloys so obtained were crushed to the powder form for analysis. The nature of the thus prepared sampleswas confirmed by the X-ray diffraction patterns 79 obtained with an Rigaku –D/MAX diffractometer with Cu K α radiation (λ = 0.15406 nm). Figure 1 80 shows the XRD patterns for rapidly quenched and thermally annealed at 200 ⁰C alloys. XRD pattern 81 82 of rapidly guenched samples reveals the amorphous nature of the alloy as she do not show any spectacular peak while XRD pattern of samples annealed at 200 °C show the polycrystalline nature 83 84





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Figure.1. X-ray diffraction spectra of GST alloy composition Ge₁₅Sb₁₅Te₇₀.
1. X-ray scan of alloy quenched in liquid nitrogen.
2. X-ray scan of thermally annealed alloy at 200^oC.

Bulk samples were characterized using energy dispersive X-ray spectroscopy (EDAX)(Zeiss EVO 40
 EP with EDAX attachment operated at 20 kV) for theanalysis of compositions.EDAX results indicate

92 that the atomic percentages of Ge₁₅Sb₁₅Te₇₀ compositions are close to starting elements (Ge-15.2, Sb- 15.1, Te - 70.3 at. %, with accuracy of ±5%. To obtain polycrystalline samples, the bulk samples 93 94 were annealed in inert atmosphere for 120 min at temperature about 20^oC above the temperature of 95 amorphous - fcc phase transition determined by differential scanning calorimetry. Differential scanning 96 calorimetry (DSC-50, Shimadzu) at heating rate of 5 ℃/min in a nitrogen flow was used toexamine the 97 thermal properties of investigated materials. Polycrystalline samples were crushed and pressed into 98 Al pans. The temperature range studied was 20 to 630 °C. The necessary amount of amorphous 99 material (about 2-4 mg).

100 The far-IR absorption measurements were made in spectral range 20-400 cm⁻¹ at room temperature. 101 Far-IR spectra were obtained on two spectrometers. A spectrometer designed at the Leningrad State University [14]and then modified with an OAP-7 and a new filtration system was used in the frequency 102 range from 20 to 150 cm⁻¹. The spectra in the range 150 to 400 cm⁻¹ were recorded with Hitachi FIS-103 104 21 (Japan) spectrometer. The resolution at a signal-to-noise ratio of the order of 100 was 1 to 2 cm⁻¹. 105 The peak positions of the spectral bands were determined to within 1 - 2 cm⁻¹. All the measurements were madeby using the polyethylene pellet method. The glassy and polycrystalline samples were 106 thoroughly ground to an average size of 10-15 µm in dry argon to preclude surface oxidation. The 107 pellets were made by mixing 2-mg portion of a sample with 200 mg of spectroscopic grade 108 109 polyethylene. The mixture was compacted into pellets with a hydraulic press (10 tons cm⁻² for 5 min under vacuum). To take into account the absorption of polyethylene, it was used as reference. The 110 spectrum of a sample was divided by the reference spectrum to eliminate the absorption of 111 112 polyethylene.

114 3. RESULTS AND DISCUSSION

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The figure2 shows the far-IR transmission spectra of polycrystalline and glassy GST material of composition $Ge_{15}Sb_{15}Te_{70}$ in the region 20 - 400 cm⁻¹. The recorded spectra have the form of an almost structureless plateau, with several peaks of the overlapping absorption bands seen on its background. The most distinct of these are the bands at ~ 60 cm⁻¹, 94 cm⁻¹, 110 cm⁻¹, 129 cm⁻¹, 159 cm⁻¹, and 217 cm⁻¹ in the spectrum of glassy and at 45 cm⁻¹, 56 cm⁻¹, 65 cm⁻¹, 92 cm⁻¹, 109 cm⁻¹, 127 cm⁻¹, 157cm⁻¹, and 214 cm⁻¹ in the spectrum of polycrystalline material.



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123 . Figure 2. Far-infrared spectra of GST system composition $Ge_{15}Sb_{15}Te_{70}$ in the 124 amorphous (a) and the crystalline (b) state.

125 Thesebands were tentative assigned on the basis of calculations in which the GTS alloy is regarded 126 as a 3D polymer. Its structure was expressed in the form of the two covalently bound structural units: 127 tetrahedral "molecules" of germanium tetratelluride (GeTe₄) in regions with a short-range order and 128 pyramidal "molecules" of antimony tritellurideSbTe₃. The validity of this approach follows from the 129 close relationship between the vibrational spectra of molecules or molecular ions and those of glass-130 like solids containing the same or analogous structural groups in the form of regionswith a short-range 131 order.

132 The GeTe₄ molecule has 5 atoms and thus supports 15 modes of motion, including 3 pure rotations 133 and 3 pure translations. The T_d symmetry of the GeTe₄ molecule leads to 6 zone-center modes 134 following the decomposition in standard notation [15]:

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$$\Gamma(T_d) = A1 + E + F_1 + 3F_2$$
,

where F_1 corresponds to a pure rotation (noted as R in the literature), and one of the F_2 is a pure translation (designated as T). R and T modes are related to the external modes. The modes derived from the tetrahedral A_1 , E and $2F_2$ modes are usually named as v1, v2, v3 and v4, respectively, and are regarded as internal modes of the GeTe₄ tetrahedra in Ge₁₅Sb₁₅Te₇₀.

The modes which predicts a group theory for a tetrahedral symmetry and the calculated phonon frequencies at the $\Gamma(T_d)$ -point are listed in Table 1. They are attributed as shown in Table 1 below in accordance with the assignments of the germanium tetraiodide (Gel₄)molecular vibrations [16, 17].

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144 Table 1.Normal vibrational modes of the Gel₄ and GeTe₄ tetrahedra.

145 From the spectrum of a Ge-Sb-Te alloy composition Ge₁₅Sb₁₅Te₇₀.

	Normal vibrational mode (cm ⁻¹) [*]									
substance	v1	v2	v3	v4	vR	vT				
Gel ₄	159	60	264	80	56 [19]	-				
GeTe₄	125	87	209	78	60	55				
$Ge_{15}Sb_{15}Te_{70}$	127	92	214	80	65	60				

146 v1 and v3 – the stretching modes for symmetric and antisymmetric vibrations, respectively;

147 v2 and v4 – bending modes for symmetric and antisymmetric vibrations, respectively;

148 vR and vT- external modes for pure rotation and pure translation, respectively.

The analogy between the modes of the $GeTe_4$ and the GeI_4 tetrahedra can be used because the mass ratio of germanium tetraiodide is very similar to germanium tetratelluride, and, therefore, a correspondence in the respective frequencies can be expected.

Although only the v3 and v4 modes are IR-active in tetrahedral GeX₄ molecules, it is expected that all the fundamental modes appear in IR absorption spectra because of the violation of the selection rules in polycrystalline and glassy materials. In addition, the structure of crystalline GST is only "rock salt like" [1].

156 The SbTe₃ molecule crystallizes in a structure with symmetry C_{3v} . There are four atoms per unit cell, and, consequently, SbTe₃ exhibits 12 phonon modes, including external R and T modes for each 157 wavevector K. A group-theoretical analysis yields for K = 0 the following reduction: Γ_{tot} (C_{3v}) = 2A_{1g} 158 +2A_{1u} + 2E_g + 2E_u, which splits into acoustical modes and IR- orRaman-active optical modes as follows: $\Gamma_{acoustical} = A_{1u} + E_u$, $\Gamma_{Raman} = 2A_{1g} + 2E_g$ and $\Gamma_{IR} = A_{1u} + E_u$, where one of the A_{1u} corresponds to a pure rotation(designated as vR here) and one of the E_u is a pure translation(designated as vT here). 159 160 161 162 Following the spectroscopic notation, the stretching modes are designated by v1 and v3 for the totally 163 symmetric and the antisymmetric (double-degenerate) vibrations, respectively. Analogously, the 164 bending modes correspond to v2 (A1) and v4 (E). All four vibrations are both infrared- and Raman-165 active.

166 The modes predicted by the group theory for a pyramidal symmetry and the calculated phonon 167 frequencies at the Γ_{tot} (C_{3v})-point are listed in Table 2. They are attributed as shown in Table 2 below 168 in accordance with the assignments of the of the antimony triiodide (Sbl_3) molecular vibrations [18].

169 The IR band assignments given here can be further substantiated by comparing the ratios between

170

171 Table 2.Normal vibrational modes of the Sbl₃ and SbTe₃ pyramids.

substance	v1	v2	v3	v 4	vR	vT	v3/v1	v4/v2
Sbl ₃	177	89	147	71	67,5	45,5	0.83	0.80
SbTe ₃	164	122	145	106	53	46	0.88	0.86
$Ge_{15}Sb_{15}Te_{70}$	157	127	142	109	56	45	0.90	0.87

172 From the spectrum of a Ge-Sb-Te alloy composition Ge₁₅Sb₁₅Te₇₀.

173 v1 and v3 – the stretching modes for symmetric and antisymmetric vibrations, respectively;

174 v2 and v4 – bending modes for symmetric and antisymmetric vibrations, respectively;

175 vR and vT- external modes for pure rotation and pure translation, respectively.

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the experimental frequencies of the stretching modes (v3/v1) and of the bending (v4/v2) modes for Sbl₃ with similar values obtained in calculating the phonon modes for SbTe₃. The important aspect of this comparison is that the frequency ratios v3/v1 and v4/v2 are essentially the same for the Sbl₃ and SbTe₃.

181 The calculated assignments of the IR bands in the spectrum of the crystalline Ge15 Sb15 Te70, 182 presented in the Tables 1 and 2 are qualitatively comparable with assignment of Raman frequencies 183 of GST materials, reported in literature [19, 20]. So, main Raman bands peaked at 127 cm⁻¹ and 157 cm⁻¹ in cubic Ge₂₂Sb₂₂Te₅₆ were also associated with stretching modes of GeTe₄ tetrahedra and 184 SbTe₃ pyramids, respectively. The theoretical calculations made in the study show that bands peaked 185 at 127 cm⁻¹ and 157 cm⁻¹ have also a contribution of $v2(A_{1u})$ and $v1(E_u)$ modes of SbTe₃, respectively. 186 The lower frequency band, covering 85-115 cm⁻¹ frequency region of the far-IR spectrum of 187 188 Ge₁₅Sb₁₅Te₇₀ is comparable in intensity with the peaks of stretching modes GeTe₄ and SbTe₃ and is 189 formed by at least two overlapping bands at 92 cm⁻¹ and 110 cm⁻¹. According to the calculations, these 190 can be attributed to bending modes v2 (E) and v4 (E_u) of the GeTe₄ tetrahedra and SbTe₃ pyramids, 191 respectively. In the frequency range 180 - 330 cm⁻¹, at least four small bands peaked at ~214, 260, 300, and 320 cm⁻¹ can be distinguished in the far-IR spectrum of crystalline Ge₁₅Sb₁₅Te₇₀ (see Figure 192 2). The first of these, lying at 214 cm⁻¹, can apparently be assigned to v3 (F_2) asymmetric stretching modes of GeTe₄ tetrahedra. An analogous feature was observed in Raman spectra of bulk crystalline 193 194 195 GeTe, which are isostructural to $Ge_{22}Sb_{22}Te_{56}$ [21]. The weak bands at 260 and 320 cm⁻¹ are quite probably due to the so-called 2 - phonon vibrational absorption arising from excitations of fundamental 196 molecular stretching modes of $GeTe_4$ (at 127cm⁻¹) and SbTe₃ (at 157cm⁻¹), respectively; while absorption band near 300 cm⁻¹ can be attributed to vibrations of Ge-Ge ethane-like clusters [22]. The 197 198 absorption bands at the frequencies 95 cm⁻¹ and 143 cm⁻¹, which were assigned to the modes v2(E)199 200 and v3(E) of the GeTe₄ and SbTe₃, respectively, in the far-IR spectrum of the Te-rich Ge₁₅Sb₁₅Te₇₀ 201 might also contain the contribution of the fragments consisting only of tellurium atoms [23]. 202 The proposed interpretation of the IR-spectrum of polycrystalline Ge₁₅Sb₁₅Te₇₀ can be used for the attribution of the absorption bands at v > 80 cm⁻¹ of the IR-spectrum of the samealloy in amorphous 203 204 (glassy) state. This is so because experimental spectra of these states practically do not differ from 205 each other in range in which intramolecular modes of the structural units of Ge₁₅Sb₁₅Te₇₀, are

206 manifested, except for a negligible red shift and changes in the shape of bands in the IR-spectrum of 207 polycrystalline $Ge_{15}Sb_{15}Te_{70}$ with respect to that of the amorphous state. Because, as a rule, the 208 crystalline phase is more ordered than the amorphous one, this fact shows unusual similarity of the 209 amorphous and crystalline structures $Ge_{15}Sb_{15}Te_{70}$ of short range order and leads to an assumption of rather similar degree of topological 'disordering' between these two forms. This means that the local
 structure of the building blocks (tetrahedra and pyramids) of Ge₁₅Sb₁₅Te₇₀ remains basically
 unchanged upon the phase transition from the crystalline to the amorphous phase.

At the same time, an analysis of Figure 2 readilyshows that the amorphous to crystal transition appear in the IR- spectra of $Ge_{15}Sb_{15}Te_{70}$ as drastic changes at v < 80 cm⁻¹, where the lattice absorption due to external librational and translational degrees of freedom of the structural units of crystal is manifested.

The presence of absorption peaks at 65, 56, and 45 cm⁻¹ (modes vR(F1), vT(F2) and vT(E), respectively) in the IR spectrum of polycrystalline $Ge_{15}Sb_{15}Te_{70}$ is due, according to the calculations, to the manifestation of the intermolecular librations and translations of tetrahedra and pyramids as a whole. The IR spectrum glassy $Ge_{15}Sb_{15}Te_{70}$ contains in the same frequency range only a single abnormally broad band peaked at ~ 60 cm⁻¹, which can be readily assigned as the well-known band ubiquitously present in amorphous solids, i.e., the boson peak.

223 Boson peak is attributed to the disorder-induced absorption, which is a manifestation of a low energy 224 vibrational excitations emerging from cancellation of the selection rules in an amorphous medium. The peak observed in the low-frequency spectra of glasses appeared to correspond to the peaks 225 observed at low frequencies in region of acoustic phonons of corresponding crystals [24]. Its 226 227 appearance in the low-frequency spectra of glasses at the phase transition indicates that, in the 228 absence of a long-range order, a system of coupled harmonic oscillators of crystal became a system 229 of disoriented oscillators in which only limited (on a scale of medium-range order) correlation between 230 the intermolecular vibrational motions is preserved [25]. So, the boson peak is occasionally assigned 231 to cooperative orientational motion (librations) of several molecular groups [26, 27].

That is, when IR-spectra of polycrystalline and glassy materials are compared in range in which intramolecular modes of the structural units of Ge₁₅Sb₁₅Te₇₀are manifested, there is no evidence about the contrasting local structure between the amorphous and the crystalline phases, so the appearance of the boson peak at frequencies of *intermolecular modes* suggests that the mediumrange crystalline order is also preserved in glassy Ge₁₅Sb₁₅Te₇₀. These data clearly reflect the local ordering of the atoms in the amorphous state which is similar to that in the crystalline Ge₁₅Sb₁₅Te₇₀ but lacks long-range ordering.

239 The spectral changes at low optical frequencies suggest the following scenario of a phase transition in 240 the system Ge₁₅Sb₁₅Te₇₀. Local distortions inherent "tetrahedral" and "pyramidal" structural units of 241 glassy Ge₁₅Sb₁₅Te₇₀, which facilitate the formation of Ge-Ge and Ge-Sb bonds [28], and also 242 correlated torsional oscillations (librations) of these structures with a noticeable share of vacancy form 243 medium-range (chain-like) order in the system. This is evidenced of this is the presence of the boson 244 peak in the spectrum. For the transition to occur, such a system in a metastable ordered state not 245 require a restructuring of local structures, but rather reorientation of existing clusters (the chains). In 246 essence, the order-disorder transition is in this case a relaxation transition by the fast β -relaxation 247 Johari-Goldstein mechanism [29]. Such a transition requires the minimuminput of power and has a 248 nanosecond relaxation time. 249

250 4. CONCLUSION

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In the present study the far-IR transmission spectra of polycrystalline and glassy materials of the GST system of composition Ge₁₅Sb₁₅Te₇₀were measured in the spectral range 20 - 400 cm⁻¹ at room temperature and also analyzed on the basis of theoretical calculation and of the literature data. Although the made vibrational assignments of various absorption bands given here could be considered as tentative and alternative interpretations may be proposed, these assignments and the revealed differences in the spectra is allow suggest the following possible molecular mechanism for unusual amorphization- crystallization process in GST alloys.

259 According to the majority of existing models, the presence of the boson peak, as an indication of a 260 medium-range order existing in the structure of glasses, in the low-frequency spectra is associated 261 with the correlative orientational motion (librations) of fragments a crystalline structure of glass. These 262 fragments of a crystalline structure can play the role of crystallization centers on the nanometer scale 263 and their correlative librational vibrations can be viewed as precursors of phonons due to short-range 264 ordering. Then according to this model, the recrystallization from the amorphous phase may be 265 achieved easily by reorientation of the randomly oriented fragments of a crystalline structure back to 266 the original well-aligned structure of the crystalline phase. We emphasize that in this case 267 crystallization of GST does not involve relaxation of the complete covalent network but is more like the so- called β-relaxation, where only fragments of the network relax. This process requires very small
 atomic displacements and hence can be extremely fast.

270 The results, by the example of the composition Ge₁₅Sb₁₅Te₇₀, demonstrate, that the use of far-IR 271 spectroscopy has a clear potential to characterize of the local atomic structure and dynamics of the 272 GST alloys. Further studies on the new GST materials will give possibility to improve the parameters 273 of the already developed memory elements and will provide additional information about the nature of 274 the switching effect.

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