

2
3 **Dynamics of Soil Chemical Properties and Plant nutrient status as influenced**
4 **by application of lime, phosphorus and compost**

5 **Abstract**

6 *Declining soil fertility along with soil acidity are major soil degradation problems affecting crop*
7 *production in Ethiopian highlands. However, little effort has been done to investigate different*
8 *soil amendment measures on soil chemical properties and nutrient status of common bean crop*
9 *on acid soils of Areka. Therefore, field studies were conducted with the objective of investigating*
10 *the effects of combined application of compost, lime and P-fertilizer on selected soil chemical*
11 *properties and plant nutrient concentration of common bean at Areka area of southern Ethiopia.*
12 *Treatments, consisted of factorial combinations of three rates of compost (0, 5 and 10 t ha⁻¹),*
13 *lime (0, 0.64 and 1.28 t ha⁻¹) and phosphorus (0, 23 and 46 kg P₂O₅ ha⁻¹) were laid out in a*
14 *randomized complete block design with three replications. Data on several soil chemical*
15 *properties and tissue nutrient status were collected. The results revealed that relative to the*
16 *control treatment, the application of 10 t ha⁻¹ of compost alone increased soil pH and nitrogen by*
17 *7 and 68% in Belg and 7 and 77% in the Meher seasons, respectively. Similarly, sole application*
18 *of lime increased soil pH by 20 and 10% in Belg and Meher seasons, respectively. The*
19 *application of compost at the rates of 5 t ha⁻¹ also resulted in a corresponding increase in leaf*
20 *tissue N concentration during Belg and Meher seasons. Further significant interaction effects of*
21 *compost × lime × phosphorus were found for soil available P and tissue P concentration in both*
22 *seasons. Combined application of compost, lime and P at their highest rates resulted in an*
23 *increase in available P by 221% in belg and 144% in meher seasons compared to the control*
24 *treatment. In conclusion, separate as well as combined application of compost, lime and P in*
25 *both seasons can improve the fertility of the soil in the study area.*

26
27 **Keywords:** Acid-soil, available-phosphorus, tissue-nutrient-concentrations, soil-fertility, soil
28 chemical properties

30

1. INTRODUCTION

31 Soil fertility depletion in Africa is one of the major problems for agricultural production.
32 Inadequate nutrient and organic matter supply constitutes the principal cause for declining soil
33 fertility and productivity in much of sub-Saharan Africa (SSA) [1]. Soil degradation primarily
34 due to intensive cultivation, nutrient depletion and soil erosion is a serious problem in Ethiopian
35 highlands [1, 2, and 3]. Major nutrients such as nitrogen (N) and phosphorus (P) are the most
36 limiting nutrients due to low input application [4]. Fertilizer application rate in the country also
37 have been reported to be minimal [4]. As a result, the average level of nutrient losses in Ethiopia
38 in 2002/2004 cropping season has been estimated at 49 kg ha⁻¹ year⁻¹ for N, P, and Potassium (K)
39 fertilizer [5], which indicates the extent of nutrient mining and decline in soil fertility mainly due
40 to inadequate input application. Soil P is also insufficient for crop growth in most agricultural
41 systems in the tropics including Ethiopia, and P must be provided as an external input due to low
42 input application, conventional tillage practices, nutrient recycling from organic sources, etc [6].

43

44 Further, soil chemical properties including acidity, salinity and nutrient concentration combined
45 with soil texture and bulk-density influence root development and nutrient uptake [7].
46 Approximately 30% of the total land area in the world consists of acid soils and as much as 50%
47 the world's potentially arable land is acidic [8], which significantly affect crop production
48 worldwide. Soil acidification is also one of the major challenging constraints to increase beans
49 productivity in Ethiopia [9]. [10] reported that out of the total acidic soils of Ethiopia about 28 %
50 are moderate to weakly acidic (pH of 5.5 - 6.7); 13 % are strong to moderately acidic (pH < 5.5)
51 and nearly one-third have aluminum toxicity problem. Soils of the study area also had strong to
52 moderate soil acidity condition [11].

53

54 Soil acidity, which is one form of soil degradation, is caused by the release of Al from Al
55 containing clay minerals upon weathering and leaching of cations from the edge of clays by
56 excess precipitation [12, 13]. Further, hydrolysis of hydroxyl-aluminum leads to increase in the
57 hydrogen ions concentration, thereby decreasing the soil pH values below seven. Moreover, mis-
58 management such as intensive use of ammonia-containing fertilizers also counted to be another
59 cause of soil acidification by decreasing the soil pH [12, 14, 15]. [13] and [16] also indicated the
60 decomposition of organic matter to be another source of soil acidity.

61
62 The major effects of soil acidity encompasses limited availability of soil nutrients such as
63 phosphorus, calcium (Ca), magnesium (Mg) and molybdenum (Mo) and toxic effects of certain
64 micronutrients such as aluminium (Al) and manganese (Mn), which often lead to reduced root
65 growth [17,15]. In acid soils, P is the major growth limiting nutrient due to fixation by clays
66 dominated with Al and iron (Fe) hydroxides [18]. As a consequence, beside to Al and Mn
67 toxicities, P limitation is indicated to be one of the most important nutrient constraints on acid
68 soils [19]. Particularly, P deficiency has been indicated to be a primary limitation to bean
69 production in developing countries [20]. The majority of Ethiopian soils are also characterized by
70 P deficiency because of high P fixation in highly weathered soils and low P of parental material
71 [21].

72
73 Research results emanating from different locations suggest combined use of organic and
74 inorganic nutrient sources to address soil fertility problems in a sustainable manner [22, 1]. There
75 are also immense research works in Ethiopia and elsewhere in the world which, demonstrated the
76 possibility of addressing soil fertility decline problems by separate application of inorganic and
77 organic nutrient sources, combined application of both inorganic as well as organic nutrient
78 sources [1, 23, 24], separate application of lime [25], combined application of lime with organic
79 nutrient sources [26, 27, 25] as well as lime with inorganic nutrient sources. [11] also suggested
80 the need of improving the fertility status of the soils of Areka owing to its low fertility in terms
81 of both micro and macronutrients status.

82
83 ~~However, little effort has been done to investigate the effects of combined or separate application~~
84 ~~of compost, P and lime on soil chemical properties of soils of Areka.~~ This study was to
85 investigate the effects of combined application of lime, P and compost on selected soil chemical
86 properties and plant tissue nutrient status.

87 88 **2. MATERIALS AND METHODS**

89 **2.1. Description of the Study site**

90 The field experiment was conducted at Areka Agricultural Research Centre (AARC), which is
91 situated in the SNNPRS between 7°3'25' north latitude and from 37°40'52'' east longitude. The

92 altitude of the experimental site reaches to 2230 meters above sea level [11]. The Center has a
93 bimodal rainfall pattern. Accordingly the first rainy season is *belg* starting from April and
94 extending to mid-July, while *Meher* season begins in late July and extends to October. The mean
95 annual rainfall for 12 years (1988 to 2000) was 1520 mm [11]. The major soil type of the center
96 is Haplic Alfisols (FAO, classification), which is very deep and clayey in texture [11].

97

98 **2.2. Description of the Experimental Material**

99 The cultivar used for this experiment was *Dinkinesh*, which was P-efficient and promoted from
100 the greenhouse experiment. Triple super phosphate (TSP) [$\text{Ca}(\text{H}_2\text{PO}_4)_2$] (21% P) and Urea [CO
101 $(\text{NH}_2)_2$] (46% N) was used for this experiment as source of P and N, respectively.

102

103 The liming materials used for this experiment were CaO and CaCO_3 . The purity of lime (CaCO_3)
104 used for field experiment was 89%, while the purity of CaO used for incubation experiment was
105 98%. Well prepared and decomposed compost was used for this experiment.

106 **2.3. Experimental Procedures**

107 **Lime Requirement Determination**

108 For determination of lime requirement of soils of Areka, lime (CaO) was applied at the rates of
109 0, 0.56, 1.12, 1.68, 2.24, 3.36 g kg^{-1} to soil samples each weighing 1 kg and were uniformly
110 mixed. Then, the soils were added in triplicate into a polythene tube with a capacity of
111 containing 1 kg soil. Water was added into the samples approximately to field capacity. The field
112 capacity was determined initially by weighing the dry soil together with the polythene tube (dry
113 soil + polythene tube) followed by watering the soil in the polythene tube to saturation. Then, the
114 soil in the polythene tube watered to saturation was left for 24hrs until all the gravitational water
115 drained out and the soil at this stage assumed to attain its field capacity. To estimate the moisture
116 content of the soil in the polythene tube at this stage (at field capacity), the wet soil (wet soil +
117 polythene tube) was weighted and the weight difference between the wet (wet soil + polythene
118 tube) and dry soil (dry soil + polythene tube) was converted into liter to know the volume of
119 water retained at field capacity. Then, the next day the soil in the polythene tube left for one day
120 without watering and reweighed and the weight difference between the moisture content at field
121 capacity and the moisture content of the soil after one day was converted into liter and taken as
122 the amount of water needed to maintain the moisture content of the soil to field capacity

123 throughout the incubation period. Hence, the amount of water estimated earlier was added to the
124 soil everyday to maintain the moisture content of the soil to the approximately to field capacity
125 throughout the incubation period. The soils in the polythene tube were thoroughly mixed and
126 incubated under room temperature for a period of two weeks. Then, the soils were ground, all
127 plant debris and root were removed and its pH measured. The relationships between the amounts
128 of CaO applied and the pH values obtained were plotted and the level of CaO sufficient to raise
129 the pH of the soil to desired pH was determined. The CaCO₃ equivalent of CaO was used for
130 subsequent liming experiment.

131 **2.4. Soil sampling, sample preparation and analysis**

132 Soil sampling and analysis was done before and after harvesting of the crops. Samples were
133 randomly collected using an auger to the soil depth of 30 cm in a zigzag pattern from the
134 experimental field before planting. The samples were mixed thoroughly in a bucket and
135 composite sample were taken to analyze soil texture, soil pH, available P, soil organic carbon
136 (SOC), cation exchange capacity (CEC) and exchangeable bases (Ca, Mg and K) following
137 standard laboratory methods and procedures. Accordingly, soil texture was determined by the
138 Bouyoucas hydrometer method [28]. Soil bulk density was be determined by the undisturbed
139 core sampling method after drying the soil samples in an oven at 105°C to constant weights [29].

140 Available P was determined by Bray I method using ammonium fluoride as extractant and
141 measuring the concentration of P as described by Bray and Kurtz [30]. Soil organic carbon was
142 determined by the Walkley-Black wet digestion method [31]. Total N was determined by wet
143 oxidation procedure of the KjeldhalKjeldhal method [32]. The CEC was determined with
144 ammonium acetate saturated samples using Na from percolating NaCl solution to replace the
145 ammonium ions. The displaced ammonium ion was measured using the modified Kjeldahl
146 procedure and was reported as CEC [33]. Soil pH was determined potentiometrically in
147 supernatant suspension of 1:2.5 soils: water ratio using a combined glass electrode pH meter
148 [34]. Exchangeable bases (Ca, Mg, and K) in the soil were estimated by the ammonium acetate
149 (1M NH OAc at pH 7) extraction method as described by [35].

150 Compost was also analyzed for determining total P, N and K after wet digestion of the samples
151 following the same procedure for soil analysis.

152

153 **2.5. Plant Sampling, Sample Preparation and Analysis**

154

155 Just before flowering three fully developed leaves at the top of the plant were sampled from
156 twenty randomly selected common bean plants per plot [36] and the samples were taken for
157 analysis of tissue P and N. The samples were analyzed following standard methods and
158 procedures as described by [37].

159 **2.5. Treatments and Experimental Design**

160 The treatments consisted of three levels of P (0, 23 and 46 kg P₂O₅ ha⁻¹), three levels of lime (0,
161 0.64 and 1.28 tons ha⁻¹) and three levels of compost (0, 5, and 10 tons ha⁻¹). The experiment was
162 laid out as a randomized complete block design in a factorial arrangement and replicated three
163 times. Each treatment was randomly assigned to each plot.

164

165 **2.6. Agronomic Practices**

166

167 Land preparation of the experimental field was done properly both during *belg* and *meher*
168 seasons. In each season, the experiment was done at different experimental fields. In other
169 words, the experiment was not repeated across *belg* and *meher* seasons on the same land. The
170 field was tractor ploughed three times before planting so as to ensure better crop emergence and
171 crop stand. Lime was applied two months prior to planting of the crop. Compost was applied in
172 broadcast one month prior to the planting of the crop. Nitrogen was applied at the rate of 18 kg N
173 ha⁻¹ in the form of urea, at the active stage of vegetative growth before flowering [38].
174 Phosphorus fertilizer was applied in band at planting.

175

176 **2.7. Data Collection**

177 Data on soil chemical properties such as soil texture, bulk density, soil pH, available P, total N,
178 soil organic matter, soil exchangeable bases (Ca, Mg and K), and CEC were collected.
179 Similarly, data on leaf tissue N, and P concentration were collected.

180

181 **2.8. Statistical Analysis**

182

183 The data of the two seasons were tested for homogeneity of variance using F-test [39]. The F-test
184 indicated that the treatment means were significantly different and heterogeneous for the
185 parameters between the two seasons. Accordingly, separate analysis was done. All data were
186 subjected to analysis of variance using SAS version 8, Statistical software (SAS Institute,
187 Cary, North Carolina, U.S.A). Significant means were separated using LSD test.

188

3. RESULTS AND DISCUSSION

189 **3.1. Pre-planting Soil Physico-chemical Properties and pH curve**

190 The physico-chemical properties of the soils of the experimental field for the surface layer (i.e.,
191 0-30 cm) are presented in Table 1. According to the ratings given by [40], the pH of the soil is
192 strongly acidic, whereas the organic matter and total nitrogen contents of the soil are high; the
193 CEC of the soil is moderate; the base saturation percent is high. However, according to the latter
194 author, the available P content of the soil is low. This shows that the soil has limitations in soil
195 pH and P availability for crop production. Accordingly, managing soil pH and P availability is
196 important for enhancing plant growth and production in the study area.

197

198

199 Table 1. Physico-chemical properties of the study area before planting during *Belg* and *Meher*
 200 seasons

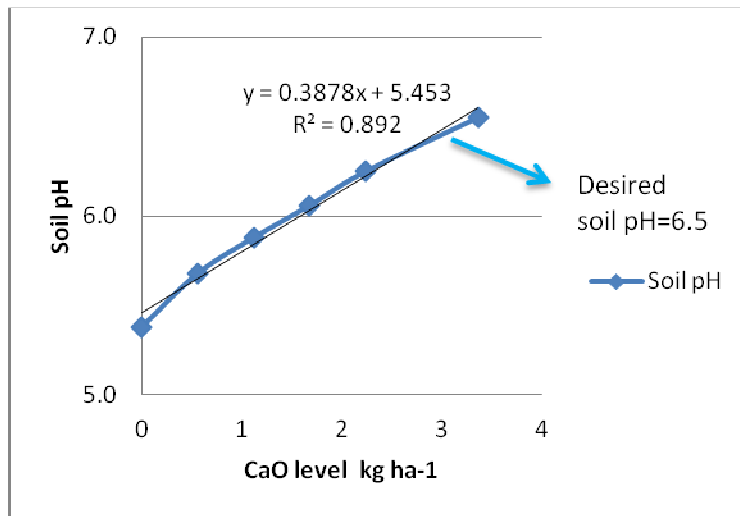
Seasons	p ^H	Total N (%)	OC (%)	OM (%)	P (mg Kg ⁻¹)	Exchangeable Cations cmol(+)/kg soil				CEC cmol _c (+)/kg soil
						Available				
						Na	K	Ca	Mg	
<i>Belg</i>	5.29	0.2	2.9	5.1	3.5	0.1	0.1	10.1	11.3	26.9
<i>Meher</i>	5.08	0.2	2.7	4.7	3.2	0.1	0.2	14.5	1.1	24.7

201 OC-Organic carbon, OM- Organic matter

202 Table 2. Chemical properties of compost used for the experiment during *Belg* and *Meher* seasons

Seasons	p ^H	N (%)	P (%)	K (%)	Ca (%)	Mg (%)
<i>Belg</i>	6.2	1.7	0.4	0.1	0.7	0.3
<i>Meher</i>	6.0	1.5	0.4	0.1	0.8	0.3

203



204

205 Figure 1. Soil pH plotted against CaO level (g kg⁻¹)

206

207 Based on the pH curve, the CaO level needed to raise the soil pH to the desired pH level *i.e.*, 6.5
208 was found to be 2.76 g kg⁻¹ (Figure 1). Accordingly, the CaCO₃ equivalent of CaO to obtain the
209 optimum pH of 6.5 was 5.426 g CaCO₃ (lime) kg⁻¹ soil, which is equivalent to 1.28 t ha⁻¹
210 assuming 89% purity of lime, 20 cm plough depth and 1.18 g cm⁻³ soil bulk density.

211 **3.1. Status of available phosphorus**

212
213 Combined application of compost, lime and P significantly ($P < 0.001$) increased soil available P.
214 Generally, an increasing trend in soil available P was observed across the compost and lime as
215 well as P rates in both seasons. The values in *belg* season ranged from 4.17 to 13.4 mg/kg, while
216 in *meher* season it ranged from 4.56 to 10.7 mg/kg. In both seasons, the highest available P was
217 obtained when the highest rates of compost, lime and P were applied compared to plots applied
218 with neither of these soil amendments. Hence, combined application of compost, lime and P at
219 rates of 10 t, 1.28 t and 46 kg P₂O₅ ha⁻¹, increased available P by more than two and one fold in
220 *belg* and *Meher* seasons respectively, compared to the control (Table 3). Further, without
221 application of lime and compost, application of P alone at rates of 46 kg P₂O₅ ha⁻¹ increased soil
222 available P nearly by 30 and 27% in *belg* and *meher* seasons, respectively compared to the
223 control (Table 3).

224
225 Different researchers reported a significant increase in available P as a result of application of
226 lime, compost and inorganic P [41, 42, and 26]. Hence, the increase in soil available P is perhaps
227 due to combined effect of the active CaO component of lime in releasing fixed and precipitated P
228 by correcting the pH [42, 26]. Similarly, the increase in concentration of soil solution P also
229 related to the direct addition of P from the applied inorganic P as well as nutrients released upon
230 mineralization from the applied compost [25, 43]. Further, liming might have enhanced
231 mineralization of organic matter added as compost thereby facilitating the release of some of the
232 essential macronutrients such as P [44]. Corroborating the results, [26] also reported that
233 combined application of lime and manure increased available P on acid soils of North Western
234 Ethiopia. In a similar study, [45] also reported high values of available P in different locations
235 ranging from 8.44 to 10.30 mg kg⁻¹ as a result of combined application of the highest rates of
236 lime and compost on acid soils.

237

238 Table 3. Interaction effect of compost, lime and P on available P during *belg* and *meher* seasons

Compost (t ha ⁻¹)	Lime (t ha ⁻¹)	Phosphorus rates (kg P ₂ O ₅ ha ⁻¹)					
		0		23		46	
		Available phosphorus (ppm)					
		<i>Belg</i>	<i>Meher</i>	<i>Belg</i>	<i>meher</i>	<i>Belg</i>	<i>meher</i>
0	0	4.17 ⁿ	4.56 ^r	4.76 ^{nm}	5.23 ^p	5.40 ^{lkm}	5.8 ^{on}
	0.64	4.67 ^{nm}	4.9 ^q	5.27 ^{lkm}	5.6 ^o	6.03 ^{kj}	6.63 ^{mk1}
	1.28	5.23 ^{lkm}	5.23 ^p	6.03 ^{kj}	5.83 ^{on}	6.57 ^{ihj}	6.40 ^m
5	0	5.13 ^{lm}	6.47 ^{ml}	5.80 ^{lkj}	6.97 ^{ji}	7.16 ^{igh}	7.46 ^g
	0.64	6.37 ^{ij}	6.0 ⁿ	7.26 ^{ghf}	6.60 ^{mk1}	8.43 ^{ed}	7.16 ^{hi}
	1.28	7.57 ^{egf}	6.70 ^{ikl}	7.97 ^{egdf}	7.40 ^{hg}	8.46 ^d	8.07 ^f
10	0	7.3 ^{ghf}	6.83 ^{jk}	8.10 ^{edf}	7.37 ^{hg}	9.90 ^c	8.0 ^f
	5	8.27 ^{ed}	8.0 ^f	9.70 ^c	9.0 ^d	9.67 ^c	10.17 ^b
	10	9.83 ^c	8.37 ^e	11.03 ^b	9.50 ^c	13.40 ^a	10.7 ^a
LSD (0.05) =		0.8668	0.2763	0.8668	0.2763	0.8668	0.2763

239 Means followed by the same letters are not statistically different at 5% probability level

240 3.2. Effect on soil pH

241 Mean values for soil pH showed a significant increase as a result of the interaction effect of
 242 compost and lime in *Belg and Meher* seasons (Table 4). The values for soil pH ranged from 5.2
 243 to 6.94 in both seasons. The highest value for soil pH (6.94) in was recorded in *Meher* season,
 244 while the lowest (5.2) value recorded in *Belg* season. However, the highest value recorded during
 245 *Meher* season was statistical at par to the values recorded during *Belg* season. The highest value
 246 was recorded as a result of application of compost at rates of 10 t ha⁻¹ along with 1.28 t ha⁻¹ lime,
 247 which increased soil pH by about 30% in *Belg* season and 33.5% in *Meher* season compared to
 248 the control (Table 4).

249

250

251 Table 4. Soil pH as influenced by interaction effect of compost and lime in *Belg* and *Meher*
 252 seasons of 2013

Compost rates (t ha ⁻¹)	<i>Belg</i>			<i>Meher</i>		
	Lime rates (t ha ⁻¹)			Lime rates (t ha ⁻¹)		
	0	0.64	1.28	0	0.64	1.28
0	5.27 ^e	6.03 ^c	6.09 ^c	5.2 ^e	5.88 ^d	6.25 ^c
5	5.22 ^e	6.14 ^c	6.53 ^b	5.8 ^d	6.25 ^c	6.51 ^b
10	5.60 ^d	6.21 ^c	6.83 ^a	5.9 ^d	6.44 ^c	6.94 ^a
LSD (0.05)=	0.1986			0.1351		

253 Means followed by the same letter are not statistically different at 5% probability level.

254 Compost fertilization has an important benefit through its liming effect by adding Ca mainly in
 255 the form of calcium carbonate [46], which might have contributed to increases in soil pH.
 256 Further, addition of lime increases soil pH through the Ca added from the lime, which react with
 257 H⁺ in the exchange site and neutralizes it, thereby increasing the pH of the soil [47].
 258 Consequently, lime and compost application might have contributed to improvement in soil pH
 259 due to the liming effect of compost and lime through the Ca added to the soil from both. In other
 260 words, the rise in soil pH as a result of application of compost and lime might be ascribed to the
 261 release of calcium ions into the soil solution through mineralization of the applied compost,
 262 which in turn hydrolyzed to form calcium hydroxide and the calcium hydroxide formed reacts
 263 with Al³⁺ ion in the soil solution to give insoluble Al (OH)₃, and further the hydroxide of the
 264 calcium hydroxide reacts with hydrogen ions to form water thereby decreasing H⁺ ion and lower
 265 the pH [22] following application of compost. In a similar manner, the likely increased Ca²⁺ ions
 266 due to lime application might have contributed to increase in soil pH, which in turn react with
 267 Al³⁺, H⁺, and Fe³⁺ ions prevalent in acid soils [48], which otherwise aggravate the soil acidity
 268 condition. Furthermore, the same authors obtained the highest pH of 6.00 with the combined
 269 application of lime and 5 t ha⁻¹ compost.

270

271 **3.3. Total nitrogen**

272 The results revealed that compost application significantly ($P<0.001$) increased soil N in *Belg*
 273 *and Meher* seasons. Consequently, in comparisons to the control, application of compost
 274 improved soil total N. During the *Belg* season, application of compost at rates of 5 and 10 t ha⁻¹
 275 increased total N by 32 and 64%, respectively compared to the control treatment (Table 5).
 276 Similarly, in the *Meher* season, application of compost at rates of 5 and 10t ha⁻¹ increased total N
 277 by about 35 and 77%, respectively (Table 5).

278 Table 5. Total nitrogen as influenced by main effects of application of compost during *Belg* and
 279 *Meher* seasons

Compost (t ha ⁻¹)	Soil parameters			
	N (%)		Organic Carbon (%)	
	<i>Belg</i>	<i>Meher</i>	<i>Belg</i>	<i>Meher</i>
0	0.19 ^c	0.17 ^c	2.84 ^c	2.78 ^b
5	0.25 ^b	0.23 ^b	2.98 ^b	2.83 ^{ba}
10	0.32 ^a	0.30 ^a	3.07 ^a	2.89 ^a
CV (%)	6.24	4.2	3.48	4.19
LSD (0.05) =	0.0086	0.054	0.0564	0.065

280 Means followed by the same letter are not statistically different at 5% probability level.

281

282 Soil N increased with increase in the levels of compost, which might be related to the release of
 283 N from the applied compost through mineralization. Corroborating the results, [1] reported
 284 highest increase in soil N from plots applied with compost and farm yard manure. Similarly, the
 285 results agree with the findings of [49], who recorded the highest nitrogen value (0.42 g Kg⁻¹) for
 286 plots applied with 8 t ha⁻¹ poultry manure on an Ultisol of Southeastern Nigeria.

287 **3.4. Soil potassium**

288 Experimental results revealed that application of compost significantly ($P<0.001$) increased soil
 289 K in both seasons. Accordingly, soil K increased due to the application compost in both seasons

290 (Table 6). For instance, during the *Belg* season, 56 and 80% increase in soil K compared to the
 291 control was recorded due to application of compost at rates of 5 and 10 t ha⁻¹, respectively.

292 Table 6. Exchangeable K as influenced by application of lime and compost during *Belg* and
 293 *Meher* seasons

Compost (t ha ⁻¹)	K (%)	
	<i>Belg</i>	<i>Meher</i>
0	0.25 ^c	0.15 ^c
5	0.39 ^b	0.16 ^b
10	0.45 ^a	0.18 ^a
Lime		
(t ha ⁻¹)		
0	0.37	0.40
0.64	0.36	0.40
1.28	0.36	0.40
CV (%) =	3.7	2.3
LSD(0.05)=	0.006	0.010

294 Means followed by the same letter are not statistically different at 5% probability level.

295
 296 The improvement in soil K might be linked to release of the nutrient from the applied compost
 297 upon decomposition [50, 24]. In agreement with the results, [51] reported significant ($P < 0.001$)
 298 increase in exchangeable soil K as a result of application of manure as compared to control in
 299 one of their experimental year 2013.

300 **3.6. Leaf nitrogen**

301 Compost application significantly ($P < 0.001$) increased leaf tissue N concentration in both
 302 seasons. Leaf tissue N concentration improved when the rate of compost increased in both
 303 seasons (Table 8). Accordingly, increasing the rate of compost from nil to 5 t ha⁻¹ resulted in a
 304 corresponding 4 and 5% increase in leaf N concentration during *belg* and *meher* seasons.

305 Table 8. Effects of compost on leaf tissue nitrogen concentration during the *belg* and the *meher*
 306 seasons

Compost (t ha ⁻¹)	Leaf tissue N concentration (%)	
	<i>Belg</i>	<i>meher</i>
0	3.92 ^b	3.81 ^c
5	4.09 ^a	4.0 ^b
10	4.16 ^a	4.15 ^a
CV (%) =	3.32	3.38
LSD (0.05) =	0.074	0.074

307 Means followed by the same letter in the same column are not statistically different at 5%
 308 probability level.

309
 310 The increase in leaf tissue N concentration might be attributed to the increased supply of N from
 311 the applied compost through mineralization and subsequent uptake by the crop [50]. In
 312 agreement with the result, [52] reported significant increase in cabbage leaf N concentration. The
 313 mean values recorded for leaf tissue N concentration are in agreement with the optimum range
 314 reported for legumes [53].

315 3.7. Leaf tissue P concentration

316 Mean values for leaf tissue P concentration showed significant increase due to combined
 317 application of compost, lime and P in both seasons (Table 9). The highest tissue P concentration
 318 in both seasons was obtained by applying compost, lime and P at corresponding rates of 10 and
 319 1.28 t ha⁻¹ and 46 kg P₂O₅, respectively, which exceeded the control by nearly one fold in both
 320 season (Table 9).

321
 322
 323
 324
 325

326 Table 9. Interaction effect of compost, lime and phosphorus on leaf tissue phosphorus
 327 concentration during the *belg* and the *meher* seasons

Compost (t ha ⁻¹)	Lime (t ha ⁻¹)	Phosphorus (kg P ₂ O ₅ ha ⁻¹)					
		0		23		46	
		Leaf tissue P concentration (%)					
		<i>Belg</i>	<i>meher</i>	<i>Belg</i>	<i>meher</i>	<i>Belg</i>	<i>Meher</i>
0	0	0.15 ⁿ	0.13 ^k	0.20 ^{kjl}	0.21 ^{igh}	0.21 ^{ijh}	0.25 ^{ced}
	0.64	0.17 ^{nm}	0.20 ^{ij}	0.21 ^{ijh}	0.22 ^{igh}	0.22 ^{ijh}	0.24 ^{fged}
	1.28	0.17 ^{nm}	0.19 ^j	0.21 ^{ijh}	0.23 ^{fge}	0.24 ^{fegd}	0.25 ^{cbd}
5	0	0.18 ^{ml}	0.20 ^{ihj}	0.20 ^{kij}	0.22 ^{fgh}	0.22 ^{fgh}	0.24 ^{fed}
	0.64	0.18 ^{kml}	0.19 ^j	0.25 ^{feed}	0.24 ^{fed}	0.26 ^{beecd}	0.25 ^{ced}
	1.28	2.0 ^{lmn}	0.19 ^{ij}	0.26 ^{bcd}	0.24 ^{fed}	0.28 ^{ba}	0.26 ^{ij}
10	0	0.23 ^{hfg}	0.24 ^{fed}	0.24 ^{hfg}	0.24 ^{fed}	0.24 ^{hfg}	0.24 ^{fed}
	0.64	0.19 ^{mn}	0.20 ^{ihj}	0.24 ^{hfg}	0.25 ^{ced}	0.27 ^{bc}	0.27 ^{cb}
	1.28	0.20 ^{lmn}	0.21 ^{igh}	0.27 ^{bc}	0.28 ^b	0.29 ^a	0.31 ^a
LSD (0.05) =		0.0196	0.0235				

328 Means followed by the same letter are not statistically different at 5% probability level.

329

330 The tissue P concentrations reported in the present study are similar to leaf tissue P concentration
 331 reported elsewhere for common bean [18]. The increase in leaf tissue P concentration is attributed
 332 to improved soil P status as a result of combined application of compost, lime and P [50]. In
 333 agreement with the results, [54] obtained significantly higher shoot P concentration in maize as a
 334 result of integrated application of organic amendments with DAP.

335

4. CONCLUSION

336 The results of this study have shown that application of compost, combined application of
 337 compost with lime, as well as combined application of (compost, lime and P) significantly
 338 improved soil chemical properties and levels of nutrients in plant tissues. The improvement in
 339 soil pH due to combined application of compost with lime and increase in available P as a result
 340 of application of (compost, lime and P) is an indicator of the potential of the soil amendments for

341 correcting the soil fertility problem in the study area. ~~Further, this implies that the soil of the~~
342 ~~study area had soil fertility problems related to soil acidity as well as soil nutrient depletion.~~
343 Hence, it can be concluded that application of compost, lime and P in combination or separately
344 has the potential to mitigate soil fertility problems at Areka.

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347 **Conflict of Interest**

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Authors have declared that no competing interests exist

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