

Original research paper

Testing Selected Soils from Bamyan Center Agricultural Research Farms for Initial Macro and Micro Nutrients with Focus on Phosphorus Availability

ABSTRACT

Phosphorus (P) and micronutrients deficiencies are common in alkaline soils. Alkaline soils make up some of the most productive agricultural lands in the Bamyan centre of Bamyan province in central Afghanistan and little is known about the phosphorus fertility status of these soils. ~~Our~~ The objectives of this study were to determine the soil fertility status of 4 soils collected from the four research farms in Bamyan centre and also to conduct P incubation studies on these soils to determine the fixation and availability of added P. The experimental design was a randomized complete block with 3 replications of each treatment. Soils used in this study had pH > 8.0, free CaCO₃ contents of 9.3-10% and texture ranged from silty clay loam to sandy loam. Seven rates of P (0, 5.6, 10.9, 16.4, 21.8, 32.8, 43.7 mg kg⁻¹) were added as monocalcium phosphate [Ca (H₂PO₄) · 2H₂O]. Soils were incubated at approximately 0.03 MPa soil tension for 15, 30, 45, 60, 75 and 90 days at 25°C and the Mehlich 3 soil test was used to determine available P. Mehlich 3 extractable P did not change consistently with time so data were averaged over all 6 sampling periods to determine the effects of P rate on Mehlich 3 P soil test levels. Soils segregated into two groups of two soils each that responded similarly in their response to P applications. Approximately 70 percent of the applied P remained available in one group of soils, while in the other group of soils, approximately 50 percent of the applied P remained available.

Key words: Bamyan soil characteristics, calcareous soil, phosphorus, Mehlich 3 soil test.

Introduction

Phosphorus (P) is an essential nutrient for plant growth and its deficiency is common in calcareous soils. Soil P is often not available at the minimum level needed for plant growth and available soil P can be improved by adding P fertilizers (Jalali and Ranjbar, 2010). Because P is a finite resource, proper management of fertilizer P is necessary to maximize crop production while minimizing the risk of P loss to the environment. Phosphorus in agricultural soils appears in three fractions: organic P (25-30%), insoluble inorganic P (about 75%) and a small soluble P fraction. Less than 10% of the total P in soil is available in the plant-animal life cycle (Ozanne, 1980).

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34 Phosphorus makes up approximately 0.2% of plant dry weight and P is a component of key
35 molecules such as phospholipids, nucleic acids and ATP, so plants cannot grow well without a
36 sufficient supply of P (Schachtman et al., 1998). Although the total amount of P in soils can be
37 relatively high, because most of this P is not present in plant-available forms it is important to
38 make sure that optimum P fertilizer management strategies are developed to ensure plant
39 productivity.

40 Little research has been conducted on calcareous soils of Afghanistan, and these are the most
41 highly productive agricultural lands in the country.

42 Not all of the P fertilizer applied to soils is available to plants due to surface adsorption and
43 precipitation processes, particularly in calcareous soils (Afif et al., 1993). For example, there
44 was a negative correlation between P fertilizer and CaCO_3 content when available P was
45 measured as resin extractable P in 20 calcareous soils of the continental USA (Jones et al.,
46 1984). For another group of calcareous soils from several countries, researchers found that P
47 availability after both 30 and 180 days of incubation was closely correlated to CaCO_3 content
48 (Sharpley et al., 1989). Larsen and Widdowson (1970) reported that P sorption in calcareous
49 soils increased with increased CaCO_3 content.

50 Ryan et al. (1985) found that solid phase CaCO_3 proved to be the most dominant phase
51 controlling P reactions in the soils they studied. However, some studies argue that the reactivity
52 of CaCO_3 could be more dependent on the specific surface area, which is related to CaCO_3
53 particle size distribution, than total CaCO_3 when relating soil properties to P reactions in
54 calcareous soils (Holford and Mattingly, 1975). The same idea has been confirmed by Borrero et
55 al. (1988).

56 Soil pH affects P solubility and plant uptake (Ortas and Rowell, 2000). Researchers suggest that
57 phosphate adsorption capacity increases as pH decreases (Bolan and Hedley, 1990). According
58 to some researchers, it is important to understand the effect of pH on P partitioning to explain
59 differences in observed P sorption levels in soil (Tunesi et al., 1999). Tunesi et al. (1999) found
60 that in Na-saturated soils, phosphate partitioning into the solid phase decreased as pH
61 increased. Moreover, according to these researchers, at higher pH values lower P sorption can
62 be observed at the initial portion of the isotherm. Tunesi et al. (1999) also suggested that a
63 higher pH value could decrease the solubility of Ca-P mineral phases and increase precipitation,
64 which would further decrease solution P.

65 An incubation study on three calcareous soils of the UK indicated that decreasing soil pH could
66 increase soil P solubility (Ortas and Rowell, 2000). They also pointed out that the formation of
67 insoluble Ca minerals was the key factor in decreasing P availability, particularly in higher pH
68 soils. Laboratory studies also have indicated that P sorption can vary with pH (Zhou et al.,

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69 2005). These researchers also believe that P sorption decreases as pH increases and as a result,
70 the surface charge becomes more negative.

71 The effect of soil pH on the dissolution of phosphate rocks and the availability of
72 inorganic P to plants has been studied by many researchers. For example, Bolan and Hedley
73 (1990) showed an increase in plant-available P from phosphate rocks with decreased soil pH.

74 Keeping in mind on the above information about the lowP availability in alkaline soil, the
75 present investigation ~~has been planned to carryare to be conducted out~~ during 2016 -2017at
76 soil science laboratory, Bamyancenter, in Bamyancenter, with the following objectives.

77 1).To determine the soil fertility status of soils to be collected from the Bamyancenter
78 farms and Department of Agriculture Research Farms and;

79 2). To conduct P incubation studies on these soils to determine the fixation and availability of
80 added P.

81 Research Methodology

82 Routine Soil Analyses

83 Four soils will be collected from four agricultural research farms of Bamyancenter. Soil samples
84 will be air dried and ground to pass through a 2-mm sieve for all laboratory analyses. The
85 Bamyancenter soil fertility laboratory will be used to conduct all the analysis. All the
86 equipments (reagents, filter papers, beakers etc.. which are necessary to conduct this
87 experiment are available inside the soil fertility lab of Bamyancenter) and the reference
88 book that will be used for each of the necessary test is " Recommended Chemical Soil Test
89 Procedures for The North Central Region , Agricultural Experiment Stations of Illinois,
90 Indiana, Iowa, Kansas, Michigan, Minnesota, Missouri, Nebraska and the U.S. Department of
91 Agriculture, No 221).

92 3.2. Phosphorus Incubation Study

93 The incubation experiment will be a split plot design with three blocked replicates.
94 Whole unit treatments will be 4 soils from Bamyancenter, with P applied at 7 different rates (0, 5.6,
95 10.9, 16.4, 21.8, 32.8, 43.7 mg P kg⁻¹). Subsamples will be taken 6 times after 15, 30, 45, 60,
96 75 and 90 days of incubation and Mehlich 3 P will be measured on each sample. Finally, the
97 statistical analysis (ANOVA) will be conducted to analyze the data.

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99 Result and Discussion

100 Soil Chemical and Physical Properties

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101 The 4 soils used in this experiment varied in their initial P contents and other soil chemical
102 and physical properties (Table 1 and Table 2). The soils' initial total P ranged from 19.5 to 24.6
103 mg kg⁻¹ (Table 1). Soils pH and percent CaCO₃ ranged from 8.3 to 8.7 and 9.3 to 10.7%,
104 respectively. Soil textural class ranged from silty clay loam to sandy loam and clay content
105 ranged from 18-36% (Table 2).

106 **Effect of Applied Phosphorus on Mehlich 3 Phosphorus**

107 Mehlich 3 soil test P increased with P additions at all sampling times, but the magnitude of
108 increase in soil test P varied among incubation times and did not follow any pattern that could
109 be explained based on soil properties or time. Therefore, ~~we averaged the~~ treatment response
110 ~~was aeraged~~ over time and only the effect of P rates on the change in Mehlich 3 soil P test
111 level was considered ~~only the effect of P rates on the change in Mehlich 3 soil P test level~~. Soils
112 1 and 2 responded similarly to each other, but significantly different ($P < 0.01$) from soils 3 and
113 4, with respect to the change in Mehlich 3 soil test P with P additions (Table 3). Even though
114 soils 1 and 2 differed in texture, pH, and initial P content responded similarly in P application
115 rate (Table 1, Table 2, and Table 3). These soils all had greater initial soil test P levels compared
116 to soils 3 and 4 (Table 1).

117 The predicted values for the increase in Mehlich 3 soil test P with P addition for these two
118 soils were described by the following equation:

$$119 Y = 8.97 + 0.01X^2$$

120 Where Y equals the change in soil test P and X equals P addition with both variables expressed
121 in mg P kg⁻¹ soil. The slope and intercepts were different from zero ($P < 0.0001$) and the
122 correlation between P application and change in Mehlich 3 soil test P level was highly significant
123 ($r^2 = 0.93$; $P < 0.0001$; Table 3). At the greatest P addition of 43.7 mg P kg⁻¹, The Mehlich 3 soil
124 test P value changed approximately 30 mg P kg⁻¹. Approximately 70% of the added P remained
125 available in these soils.

126 Bolan and Hedley (1990) suggested that phosphate adsorption capacity increases as pH
127 decreases. Moreover, Ortas and Rowell (2000) reported that soil pH affects P solubility and
128 plant uptake. Soil 2 had a lower clay content (18%) compared to soil 1 (36%). We expected
129 that less P might have been sorbed by soil 1 compared to all other soils. Fox and Kamprath
130 (1970) reported a positive correlation between the clay content and P adsorption capacity of
131 soils and Jones et al. (1984) found a negative correlation between P fertilizer and CaCO₃
132 content and changes in resin extractable P with P additions in 20 calcareous soils of the
133 continental USA. Sharply et al. (1989). Also found that P availability after 30 and 180 days of
134 incubation was closely correlated to the CaCO₃ content of calcareous soils from several
135 countries. Soils 3 and 4 also responded similarly to P additions as expressed by the change in
136 Mehlich 3 soil test P values. Soils 3 and 4 were both silty clay loam with a pH ranges from 8.6-
137 8.7 (Table 1).

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138 The predicted values for the increase in Mehlich 3 soil test P with P additions for these two
139 soils were estimated by the following equation:

140 $Y = 2.90 + 0.01X^2$

141 Where Y equals the change in soil test P and X equals P addition with both variables expressed
142 in mg P kg⁻¹ soil. The slope and intercepts were different from zero ($P < 0.0001$) and the
143 correlation between P application and change in Mehlich 3 soil test P level was highly significant
144 ($r^2 = 0.93$; $P < 0.0001$; Table 3). In these soils we found roughly a 21 mg P kg⁻¹ increase in
145 Mehlich 3 soil P test level after adding 43.7 mg P kg⁻¹. Approximately 50% of the added P
146 remained available in these soils, which was significantly less than in soils 1 and 2. These results
147 were not unexpected and Soils 3 and 4 had lesser initial soil test P values than Soils 1 and 2.
148 Soil 4 had the greatest pH. In an incubation study on three calcareous soils of the UK, Ortas and
149 Rowell (2000) found that decreasing soil pH could increase soil P solubility, and Fox and
150 Kamprath (1970) found a positive correlation between clay content and soil P adsorption
151 capacity.

152 The percent free CaCO₃ for Soils 3 and 4 also were ranged from 10.1-10.6 % and according
153 to previous researchers, P sorption in calcareous soils increased with increased CaCO₃ content
154 (Larsen and Widdowson, 1970).

155 Conclusion

156 The 4 calcareous soils from Bamyancenter varied in their initial macro and micro nutrients
157 content and many other soil properties. The rate and time effects varied among all 4 soils and
158 time effect on change in Mehlich 3 soil test P levels were not consistent. Soil Mehlich 3 P
159 increased with added P and the availability of added P differed between two groups of soils.
160 These two groups of soils were separated between high P testing soils and low P testing soils.

161 Recommendation

162 1) Based on results, the organic content in Bamyancenter soil is low, so it is recommended to apply
163 the different sources of organic fertilizers, especially ~~decomposed~~ composted animal manures
164 on the cultivated land.

165 2) According to many studies, the application of both chemical and organic fertilizers highly
166 ~~increasing-increase~~ the efficiency of chemical fertilizers. So, it is recommended to apply both
167 chemical and organic fertilizer simultaneously on the field.

168 3) In mechanical agriculture, the application of fertilizers must be according to availability and
169 amount of essential nutrients in the soil and needs of cultivated plants. So, the results ~~in from~~
170 this research ~~will-should~~ be able to at least ~~a raw data for correct applying~~ assist farmers in
171 applying the correct doses of chemical fertilizers.

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Table 1: Origin and selected Properties of soils collected from the Bamyancentre in the Bamyan province of Central Afghanistan.

Soil ID	Soil Origin	pH	EC	CEC	OM	N	P	K	Ca	Mg	CaCO ₃
			dS/m	cmol kg ⁻¹	%						
1	Bamyan Uni. Ag. Farm A.	8.4	0.5	29.1	0.9	51.4	24.6	92.5	4900	780	9.3
2	Bamyan Uni. Ag. Farm B.	8.3	1.3	30.4	1.1	44.8	23.7	106	5300	685	10.7
3	Mollagholam A.	8.6	0.6	26.1	0.6	48.2	22.1	86.4	5150	700	10.1
4	Mollagholam B	8.7	0.3	25.6	0.5	44.7	19.5	94.2	5400	650	10.6

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Table 2: Origin and selected Properties of soils collected from the Bamyancentre in the Bamyan province of Central Afghanistan.

Soil ID	Soil Origin	S	Mn	Fe	Cu	Zn	Sand	silt	Clay	Textural Class
		-----mg kg ⁻¹ -----					-----%-----			
1	Bamyan Uni. Ag. Farm A.	9.3	64	27	4	3	48	16	36	Sandy clay
2	Bamyan Uni. Ag. Farm B.	12.7	71	23	6	2	65	17	18	Sandy loam
3	Mollagholam A.	9.5	65	25	7	3	18	50	32	Silty clay loam
4	Mollagholam B	7.8	67	18	5	4	19	52	29	Silty clay loam

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Table 3: Analysis of variance and regression coefficients for the effect of applied phosphorus on Mehlich 3 P in 4 Bamyan Centre soils.

R-Square 0.93	Pr>F <0.0001	CoeffVar 16.84	Root MSE 2.12	M-3 P 12.52	
Source	Df	Type II SS	Mean square	F Value	Pr>F
Soil	1	330.28	330.28	73.52	<0.0001**
R ²	1	1657.1	1657.1	368.52	<0.0001**
Parameter	Estimate	Stand Error	T Value	Pr> t	
Soil 1 2	8.960	0.60824	14.73	<0.0001**	
Soil 3 4	2.902	0.60824	14.76	<0.0001**	
R ²	0.0102	0.00052	19.2	<0.0001**	

185

** = significant at P ≤ 0.01, * = significant at P ≤ 0.05, and ns = non-significant

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