

**Original research paper****Testing Selected Soils from Bamyan Center Agricultural Research Farms for Initial Macro and Micro Nutrients with Focus on Phosphorus Availability****ABSTRACT**

Phosphorus (P) and Micronutrients deficiencies are common in alkaline soils. Alkaline soils make up some of the most productive agricultural lands in the Bamyan centre of Bamyan province in central Afghanistan and little is known about the phosphorus fertility status of these soils. Our objectives were to determine the soil fertility status of 4 soils collected from the four research farms in Bamyan centre and also to conduct P incubation studies on these soils to determine the fixation and availability of added P. The experiment design was a randomized complete block with 3 replications of each treatment. Soils used in this study had pH > 8.0, free CaCO<sub>3</sub> contents of 9.3-10% and texture ranged from silty caly loam to sandy loam. Seven rates of P (0, 5.6, 10.9, 16.4, 21.8, 32.8, 43.7 mg kg<sup>-1</sup>) were added as either monocalcium phosphate [Ca (H<sub>2</sub>PO<sub>4</sub>) · 2H<sub>2</sub>O]. Soils were incubated at approximately 0.03 MPa soil tension for 15, 30, 45, 60, 75 and 90 days at 25 ° C and the Mehlich 3 soil test was used to determine available P. Mehlich 3 extractable P did not change consistently with time so data were averaged over all 6 sampling periods to determine the effects of P rate on Mehlich 3 P soil test levels. Soils segregated into two groups of two soils each that responded similarly in their response to P applications. Approximately 70 percent of the applied P remained available in one group of soils, while in the other group of soils, approximately 50 percent of the applied P remained available.

**Key words:** Bamyan soil characteristics, Calcareous soil, phosphorus, Mehlich3 soil test.

**Introduction**

Phosphorus (P) is an essential nutrient for plant growth and its deficiency is common in calcareous soils. Soil P is often not available at the minimum level needed for plant growth and available soil P can be improved by adding P fertilizers (Jalali and Ranjbar, 2010). Because P is a finite resource, proper management of fertilizer P is necessary to maximize crop production while minimizing the risk of P loss to the environment. Phosphorus in agricultural soils appears in three fractions: organic P (25-30%), insoluble inorganic P (about 75%) and a small soluble P fraction. Less than 10% of the total P in soil is available in the plant-animal life cycle (Ozanne, 1980).

33 Phosphorus makes up approximately 0.2% of plant dry weight and P is a component of key  
34 molecules such as phospholipids, nucleic acids and ATP, so plants cannot grow well without a  
35 sufficient supply of P (Schachtman et al., 1998). Although the total amount of P in soils can be  
36 relatively high, because most of this P is not present in plant-available forms it is important to  
37 make sure that optimum P fertilizer management strategies are developed to ensure plant  
38 productivity.

39 Little research has been conducted on calcareous soils of Afghanistan, and these are the  
40 most highly productive agricultural lands in the country.

41 Not all of the P fertilizer applied to soils is available to plants due to surface adsorption and  
42 precipitation processes, particularly in calcareous soils (Afif et al., 1993). For example, there  
43 was a negative correlation between P fertilizer and  $\text{CaCO}_3$  content when available P was  
44 measured as resin extractable P in 20 calcareous soils of the continental USA (Jones et al.,  
45 1984). For another group of calcareous soils from several countries, researchers found that P  
46 availability after both 30 and 180 days of incubation was closely correlated to  $\text{CaCO}_3$  content  
47 (Sharpley et al., 1989). Larsen and Widdowson (1970) reported that P sorption in calcareous  
48 soils increased with increased  $\text{CaCO}_3$  content.

49 Ryan et al. (1985) found that solid phase  $\text{CaCO}_3$  proved to be the most dominant phase  
50 controlling P reactions in the soils they studied. However, some studies argue that the reactivity  
51 of  $\text{CaCO}_3$  could be more dependent on the specific surface area, which is related to  $\text{CaCO}_3$   
52 particle size distribution, than total  $\text{CaCO}_3$  when relating soil properties to P reactions in  
53 calcareous soils (Holford and Mattingly, 1975). The same idea has been confirmed by Borrero et  
54 al. (1988).

55 Soil pH affects P solubility and plant uptake (Ortas and Rowell, 2000). Researchers suggest  
56 that phosphate adsorption capacity increases as pH decreases (Bolan and Hedley, 1990).  
57 According to some researchers, it is important to understand the effect of pH on P partitioning  
58 to explain differences in observed P sorption levels in soil (Tunesi et al., 1999). Tunesi et al.  
59 (1999) found that in Na-saturated soils, phosphate partitioning into the solid phase decreased  
60 as pH increased. Moreover, according to these researchers, at higher pH values lower P  
61 sorption can be observed at the initial portion of the isotherm. Tunesi et al. (1999) also  
62 suggested that a higher pH value could decrease the solubility of Ca-P mineral phases and  
63 increase precipitation, which would further decrease solution P.

64 An incubation study on three calcareous soils of the UK indicated that decreasing soil pH  
65 could increase soil P solubility (Ortas and Rowell, 2000). They also pointed out that the  
66 formation of insoluble Ca minerals was the key factor in decreasing P availability, particularly in  
67 higher pH soils. Laboratory studies also have indicated that P sorption can vary with pH (Zhou et

68 al., 2005). These researchers also believe that P sorption decreases as pH increases and as a  
69 result, the surface charge becomes more negative.

70 The effect of soil pH on the dissolution of phosphate rocks and the availability of  
71 inorganic P to plants has been studied by many researchers. For example, Bolan and Hedley  
72 (1990) showed an increase in plant-available P from phosphate rocks with decreased soil pH.

73 Keeping in mind on the above information about the low P availability in alkaline soil, the  
74 present investigation has been planned to carry out during 2016 -2017 at soil science  
75 laboratory, Bamyán university, in Bamyán center, with the following objectives.

76 1).To determine the soil fertility status of soils to be collected from the Bamyán University  
77 farms and Department of Agriculture Research Farms and;

78 2). To conduct P incubation studies on these soils to determine the fixation and availability of  
79 added P.

## 80 **Research Methodology**

### 81 **Routine Soil Analyses**

82 Four soils will be collected from four agricultural research farms of Bamyán. Soil samples  
83 will be air dried and ground to pass through a 2-mm sieve for all laboratory analyses.The  
84 Bamyán University soil fertility laboratory will be used to conduct all the analysis. All the  
85 equipments ( reagents, filter papers, beakers etc.. which are necessary to conduct this  
86 experiment are available inside the soil fertility lab of Bamyán University) and the reference  
87 book that will be used for each of the necessary test is “ Recommended Chemical Soil Test  
88 Procedures for The North Central Region , Agricultural Experiment Stations of Illinois,  
89 Indiana, Iowa, Kansas, Michigan, Minnesota, Missouri, Nebraska and the U.S. Department of  
90 Agriculture, No 221).

### 91 **3.2. Phosphorus Incubation Study**

92 The incubation experiment will be a split plot design with three blocked replicates.  
93 Whole unit treatments will be 4 soils from Bamyán, with P applied at 7 different rates (0, 5.6,  
94 10.9, 16.4, 21.8, 32.8, 43.7 mg P kg<sup>-1</sup>). Subsamples will be taken 6 times after 15, 30, 45, 60,  
95 75and 90 days of incubation and Mehlich 3 P will be measured on each sample. Finally, the  
96 statistical analysis (ANOVA) will be conducted to analyze the data.

97

## 98 **Result and Discussion**

### 99 **Soil Chemical and Physical Properties**

100 The 4 soils used in this experiment varied in their initial P contents and other soil chemical  
101 and physical properties (Table 1 and Table 2). The soils initial total P ranged from 19.5 to 24.6  
102 mg kg<sup>-1</sup> (Table 1). Soils pH and percent CaCO<sub>3</sub> ranged from 8.3 to 8.7 and 9.3 to 10.7%,  
103 respectively. Soil textural class ranged from Silty clay loam to sandy loam and clay content  
104 ranged from 18-36% (Table 2).

### 105 **Effect of Applied Phosphorus on Mehlich 3 Phosphorus**

106 Mehlich 3 soil test P increased with P additions at all sampling times, but the magnitude of  
107 increase in soil test P varied among incubation times and did not follow any pattern that could  
108 be explained based on soil properties or time. Therefore, we averaged treatment response over  
109 time and considered only the effect of P rates on the change in Mehlich 3 soil P test level. Soils  
110 1 and 2 responded similarly to each other, but significantly different ( $P < 0.01$ ) from soils 3 and  
111 4, with respect to the change in Mehlich 3 soil test P with P additions (Table 3). Even though  
112 soils 1 and 2 differed in texture, pH, and initial P content responded similarly in P application  
113 rate (Table 1, Table 2, and Table 3). These soils all had greater initial soil test P levels compared  
114 to soils 3 and 4 (Table 1).

115 The predicted values for the increase in Mehlich 3 soil test P with P addition for these two  
116 soils were described by the following equation:

$$117 Y = 8.97 + 0.01X^2$$

118 Where Y equals the change in soil test P and X equals P addition with both variables expressed  
119 in mg P kg<sup>-1</sup> soil. The slope and intercepts were different from zero ( $P < 0.0001$ ) and the  
120 correlation between P application and change in Mehlich 3 soil test P level was highly significant  
121 ( $r^2 = 0.93$ ;  $P < 0.0001$ ; Table 3). At the greatest P addition of 43.7 mg P kg<sup>-1</sup>, The Mehlich 3 soil  
122 test P value changed approximately 30 mg P kg<sup>-1</sup>. Approximately 70% of the added P remained  
123 available in these soils.

124 Bolan and Hedley (1990) suggest that phosphate adsorption capacity increases as pH  
125 decreases. Moreover, Ortas and Rowell (2000) reported that soil pH affects P solubility and  
126 plant uptake. Soil 2 had a lower clay content (18%) compared to soils 1 (36%). We expected  
127 that less P might have been sorbed by soil 1 compared to all other soils. Fox and Kamprath  
128 (1970) reported a positive correlation between the clay content and P adsorption capacity of  
129 soils and Jones et al. (1984) found a negative correlation between P fertilizer and CaCO<sub>3</sub>  
130 content and changes in resin extractable P with P additions in 20 calcareous soils of the  
131 continental USA. Sharply et al. (1989). Also found that P availability after 30 and 180 days of  
132 incubation was closely correlated to the CaCO<sub>3</sub> content of calcareous soils from several  
133 countries. Soils 3 and 4 also responded similarly to P additions as expressed by the change in  
134 Mehlich 3 soil test P values. Soils 3 and 4 were both silty clay loam with a pH ranges from 8.6-  
135 8.7 (Table 1).

136 The predicted values for the increase in Mehlich 3 soil test P with P additions for these two  
137 soils were estimated by the following equation:

138  $Y = 2.90 + 0.01X^2$

139 Where Y equals the change in soil test P and X equals P addition with both variables expressed  
140 in mg P kg<sup>-1</sup> soil. The slope and intercepts were different from zero ( $P < 0.0001$ ) and the  
141 correlation between P application and change in Mehlich 3 soil test P level was highly significant  
142 ( $r^2 = 0.93$ ;  $P < 0.0001$ ; Table 3). In these soils we found roughly a 21 mg P kg<sup>-1</sup> increase in  
143 Mehlich 3 soil P test level after adding 43.7 mg P kg<sup>-1</sup>. Approximately 50% of the added P  
144 remained available in these soils, which was significantly less than in Soils 1 and 2. These results  
145 were not unexpected and Soils 3 and 4 had lesser initial soil test P values than Soils 1 and 2. Soil  
146 4 had the greatest pH. In an incubation study on three calcareous soils of the UK, Ortas and  
147 Rowell (2000) found that decreasing soil pH could increase soil P solubility, and Fox and  
148 Kamprath (1970) found a positive correlation between clay content and soil P adsorption  
149 capacity.

150 The percent free CaCO<sub>3</sub> for Soils 3 and 4 also were ranged from 10.1-10.6 % and according  
151 to previous researchers, P sorption in calcareous soils increased with increased CaCO<sub>3</sub> content  
152 (Larsen and Widdowson, 1970).

### 153 **Conclusion**

154 The 4 calcareous soils from Bamyan center varied in their initial macro and micro nutrients  
155 content and many other soil properties. The rate and time effects varied among all 4 soils and  
156 time effect on change in Mehlich 3 soil test P levels were not consistent. Soil Mehlich 3 P  
157 increased with added P and the availability of added P differed between two groups of soils.  
158 These two groups of soils were separated between high P testing soils and low P testing soils.

### 159 **Recommendation**

160 1) Based on results, the organic content in Bamyan soil is low, so recommended to apply the  
161 different sources of organic fertilizers, especially decomposed animal manures on the cultivated  
162 land.

163 2) According to many studies, the application of both chemical and organic fertilizers highly  
164 increasing the efficiency of chemical fertilizers. So, it is recommended to apply both chemical  
165 and organic fertilizer simultaneously on the field.

166 3) In mechanical agriculture, the application of fertilizers must be according to availability and  
167 amount of essential nutrients in the soil and needs of cultivated plants. So, the results in this  
168 research will be at least a raw data for correct applying doses of chemical fertilizers.

169

170

171

172

173

174 Table 1: Origin and selected Properties of soils collected from the Bamyan centre in the Bamyan  
 175 province of Central Afghanistan.

Soil ID	Soil Origin	pH	EC dS/m	CEC cmol kg <sup>-1</sup>	OM %	-----mg kg <sup>-1</sup> -----					CaCO <sub>3</sub> %
						N	P	K	Ca	Mg	
1	Bamyan Uni. Ag. Farm A.	8.4	0.5	29.1	0.9	51.4	24.6	92.5	4900	780	9.3
2	Bamyan Uni. Ag. Farm B.	8.3	1.3	30.4	1.1	44.8	23.7	106	5300	685	10.7
3	Mollagholam A.	8.6	0.6	26.1	0.6	48.2	22.1	86.4	5150	700	10.1
4	Mollagholam B	8.7	0.3	25.6	0.5	44.7	19.5	94.2	5400	650	10.6

176  
 177 Table 2: Origin and selected Properties of soils collected from the Bamyan centre in the Bamyan  
 178 province of Central Afghanistan.

Soil ID	Soil Origin	-----mg kg <sup>-1</sup> -----					-----%-----			Textural Class
		S	Mn	Fe	Cu	Zn	Sand	silt	Clay	
1	Bamyan Uni. Ag. Farm A.	9.3	64	27	4	3	48	16	36	Sandy clay
2	Bamyan Uni. Ag. Farm B.	12.7	71	23	6	2	65	17	18	Sandy loam
3	Molla gholam A.	9.5	65	25	7	3	18	50	32	Silty clay loam
4	Molla gholam B	7.8	67	18	5	4	19	52	29	Silty clay loam

179  
 180 Table 3: Analysis of variance and regression coefficients for the effect of applied phosphorus on Mehlich  
 181 3 P in 4 Bamyan Centre soils.

R-Square	Pr>F	Coeff Var	Root MSE	M-3 P	
0.93	<0.0001	16.84	2.12	12.52	
Source	Df	Type II SS	Mean square	F Value	Pr>F
Soil	1	330.28	330.28	73.52	<0.0001**
R <sup>2</sup>	1	1657.1	1657.1	368.52	<0.0001**
Parameter	Estimate	Stand Error	T Value	Pr> t	
Soil 1 2	8.960	0.60824	14.73	<0.0001**	
Soil 3 4	2.902	0.60824	14.76	<0.0001**	
R <sup>2</sup>	0.0102	0.00052	19.2	<0.0001**	

182 \*\* = significant at P ≤ 0.01, \* = significant at P ≤ 0.05, and ns = non-significant

183

184

185

186 **References:**

- 187 Afif, E., A. Matar, and J. Torrent. 1993. Availability of phosphate applied to calcareous soils of  
188 est Asia and North Africa. *Soil Sci Soc. Am. J.* 57:756-760.
- 189 Berrero, C., F. Pena, and J. Torrent. 1988. Phosphate sorption by calcium carbonate in some soils  
190 of the Mediterranean part of Spain. *Geoderma.* 42:261-269.
- 191 Bolan, N.S., and M.J. Hedley. 1990. Dissolution of phosphate rocks in soils 2. Effect of pH on the  
192 dissolution and plant availability of phosphate rock in soil with pH dependent charge.  
193 *Fertilizer and lime research. Massey University, New Zealand.* 125-134.
- 194 Fox, R.L., and E.J. Kamprath. 1970. Phosphate sorption isotherms for evaluating the phosphate  
195 requirements of soils. *Soil Sci. Soc. Am. Proc.* 34:902-907.
- 196 Jalali, M., and F. Ranjbar. 2010. Aging effects on phosphorus transformation rate and  
197 fractionation in some calcareous soils. *Geoderma.* 155:101-106.
- 198 Jones, C.A., C.V. Cole, A.N. Sharpley, and J.R. Williams. 1984. A simplified soil and plant  
199 phosphorus model: I. Documentation. *Soil Sci. Am.* 48:800-805.
- 200 Holford, I.C.R., and G.E.G. Mattingly. 1975. Phosphate sorption by Jurassic oolitic limestones.  
201 *Geoderma.* 13:257-264.
- 202 Larsen, S., and A.E. Widdowson. 1970. Evidence of dicalcium phosphate precipitation in a  
203 calcareous soil. *Soil Sci.* 21:364-367.
- 204 Ortas, I., and D.L. Rowell. 2000. Effect of pH on amount of phosphorus extracted by 10 mM  
205 calcium chloride from three Rothamsted soils. *Comm. in Soil Sci. and Plant Anal.*  
206 31:2917-2923.
- 207 Ozanne, P.G., 1980. Phosphate nutrition of plants—a general treatise. pp.559-589. In F.E.  
208 Khasawneh, E.C. Sample, and E.J. Kamprath (ed.). *The role of phosphorus in Agriculture.*  
209 *Am. Soc. of Agron. Madison, WI.*
- 210 Ramirez, R., S.M. Fernandez, and J.I. Lizaso. 2001. Changes of pH and available phosphorus and  
211 calcium in rhizosphere of aluminum-tolerant maize germplasm fertilized with phosphate  
212 rock. *Comm. in Soil Sci. and Plant Anal.* 32:1551-1565.

- 213 Ryan, J., D. Curtin, and M.A. Cheema. 1985. Significance of iron oxides and calcium carbonate  
214 particle size in phosphate sorption by calcareous soils. *Soil Sci. Soc. Am. J.* 49:74-76.
- 215 Schachtman, D.P., R.J. Reid, and S.M. Ayling. 1998. Phosphorus uptake by plants: from soil to  
216 cell. Department of Botany. University of Adelaide, SA, Australia.116:447-453.
- 217 Sharpley, A.N., U. Singh, G. Uehara, and J. Kimble. 1989. Modeling soil and plant phosphorus  
218 dynamics in calcareous and highly weathered soils. *Soil Sci. Soc. Am. J.* 53:153-158.
- 219 Tunesi, S., V. Poggi, and C. Gessa. 1999. Phosphate adsorption and precipitation in calcareous  
220 soils: the role of calcium ions in solution and carbonate minerals. *Nutrient Cycling in*  
221 *Agroecosystems.* 53:219-227.
- 222 Zhou, A.M., H.X. Tang, and D.S. Wang. 2005. Phosphorus adsorption on natural sediments:  
223 modeling and effects of pH and sediment composition. *Water Res.* 39:1245-1254.