

# Original Research Article

## Formulation of lubricant from calabash seed oil

### ABSTRACT

The engine oil industry is faced with complex chemical reactions and difficult chemical engineering involved in the formulation of synthetic based engine oils, and therefore, the need to source for alternative base oils for engine oil formulation from vegetable oils has become urgent and inevitable. This research is aimed at formulation of lubricant using calabash seed oil (CSO). An experiment designed (Mixture Design Method using Minitab 17) was used to obtain the blend of CSO (28.75%), SN 500 (68.75%), and additive (2.50%) with improved physicochemical parameters. The lubricant obtained had kinematic viscosities  $9.30 \pm 2.11$  cSt (at  $100^\circ\text{C}$ ) and  $53.11 \pm 1.03$  cSt (at  $40^\circ\text{C}$ ), a viscosity index of  $167 \pm 0.51$ , flash point of  $240 \pm 2.01^\circ\text{C}$ , and pour point of  $-28 \pm 1.31^\circ\text{C}$ . The lubricant obtained in this research had quality parameters that are comparable to those of synthesised environmentally acceptable engine oils, and are within the standard for engine oils.

*Keywords: Formulation, Environment, Lubricant, Calabash Seed Oil*

### 1. INTRODUCTION

The engine oil industry is faced with numerous challenge such as renewability of the base oils [1], availability and accessibility of the products, compatibility of the products with modern machines and equipment, and environmental acceptability of the products.

Leaders in the engine oil industry have accepted the challenges and are formulating specialised products in conformity with the aforementioned challenge but the formulation processes are associated with costs because of the complex chemical reactions and difficult chemical engineering involved in the production of synthetic based engine oils [2].

The use of vegetable oils for the production of environmetally acceptable engine oils is a promising development in addressing the drawbacks associated with the use of conventional mineral oils, the complex methods employed in the formulation of synthetic engine oils, and additionally, economic relevance in the local communities where vegetable oil crops are grown and processed [1,3-7].

This study is aimed at using calabash seed oil (CSO) for the formulation of lubricating engine oil.

### 2. METHODS

#### 2.1 Sample Collection and Treatment

Calabash seeds were obtained from Sokoto Metropolis, Sokoto State, North-western Nigeria, and authenticated by Botany Unit, Department of Biological Science, Faculty of

41 Sciences, Usmanu Danfodiyo University, Sokoto. The seeds were dehulled, dried, ground  
42 into powder, and sieved to obtain a homogeneous powder. The powdered calabash seed  
43 (300.00 g) was weighed and preserved for oil extraction [8,9].

44

## 45 **2.2 Oil Extraction**

46

47 Soxhlet extractor was employed in the extraction of calabash seed oil (CSO) with n-hexane  
48 (6:1 w/w% of solvent/sample) as the extracting solvent (in 500 cm<sup>3</sup> round-bottomed flask).  
49 The sample (50.00 g per each extraction) was heated at 60°C for 5 hours, for complete  
50 extraction [4,9]. The oil was separated from the solvent with the aid of a rotary evaporator.  
51 The percentage yield of the CSO was calculated using equation 1:

52

$$53 \quad \% \text{ Yield} = \text{Weight of Oil}(g) \times 100 / \text{Weight of the Sample}(g) \quad \dots\dots\dots 1$$

54

55

## 56 **2.3 Lubricant Formulation**

57

58 In order to achieve a better thermal and oxidative stability for the CSO [5,10], the CSO was  
59 blended with SN 500 mineral based oil and additive (poly alkylmethacrylate). A design of  
60 experiment (Mixture Design method of Minitab 17) was used to obtain the best blend with  
61 improved quality parameters from the raw CSO extract. The blend of CSO, SN 500, and  
62 additive was done in a conical flask at a temperature of 45°C, and stirred at 600 rpm for 15  
63 minutes, while a heating mantle equipped with a magnetic stirrer was used to achieve a  
64 homogeneous mixture [5,9].

65

66

67

68

## 69 **2.4 Physicochemical Parameters of CSO and the Lubricant**

70

71

### 72 **2.4.1 Determination of Kinematic Viscosity**

73

74 The oils were each poured into a viscometer tube and mounted upright in the viscometric  
75 bath which was maintained at 40 or 100°C. The oil in the tube was allowed to gain  
76 equilibrium for 15 minutes. When the equilibrium temperature was achieved, the oil level in  
77 the viscometer tube was adjusted using a suction pump to 7 mm above the upper mark of  
78 the viscometer tube. The time (*t*) taken for the oil to move from the upper mark to the lower  
79 mark of the viscometer tube was recorded [9]. The kinematic viscosity (KV) was obtained  
80 via equation 2:

81

$$82 \quad KV (cSt) = C (cSt s^{-1}) \times t (s) \quad \dots\dots\dots 2$$

83

84

85 Where, **KV** is the kinematic viscosity; **C** is the calibration constant of the viscometer;  
86 **t** is the time.

### 87 **2.4.2. Determination of Viscosity Index**

88

89 Viscosity index (VI) of the oils were each obtained using values of kinematic viscosity  
90 obtained at 40 and 100°C with standard measurement table as determined by ASTM D-2270  
91 method.

92 **2.4.3. Determination of Pour Point**

93  
94 Pour point tester of accuracy of  $\pm 3^{\circ}\text{C}$  was used to determine the pour points of the CSO and  
95 lubricating oil. The tester used methanol as the cooling solvent and has a minimum  
96 temperature of  $-68^{\circ}\text{C}$ . The oil ( $45\text{ cm}^3$ ) was poured into a test jar to the levelled mark. Then  
97 the tester was cooled to  $-37^{\circ}\text{C}$ . While cooling the tester, the oil jar was heated to  $45^{\circ}\text{C}$  with  
98 the aid of a water bath. The oil jar was cooled with another water bath to a temperature of  
99  $27^{\circ}\text{C}$ . When the pour point tester had reached  $-36^{\circ}\text{C}$ , the oil jar was placed in a horizontal  
100 position in the hole at the top of the tester and the pour point temperature was taken after 5  
101 seconds when the oil showed no movement [5].

102 **2.4.4. Determination of Flash Point**

103  
104 The flash point of the CSO and lubricating oil were each determined by heating a cup  
105 holding the oil and moving a flame over the oil at regular temperature, starting with a  
106 temperature of  $28^{\circ}\text{C}$  below the expected flash point of the oil. The bulb of the thermometer  
107 was immersed in the sample in order to allow monitoring and reading of the temperature at  
108 flash point. The flash occurred in the cup containing the CSO when the temperature of the oil  
109 had reached its flash point [9].  
110  
111

112 **2.4.5 Determination of Free Fatty Acid**

113  
114 The oil (2.00 g each of CSO and lubricating oil) was placed in a dry  $250\text{ cm}^3$  -conical flask.  
115  $50\text{ cm}^3$  of ethanol and few drops (2-3) of phenolphthalein indicator were added. The mixture  
116 was heated at  $60^{\circ}\text{C}$  in a water bath for 10 minutes and then cooled. The mixture was titrated  
117 with 0.1 M KOH to the endpoint (with consistent shaking). A dark pink colour was observed  
118 and the volume of KOH used for the titration was recorded as the titre value [5]. The acid  
119 value and the free fatty acid value were calculated using equation 3 and 4:  
120

121 Acid value  $\left(\frac{\text{mgKOH}}{\text{g sample}}\right) = \frac{\text{Volume KOH}(\text{cm}^3) \times \text{N KOH}(\text{mmol/cm}^3) \times 56.1 (\text{mg/mmol})}{\text{sample weight (g)}} \dots\dots\dots 3$

122  
123  
124 Free Fatty Acid  $\left(\frac{\text{mgKOH}}{\text{g sample}}\right) = 0.5 \times \text{Acid value} \dots\dots\dots 4$   
125

126 Where, **KOH** is potassium hydroxide; **N** is the molar concentration of KOH; and  
127 **56.10** is the molecular weight of KOH  
128  
129

130 **3. RESULTS AND DISCUSSION**

131  
132 The formulated CSO lubricant with optimum physicochemical properties was obtained from  
133 the proportion with details as CSO (28.75% wt), SN 500 (68.75% wt), and additive (2.50%  
134 wt) (Table 1).  
135 An index for evaluating the internal resistance in the motion of engine oil is kinematic  
136 viscosity [11]. Kinematic viscosities of the lubricant were found to be  $53.11 \pm 1.03$  cSt at  $40^{\circ}\text{C}$   
137 and  $9.30 \pm 2.11$  cSt at  $100^{\circ}\text{C}$  (Table 2) which are lower compared to the values for

138 synthesised oil (Appendix I) but are within the standard for engine oils [9]. Thus, the lubricant  
139 is useful for engine oil application.

140 The effect of changing temperatures on the kinematic viscosity of lubricating fluid is called  
141 viscosity index (VI) and it is inversely proportional to temperature: a higher change in  
142 viscosity in response to temperature variation indicates small viscosity index [11,12]. The  
143 viscosity index of the lubricant was found to be  $167 \pm 0.51$  (Table 2) which is comparable to  
144 the available commercial lubricating oil. This shows that the lubricant will experience less  
145 change in its viscosities as a result of variations in temperatures during applications.

146 The value of flash point of the lubricant was found to be  $240 \pm 2.0$  °C (Table 2). Flash point is  
147 a useful lubricating oil property which suggests the minimum temperature at which the oil  
148 generates ignitable vapour [13]. The high flash point of the produced lubricant suggests a  
149 higher and complex nature of its molecular structure, and hence, has low risk associated  
150 with vapourisation during transport and storage.

151 However, the pour point of the lubricant was found to be  $-28 \pm 1.31$  °C (Table 2). According to  
152 Gobinda et al. [14], vegetable oil base stocks solidify at low temperatures making them less  
153 useful for some applications. Though the pour point of the lubricant is within the standard for  
154 engine oil [9], the produced lubricant would require a calculated amount of pour point  
155 depressant (PPD) for it to be applicable in extreme low temperature conditions as compared  
156 to synthesised engine oils. The PPDs will minimise the negative effects of precipitation at  
157 low temperatures during usage.

158 Free fatty acid component of crude vegetable oil is an important factor used to determine the  
159 food or oleochemical application of the vegetable oil; percentage free fatty acid greater than  
160 5% suggests that the oil could be useful for the production of biodiesel, biolubricant, and  
161 bioplastics [15]. The free fatty acid value of the crude extract ( $1.06 \pm 12$ ) decreases to  
162  $0.85 \pm 12$  in the formulated lubricating oil (Table 2), this is due to the effect of additive in the  
163 formulated oil.

164  
165  
166

**Table 1. Optimised Blends for the Lubricant Formulation**

Run Order	CSO	SN 500	Additive	KV @ 100°C (cSt)	KV @ 40 °C (cSt)	VI	FP (°C)	PP (°C)
1	0.00	90.00	10.00	9.00	89.20	75	233	-10
2	68.75	23.75	7.50	8.02	34.43	219	146	-13
3	23.75	68.75	7.50	10.50	50.18	207	241	-19
4	90.00	0.00	10.00	8.11	20.44	270	153	-24
5	28.75	68.75	2.50	9.31	53.00	166	238	-28
6	68.75	28.75	2.50	7.70	41.30	163	144	-16
7	10.00	90.00	0.00	10.70	65.20	157	236	-14
8	28.75	68.75	2.50	9.30	53.11	167	240	-28

9	90.00	0.00	10.00	8.11	20.62	268	153	-23
10	23.75	68.75	7.50	10.51	50.13	208	241	-18
11	90.00	10.00	0.00	8.60	44.90	172	166	-28
12	68.75	28.75	2.50	7.60	41.90	155	145	-15
13	68.75	23.75	7.50	8.02	34.47	217	145	-13
14	47.50	47.50	5.00	9.59	49.32	183	232	-29
15	0.00	90.00	10.00	9.50	89.17	85	231	-11
16	90.00	10.00	0.00	8.60	44.90	174	164	-27
17	47.50	47.50	5.00	9.50	49.32	182	233	-28
18	10.00	90.00	0.00	10.70	65.20	156	236	-14

Key: **CSO** = Calabash Seed Oil; **KV** = Kinematic Viscosity;  
**VI** = Viscosity Index; **FP** = Flash Point; **PP** = Pour Point

167  
168  
169  
170  
171  
172  
173

**Table 2. Physicochemical Properties of CSO and the Lubricant**

Parameters	Units	CSO	Lubricant	Engine oil range*
Kinematic Viscosity @ 40°C	cSt	21.78±1.22	53.11±1.03	> 28.80
Kinematic Viscosity @ 100°C	cSt	6.55±1.01	9.30±2.11	> 4.10
Viscosity Index		266±2.11	167±0.51	> 90.00
Flash Point	°C	145±1.01	240±2.01	> 150.00

Pour Point	°C	-11±0.01	-28±1.31	< -5.00
Free Fatty Acid	mg KOH g <sup>-1</sup>	1.06±12	0.85±12	-

Key: **CSO** = Calabash Seed Oil; ( $\pm$ ) = Mean Value Plus or Minus Standard Deviation (n = 3); (\*) = Owuna et al., 2018

174  
175  
176  
177

#### 4. CONCLUSION

179

This research was undertaken in order to confirm the use of CSO for the formulation of engine oil. The results obtained showed that a blend of CSO with mineral based oil (SN 500) and additive was used to achieve a formulation of lubricant that has quality parameters that are within the standard for engine oils and are comparable to commercial engine oils. The lubricant can be used as substitute for any applications where synthesised engine oils are applicable.

186

187

188

#### REFERENCES

189

1. Jagadeesh, K. M., Satish, V. K., Venkatesh, K. and Kathyayini, N. Environmentally Friendly Functional Fluids from Renewable and Sustainable Sources – A Review. *Renewable and Sustainable Energy Reviews*. (2018);18:1787-1801.

190

191

2. Mobarak, H. M., Niza, M. E., Masjuki, H. H., Kalam, M. A., Al Mahmud, K. A .H., Habibullah, M. and Ashraful, A. M. The Prospects of Biolubricants as Alternative in Automotive Applications. *Renewable and Sustainable Energy Reviews*. (2014);33:34-43.

192

193

194

195

3. Abadi, B. and Shimels, A. A Review on Potential and Status of Biofuel Production in Ethiopia. *Journal of Plant Sciences*. (2017);5:82-89.

196

197

4. Muhammad, C., Usman, Z. and Agada, F. Biodiesel Production from Ceiba pentandra Seed Oil Using CaO Derived from Snail Shell as Catalyst. *Petroleum Science and Engineering*. (2018);2:7-16.

198

199

200

5. Dabai, M. U., Owuna, F. J., Sokoto, M. A. and Abubakar, A. L. Assessment of Quality Parameters of Ecofriendly Biolubricant from Waste Cooking Palm Oil. *Asian Journal of Applied Chemistry Research*. (2018);1:1-11.

201

202

203

6. Adolf, O., Akwasi, A., Gyang, N. O., Amoa, C. A. and Akorfa, A. A. Comparative Assessment of some Physico-chemical Properties of Seed Oils of *Parkia biglobosa* and *Monodora myristica* with some Commercial Oils. *African Journal of Food Science*. (2018);12:1-5.

204

205

206

207

7. Anuchaya, D., Vijay, K. D., and Dhanapati, D. Evaluation of the effectiveness of potato peel extract a natural antioxidant on biodiesel oxidation stability. *Industrial Crops & Products*. (2018);123:454-460.

208

209

210

8. Mukhtar, M., Muhammad, C., Dabai, M. U. and Muhammad, M. Ethanolysis of Calabash (*Langeneria sinceraria*) Seed Oil for the Production of Biodiesel. *American Journal of Energy Engineering*. (2014);2:141-145.

211

212

213

9. Owuna, F. J., Dabai, M. U., Sokoto, M. A., Muhammad, C., and Abubakar, A. L. Use of *Lagenaria siceraria* Seed Oil for the Production of Environmentally Friendly Biolubricant. *American Journal of Applied and Industrial Chemistry*. (2018);2:1-7

214

215

216

- 217 10. Jeevan, T. P. and Jayaram, S. R. Experimental Investigation on the Performance of  
 218 Vegetable Oil Based Cutting Fluids in Drilling AISI 304L Using Taguchi Technique.  
 219 Tribology Online. (2018);13:50-56.
- 220 11. Ratchadaporn. I., Natthawud, D., Churat, T., Jutaporn, C. and Rameshpraba  
 221 Ramaraj. An Experimental Investigation of Palm Oil as an Environment Friendly  
 222 Biolubricant. The 22nd Tri-U International Joint Seminar and Symposium, Jangsu  
 223 University, China, October18 - October 23. (2015);2015:1-4.
- 224 12. Ebtisam, K. H., Elmelawu, M. S., Salah, A. K. and Elbasuny, N. M. Manufacture of  
 225 Environment Friendly Biolubri cants from Vegetable Oils. Egyptian Journal of  
 226 Petroleum. (2017);26:53-59.
- 227 13. Janes, A. and Chaineaux, J. Experimental Determination of Flash Points of  
 228 Flammable Liquid Aqueous Solutions. Chemical Engineering Transactions.  
 229 (2013);31:943-948.
- 230 14. Gobinda, K., Pranab, G. and Bragendra, K. S. Chemically Modified Vegetable Oils to  
 231 Prepare Green Lubricants. Lubricants. (2017);44:1-17.
- 232 15. Japir, A. A.-W., Salmon, S., Derawi, D., Bahadi, M., Al-Shuja'a, S. and Yusop, M.  
 233 Physicochemical Characteristics of High Free Acid Crude Palm Oil. Oilseeds & Fats  
 234 Crops and Lipids. (2017);24:1-9.

235

236

## DEFINITIONS

237

**ASTM** = American Society for Testing and Materials

238

**CSO** = Calabash Seed Oil

239

**HTHS** = High Temperature High Shear

240

**mg/l** = Milligram per Litre

241

**mPa** = Millipascal

242

**SEA** = Society of Automotive Engineers

243

**Wt%** = Weight Percent

244

245

## APPENDIX

246

247

### Appendix I: Fully Synthesised Mobil 1 5W-30 Oil (Mobil) Technical Data Sheet

248

#### Typical Properties

<b>Mobil 1 5W-30</b>	
SAE Grade	5W-30
Viscosity @ 100°C, cSt (ASTM D445)	11.0
Viscosity, @ 40°C, cSt (ASTM D445)	61.7
Viscosity Index	172
Sulfated Ash, wt% (ASTM D874)	0.8
HTHS Viscosity, mPa•s @ 150°C (ASTM D4683)	3.1
Pour Point, °C (ASTM D97)	-42
Flash Point, °C (ASTM D92)	230
Density @15.6 °C, mg/l (ASTM D4052)	0.855

249

250

Source: [http://mobil.moovelub.com/sites/default/files/mobil\\_1\\_5w-30.pdf](http://mobil.moovelub.com/sites/default/files/mobil_1_5w-30.pdf)

251