Original Research Article

EFFECT OF DIFFERENT ETCHING TIME ON BOND STRENGTH OF COMPOSITE RESIN TO DENTIN

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ABSTRACT:

- 7 The aim of the study is to to evaluate the bond strength of composite resin to
- 8 dentin surface with various etching protocol. Three sample groups, consisting of
- 9 ten mandibular first molar teeth in each group were established. The buccal
- surfaces of the samples were grinded with the help of straight fissured diamond
- abrasive upto dentin surface. Group I- Samples will be etched for 15 seconds,
- Group II- Samples will be etched for 30 seconds, Group III- Samples will be
- etched for 60 seconds. All the samples will be then washed for 1 min and dried.
- Samples will be then bonded for 15 seconds and restored with composite resin
- and cured. Samples will be subjected under universal testing machine for bond
- 16 failure.

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- 17 **KEYWORDS:** Etching, adhesives, 37% phosphoric acid, universal testing
- machine, dentin adhesives.

INTRODUCTION:

- 20 Over the years, phosphoric acid etching has become the standard procedure for
- 21 bonding to improve the surface characteristics before the application of
- adhesive bonding agents and fissure sealant [1]. The penetration of adhesive
- resin monomers into the porous zone results in the formation of resin tags,
- thereby establishing micromechanical interlocking within the etched surface.
- 25 Therefore, regardless of the adhesive system, using phosphoric acid supports
- achievement of a strong and durable bond [2]. Phosphoric acid etching gel is
- applied to dentin substrate to remove the smear layer. After rinsing, the dentin
- surface becomes demineralized with exposure of the collagen fibers. This would

leave collagen fibers exposed and susceptible to hydrolysis, possibly weakening 29 the bonding [3]. Thus, it is speculated that as shallow a demineralization as 30 possible might give the adhesive system a better chance to diffuse into the entire 31 collagen network [4]. For that, acid-etching time of 15 seconds has been 32 suggested by various authors, aiming at an adequate bond to normal dentin. 33 Dentin hybridization is a modern dental adhesion procedure, which was first 34 described by Nobuo Nakabayashi et al. (1982) [5]. Hybridized dentin begins 35 under the dentin surface after surface und subsurface demineralization and 36 adhesive monomer infiltration into exposed collagen network [6]. Thus, the 37 result of the revolutionary discovery by Nakabayashi and colleagues has opened 38 new horizons of restorative dentistry. Over the years, phosphoric acid etching 39 has become the standard procedure for adhesive dentistry. In 1954 Buonocore 40 introduced acid etching procedure as a pretreatment method that enhances the 41 strength bonding of composite resins for the first time [7]. It's clinical 42 application presented in 1976 by Cueto and Buonocore. The dentin surface 43 becomes demineralized with exposure of the collagen fibers [8]. To obtain 44 adequate resin-dentin bonding, resin monomers must penetrate this 45 demineralized surface dentin in order to produce hybridization. The clinical 46 47 success of restorative material depends upon a good adhesion with dentinal surface so as to resist various dislodging forces acting within the oral cavity [9]. 48 The need for restorative material, with better bond characteristics and strength 49 to withstand the stress of masticatory forces, leads to the recent advances in the 50 restorative dentistry. The composite resins are one of the commonest restorative 51 material used now a days due to its aesthetic appearance but they have some 52 imperfections such as polymerization shrinkage [10]. 53

The present study was to to evaluate the bond strength of composite resin to dentin surface with various etching protocol.

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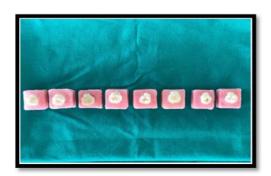
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MATERIALS AND METHOD:

30 extracted human mandibular molar teeth with neither carious lesions nor restorations, which were recently extracted for periodontal or orthodontic reason, were selected for this in vitro study. Each tooth underwent scaling and root planing with an ultrasonic device to remove residual organic tissue. Then, the teeth were immersed in 2.5% sodium hypochlorite solution and rinsed with running water for 10 min.

Acrylic blocks were prepared by cold cure acrylic resin material. The selected molars were embedded into the blocks. The blocks were then put in water to avoid expansion of the material. The buccal surfaces of the samples grinded

with the help of straight fissured diamond abrasive upto dentinal surface



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Acrylic Mold Prepared For Each Samples

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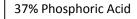
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The buccal surfaces of the samples grinded with the help of straight fissured diamond abrasive upto dentin surface

- Dentinal surfaces were acid etched with 37% phosphoric acid into following
- 73 groups:
- 74 Group A:- 15 seconds
- 75 Group B:- 30 seconds
- 76 Group C:- 60 seconds







Etching of Dentin Surface

Samples were then bonded and light cured with LED light for 20 seconds and restored with light cure composite resin. The specimens were stored in distilled water for 24hrs.



Application of Bond



Curing After Application of Bond

Specimens were then transferred to the Universal testing machine with a crosshead speed of 0.5 mm/min until fracture with tip diameter 1.5mm. Subjected to compressive test determination which created buckling of the restoration which results in formation of a tensile stress in the dentinal walls.





Specimens transferred to the UTM with a crosshead speed of 0.5 mm/min until fracture with tip diameter 1.5mm.

Universal Testing Machine

The load required to debond the specimen was recorded. Placed in the lower assembly of the machine and the force was applied with the help of a knife-like mandrel which engaged the blocks and dislodged it. Bond strength was calculated according to the following formula and expressed in kilo newton

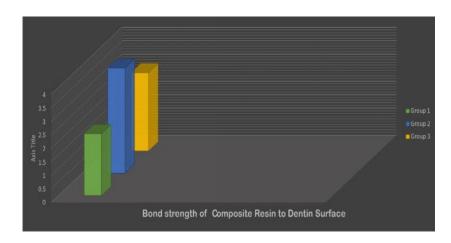


The load required to debond the specimen was recorded

101 (KN):

Stress = Failure load (N)/surface area (mm²)

RESULT & STATISTICAL ANALYSIS



Bond strength is much higher in group B than in group C and A with a confidence of more than 98.828%. (P value -0.0675)

Groups	Mean deviation	Standard deviation	Sum	Average	Variance
A	4.88	3.15	16.7543	2.347	0.858
В	8.85	0.37	21.664	3.76	1.839
С	6.63	0.73	17.663	2.66	0.676

SOURCE OF VARIATION	SS	df	MS	F	P-value	F crit
Between groups	2.33	1	2.4457	21.5674	0.0675	2.5567
Total	2.49	17				

DISSCUSION:

- Acid etching technique using phosphoric acid is well accepted for various 115 applications in dentistry. Among mechanical properties bond strength of 116 restorative materials is important because it provides sufficient strength to resist 117 intraoral compressive & tensile forces that are produced in function & 118 parafunction [11]. Different etching times with the same phosphoric acid 119 concentration result in different morphologic changes of demineralized dentin 120 surface. There is a direct correlation between etching time and the depth of 121 demineralized zone. The hybrid layer thickness correlates directly to the etching 122
- time [12]. Increased etching time demineralizes dentin surface to a depth greater
- than that to which resin monomers can penetrate, producing a thick, poorly
- infiltrated hybrid layer [13].
- The acid etching time recommended for dentin with 37% phosphoric acid gels
- commonly employed with etch and rinse system has been 30 seconds. Ustunkol
- et al., Batra et al. and Taschner et al. claimed etching process has a significant
- effect on bond strength of methacrylate-based composite Adebayo et al. showed
- higher bond strengths of the nano hybrid composite [14].
- Koliniotou-Koumpia et al, Sampaio et al. said there is no difference of bond
- strength between nano hybrid composite and bulk fill composite on etching
- 133 [15]. Dentin is heterogeneous, consisting of hydroxyapatite and collagen. The
- degree of mineral content in dentin is quite variable, depending on whether it is
- near the DEJ or deeper in close proximity to the pulp. Acidity of monomer also
- caused change in surface chemistry and morphology of dentin, which in turn
- can influence bonding. A significantly thicker hybrid layer was noted in areas
- with perpendicular tubule orientation than in areas with parallel tubule
- orientation [16].
- Mechanical behaviour depends upon the concentration and particle size of the
- inorganic filler. For evaluating the bond strength, the study samples were stored
- in distilled water with few thymol crystals, to maintain aseptic conditions before
- cavity system, with hydrophilic components, which can dislodge moisture from
- the conditioned dentin and attain an intimate interaction at the demineralized
- intertubular and peritubular dentin, creating the hybrid layer, which is essential
- for an ideal bond to dentin. The studies conducted by Kallenos et al. and Gupta
- 147 et al [17].

- The present study showed that 30 seconds acid etching with 37% phosphoric
- acid gave better bond strength than 60 seconds and 15 seconds. Hence these
- findings confirmed that the different etching times with the same phosphoric
- acid concentration result in different morphologic changes of demineralized
- dentin surface. This was very evident in the striking changes in the number,
- diameter and surface area of dentinal tubules, intertubular surface area,
- appearance of the dentin surface porous zone containing smear layer and
- demineralized residual collagen particles with dentin demineralization products
- in acid globules, and the completely dissolved peritubular dentin cuff that
- happened after prolonged etching time [18].

CONCLUSION:

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- 1. When 37% phosphoric acid is applied a dentin substrate free of smear layer is not created for the etching time of 15 seconds.
- 2. Efficiently removed smear layer and no precipitates were observed in teeth after 30 seconds etching with 37% phosphoric acid.
 - 3. Within the limitations of this in vitro study; it can be concluded that-30seconds of etching time gives better bond strength due to longitudinal tubules which influences the dentin hybridization process in the sense of forming more resin tags that, in turn, contribute to bond strength.

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UNDER PEER REVIEW

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