1 Original Research Article 2 Gas Chromatographic FID Evaluation of roots of Manniophytum 4 fulfum 5

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8 Abstract

The phytochemical composition of the root extract of *Manniophyton fulvum* was 9 screened by gas chromatography coupled to flame ionization and pulse flame 10 11 photometric detectors. Among the twenty four known flavonoids were detected, only quercetin (107.44mg/100g), kaemoferol (66.19mg/100g), quercetrin (49.87mg/100g), rutin 12 (28.52mg/100g) and, apigenin (28.67mg/100g) representing (37.99%,23.41%, 10.14%, 13 17.64%, and 10.09%) respectively were prominent. The rest were in insignificant amount. 14 15 The roots extracts contain nine glycoside compound consisting mainly of digoxin 16 (27.166mg/100mg 58.5%), digitoxin (17.9 mg/100g (38.51%) and 0.8mg/100g (2.99%) of 17 kampferol-3-rhamnoside. The rest were of a very little amount. For saponin, the FID gas chromatography analysis revealed that euphol, saponine, sapogenin had the highest 18 19 concentration of 20.268490mg/100g, 0.3369mg/100g, and 0.065978mg/100g making up of 20 97.9%,1.62% and 0.3% percentage composition respectively. The rest were in insignificant 21 amount. Seven benzoic acid were detected by the GC- FID consisting mainly of p-hydroxyl 22 benzoic 6.288160mg/100g (65.4%), gallic acid 2.96280 mg/100g (30.8%) and gentisic acid 23 0.355 mg/100 g (3.71%). Out of the ten carotenoid detected, the highest concentration were 24 violaxanthine 47.80526 (49.95%), astraxanthine 26.62538, (27.82%), lutein 8.24911 (8.93%) 25 and neoxanthine 6.54195mg/100g. (6.84%). For phytosterol, the bioactive compound include sitosterol 14.2007mg/100g(71.53%), campesterol 3.54mg/100g (17.85%) and stigmasterol 26

2.1093mg/100g (10.6%) making up highest concentration. Three allicin compound were
detected which include diallylthiosulphinate (95.92%), methyl allyl thiosulphinate (3.50%)
and allyl methyl thiosulphinate (0.58%). These results shows that *Manniophyton fulfum* has
many bioactive molecules suggesting strong therapeutic and neutraceautical potential
suggesting their likely uses in the management and prevention of diseases.

32 Keyword: Gas chromatographic FID, phytochemical, maniophyton fulfum, neutraceatical.

33 Introduction

34 Nature has been a source of medicinal agents for thousands of years and an impressive number of modern drugs have been derived from natural sources, with many of these 35 36 isolations based on the uses of the agents in traditional medicine (Cragg and Newman, 2001). 37 Phytomedicine also known as botanical medicine or herbal medicine is a branch of science in 38 which plant based formulations are used to alleviate diseases. Recently there has been a shift 39 in universal trend from synthetic to herbal medicine, which can be termed 'Return to Nature''. 40 Medicinal plants have been known for millennia and highly esteemed all over the world as a rich source of therapeutic agents for the prevention of diseases and ailments (Sharma et al., 41 42 2008). The search for eternal health and longevity and for remedies to relieve pain and discomfort drove early man to explore his immediate natural surroundings and led to the use 43 44 of many plants, animal products, minerals, etc and the development of a variety of 45 therapeutic agents (Nair et al., 2007).

The importance of medicinal plants and traditional health system in solving health care problems of the world is gaining increasing attention. Because of this resurgence of interest, the research on plants of medicinal importance is growing phenomenally at the international level, often to the detriment of natural habitats and mother populations in the countries of origin. Most of the developing countries have adopted traditional medicine as an integral part of their culture. Historically, most medicinal preparations are derived from plants, whether in
the simple form of raw plant materials or in the refined form of crude extracts, mixers, etc.
(Krishnaraju *et al.*, 2005).

54 In the bid to maximize the potentials of phytomedicine, there is need to investigate the 55 chemical composition and biological activities of medicinal plants. The product of the 56 various investigations carried out in the past two centuries have yielded compounds for the development of modern synthetic organic chemistry as a major route for discovery of novel 57 and more effective therapeutic agents (Nair et al., 2007). Despite all efforts, out of an 58 59 estimate of about 250,000 flowering plants occurring on the planet earth, 60% have not been 60 screened for biological activity. Nigeria is blessed with abundance of medicinal plants with 61 vast therapeutic potentials which phytochemistry has not been elucidated. Such medicinal plant include Manniophyton fulvum. 62

63 Manniophyton fulvum remains one of the popular herbs amongst local traditional medicine practitioners in the south-south region of Nigeria. It belongs to the family Euphorbiaceae. It 64 65 is also geographically distributed widely in tropical Africa, from Sierra Leone to Sudan, and South-ward to Angola (Brown et al., 1913). In African Traditional medicine the root, stem, 66 bark and leaf are credited with analgesic properties, and are used to treat diarrhea, stomach 67 68 ache, cough and bronchitis (Burkil 1985). According to unconfirmed folklore in the south-69 south region of Nigeria, traditional herbalists have used the root extract of Manniophyton 70 fulvum to treat erectile dysfunctio n. The red stem sap is credited with haemostatic properties, 71 while the leaf sap is used against ear problems, caries (Harbone 1973). In Congo 72 (Brazzaville), it is considered a cicatrisant on wounds, and also good for treating dysentery, 73 piles, haemophthysis and dysmenorrhea (Bouquet and Debray, 1974; Bouquet, 1969). The 74 red stem-sap is used topically in Ivory Coast on herpes and other dermal infections. Furthermore, a decoction of the young shoots, bark and stem, the husk of the nut and the sap 75

<mark>76</mark>	are used as a	remedy for	cough in	n Congo	(Brazzaville),	Ivory	coast	and	Sierra	Leone
77	(Bouquet and D	Debray, 1974;	Bouquet	, 1969).						

78	In recent researches, the antioxidant and antidiarrheal property of the plant have been
79	determined and the result was positive (Ezeigbo et al., 2010). The leaf extract of the plant
80	was found less toxic than the root extract, being that the dosage required to cause death in
81	50% of the animals using the root extract is about 850 mg/kg whereas the leaf extract
82	produced toxicity at a dose of 1050 mg/kg (Agbaire et al., 2013).

83 MATERIALS AND METHODS

84 Chemical used.

- All chemical and drugs were obtained commercially and were of
- 86 analytical grade.
- 87

88 Collection of plant samples

- 89 Roots of Manniophyton fulvum were obtained from Umuada Ngodo Isuochi,
- 90 Umunneochi LGA, Abia State, Nigeria. For easy identification, the fresh leaves
- and fruits of the plant was also harvested. After identification by the Taxonomy
- 92 Unit, Department of Plant Science and Biotechnology, Faculty of Biological
- 93 and Physical Sciences, Abia State University, Uturu, the identity was confirmed
- 94 and authenticated by a plant taxonomist, Dr Edwin Wosu, Department of Plant
- 95 Science and Biotechnology, Faculty of Biological Sciences, College of Natural
- 96 and Applied Sciences, University of Port Harcourt, Nigeria and sample
- 97 specimens where deposited at the University Herbarium. Voucher number
- 98 (UPH No): V 1,035 and carpological number (UPH No): C 056 were
- 99 assigned to the leaves and fruits respectively.

101 Analysis of the phytochemical profile

102 Calibration, identification and quantification

103 The linearity of the dependence of response on concentration was verified by regression 104 analysis. Identification was based on comparison of retention times and spectral data with 105 standards. Quantification was performed by establishing calibration curves for each 106 compound determined, using the standards.

107 DETERMINATION OF FLAVONIODS

108 (Ros. J. Med sci. 3(2): 70-74, 2009)

The dried extract of ethanolic and aqueous extraction, was sampled and made to be free of water by ensuring constant weight for a period of time in the laboratory. 1.09g of the sample was weighed into the 250ml conical flack capacity with addition of 100ml of distilled water and boiled for 10minutes. The flavoniods extract was obtained by pouring 100ml of the boiling methanol;water (70:30) v. v into the materials. The mixture was allowed to macerate for about 6 hours and then concentrated to 5ml for gas chromatography analysis.

115 Determination of the sterol composition

116 Extraction

Sterol analysis was carried out by following the modified AOAC 970.51 official methods. The aliquot of the extracted oil was added to the screw-capped test tubes. The sample was saponified at 95 °C for 30 min, by using 3 mL of 10% KOH in ethanol, to which 0.20 mL of benzene had been added to ensure miscibility. Deionised water (3 mL) was added and 2 mL of hexane was used in extracting the non-saponifiable materials (sterol, etc.). Three extractions, each with 2 mL of hexane, were carried out for 1 h, 30 min and 30 min respectively, to achieve complete extraction of the sterols. The hexane was concentrated to 1

mL in the vial for gas chromatography analysis and 1 μL was injection into the injection port
of GC.

126 Determination of the saponin composition

127 Extraction

128 The extraction was carried out by following the modified method of Guo et al. (2009). The 129 sample was pulverized and the saponin was extracted three times with redistilled methanol. 130 The saponins were removed with 20 mL of the solvent for 20 min with the aid of the 131 sonication. The combined in extracts were concentrated to syrup under reduced pressure, and 132 then suspended in water. The suspension was extracted with petroleum ether, chloroform and 133 1- butanol saturated with water, successively, to give the respective extract after removal of 134 the solvent. The combined extract was filtered and concentrated to 1 mL in the vial for gas 135 chromatography analysis and 1 μ L was injected into the injection port of GC.

Determination of the allicin composition

137 Extraction

The allicin extraction was carried out by following the modified method Chehregani et al. (2007). The fresh samples (5.0 g) were washed, mined and added to adequate amount of water to a concentration of 12.5% (w/v). The sample was grinded in with a blender of make JANKE & K UNKEI GmbH & Co. KG IKA labortechnik. The products were squeezed through gauze cloth to remove the larger particles and the extracts were passed through a 0.2 µm filter (Millipore). The process was carried out at room temperature and the filtrate was sterilized and stored at 4 °C before analysis.

145 The extract was re-extracted with petroleum ether and later concentrated by pouring the 146 petroleum ether extract into the round bottom flask of the rotator evaporator arrangement. It

- 147 was concentrated by evaporation. Then the concentrated extract was dried of water by using
- the hydrous sodium sulphate before gas chromatography analysis.

149 Determination of the carotenoids composition

150 Extraction

151 The carotenoids extraction was carried out by following the modified method Takagi (1985). 152 The pulverized sample (5.0 g) was homogenized in 75 mL acetone and kept at room 153 temperature for 1 h in the dark. The homogenate was filtered through filter paper by suction. 154 Extraction was repeated three times with the same volume of acetone. The extracts were 155 combined and evaporated under reduced pressure and the residue was re-extracted by a 156 mixture of diethyl ether and petroleum ether in equal ratio. The extract was poured into the 157 round bottom flask of the rotator evaporator arrangement. It was concentrated by evaporation. 158 Then the concentrated extract was dried of water by using the anhydrous sodium sulphate 159 before gas chromatography analysis.

160 Determination of the glycoside composition

161 Extraction

The extraction was carried out by following the modified method of Oluwaniyi and Ibiyemi (2007). The pulverized sample (10 g) was extracted by soaking for 2 h with 10 mL of 70% alcohol and then filtered, concentrated. The redistilled hexane was used to replace the initial solvent and the hexane was concentrated to 1 mL in the vial for gas chromatography analysis and

167 Determination of the benzoic acid derivatives composition

168 Extraction

BDA content was extracted following modified method of Andary et al. (2013). Two stage

170 extraction procedures followed for the effective removal of BAD as described below.

171 Stage 1:

The sample (50.0 mg) was extracted with 5 mL of 1 M NaOH for 16 h on a shaker at ambient temperature. After extraction, the sample was centrifuged (5000 x g), rinsed with water, centrifuged again, and the supernatants were combined and placed in a disposable glass test tube and heated at 90 °C for 2 h to release the conjugated phenolic compounds. The heated extract was cooled, titrated with 4 M HCl to pH <2.0, diluted to 10 mL, with deionized water, and centrifuged to remove the precipitate. The supernatant was saved for subsequent purification and the residue was extracted further in stage 2.

179 Stage 2:

The residue from stage 1 above was extracted with 5 mL of 4 M NaOH, heated to 160 °C in Teflon. After cooling, the mixture was filtered. Supernatant was collected and the residue washed with water (deionised). The supernatants were combined and adjusted to pH <2.0with 4 M HCl. The filtrates were combined for further purification.

184 **Preparation for GC-FID**

The concentrated extract (2.0 mL) was transferred to 5.0 mL glass vial. The extract was saturated with sodium chloride slat before the addition to 250.0 μ L of ethyl acetate. The mixture was agitated manually for 10.0 min at room temperature and later centrifuged for 15 min at 2500 rpm. The organic phase was removed to a 1 mL vial. The extraction was repeated twice and the organic phases were mixed together. 50.0 μ L of the N, O-Bis (trimethylsilyl) trifluoroacetamide were added and the mixture was manually agitated for 2 min at room temperature for derivatization.

192 Chromatographic conditions

193 The gas chromatograph was an HP 6890 (Hewlett Packard, wellington, DE,USA), GC 194 apparatus, fitted with flame ionization detector (FID), powered with HP Chemstation Rev.A 195 09.01(1206) software, to identify and quantify compounds. The column was a capillary DB-196 5MS (30mx0.25mmx0.25µm thickness). The inlet and detection temperatures were 250°C 197 and 320°C. Split injection was adopted with a split ratio of 20:1. The Carrier gas was nitrogen 198 gas .The compressed air and hydrogen pressures were 38 psi and 28psi. The oven 199 programmed was; initial temperature at 60°C for 5 mins. First ramping at 10°C/min for 20 200 min was followed by a second ramping at 15°C /min for 4 min.

201 **Result**

202 Table 1 Flavonoid composition of *Manniopyton fulvum* roots determined by gas203 chromatography

Compounds	Retention	time Composition ×	10 ⁻⁴
	(min)	(mg/100g)	
Catechin	13.535	0.020	
Resveratrol	14.915	0.178	
Apigenin	16.034	286679.900	
Diadzein	16.243	1296.100	
Butein	16.449	1.738	
Naringenin	16.668	4.460	
Baiochanin	17.358	1.803	
Luteolin	17.768	18588.600	
Kaemferol	18.050	661903.900	
(-) – Epicatechin	19.396	5.157	
Salvagenin	20.466	12.478	
(-) – Epicatechin–3–gallate	21.668	0.040	
Gallocatechine	22.065	2.062	
Quercetin	22.600	1074141.300	
Isorhamnetin	23.738	24.762	
Myricetin	24.609	3.454	
Sinensetin	24.998	14.740	
Kaempferol-3-arabinoside	25.359	0.913	
Naringenin	26.059	2.413	
Quercitrin	27.287	498662.700	

Total composition		2827460.700
Isoorientin	28.524	96.121
Rutin	28.092	285225.200
Orientin	28.094	0.199
Isoquercetin	27.426	791.555

206 Table 2 Glycoside composition of *Manniopyton fulvum* roots determined by gas207 chromatography

Compounds	Retention (min)	time Composition × 10 (mg/100g)	0-3
Kampferol-3-O-rhamnoside	16.038	635.858	
Arbutin	17.469	353.984	
Salicin	18.765	57.267	
Amyygdalin	19.521	0.300	
Ouabain	20.473	2.259	
Digitoxin	21.824	17897.140	
Vitexicarpin	22.242	241.466	
Digoxin	22.606	27166.900	
Costugenin	23.970	120.116	
Total composition		46475.290	

Compounds	Retention	time Composition × 10
	(min)	(mg/100g)
Gitogenin	17.899	36.183
Solagenin	18.607	7.388
Diaosgenin	19.501	229.581
Tigogenin	19.935	20.223
Neohecogenin	21.020	3.071
Hecogenin	21.793	2.467
Sapogenin	23.239	659.776
Euphol	25.067	202684.910
Saponine	25.559	3369.490
Total composition		207013.190

Table 3 Saponin composition of *Manniopyton fulvum* roots determined by gas
 chromatography

Compounds	Retention (min)	time	Composition (mg/100g)	×	10 ⁻³
Gentisic acid	4.931		354.843		
Protocatechuic acid	8.760		0.351		
p–Hydroxybenzoic acid	10.940		6288.160		
Gallic acid	13.776		2962.840		
2,3 – Dihydroxybenzoic acid	16.555		0.229		
2,3,4 – Trihydroxybenzoic acid	18.697		3.408		
2,4,6 – Trihydroxybenzoic acid	19.793		6.256		
Total composition			9616.090		

215 Table 4 Benzoic acid composition of Manniopyton fulvum roots determined by gas

216 chromatography

217

218

219 Table 5 Carotenoid composition of Manniopyton fulvum roots determined by gas

220 chromatography

Compounds	Retention	time	Composition	×	10 ⁻²
	(min)		(mg/100g)		
Malvidin	19.808		20.099		
Carotene	21.229		38.545		
Lycopene	21.547		0.190		
Beta – cryptoxanthin	22.183		28.893		
Lutein	22.586		824.911		
Zeaxanthin	24.161		269.864		
Antheraxanthin	24.733		1.145		
Astraxanthin	25.780		2662.538		
Violaxanthin	26.400		4780.526		
Neoxanthin	27.873		654.195		
Total composition			9570.908		

221

222

223 Table 6 Phytosterols composition of Manniopyton fulvum roots determined by gas

224 chromatography

Compounds	Retention (min)	time Composition (mg/100g)	×	10 ⁻⁴
Cholesterol	19.151	0.009		
Cholestanol	20.533	0.933		
Ergosterol	21.442	0.995		

Campesterol	21.820	35436.600	
Stigmasterol	23.114	21092.500	
5-avenasterol	23.968	0.618	
Sitosterol	25.090	142006.700	
Total composition		198538.355	

Table 7 Allicin composition of *Manniopyton fulvum* roots determined by gaschromatography

Compounds	Retention (min)	time	Composition (mg/100g)	×	10 ⁻⁸
Diallyl thiosulphinate	16.203		1232.810		
Methyl allyl thiosulphinate	17.206		45.027		
Allyl methyl thiosulphinate	18.208		7.478		
Total composition			1285.310		

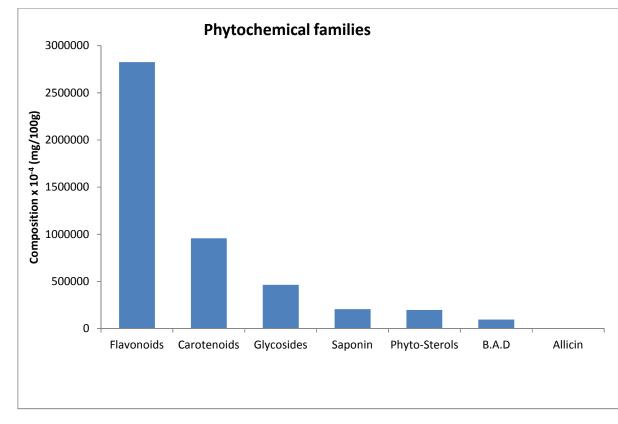


Figure 4.1 Summary of the composition of phytochemicals in *Manniophytun fulvun* roots

234 DISCUSSION

235 The results obtained from the phytochemical profile of this study revealed that the aqueous 236 root extract of *Manniophyton fulvum* contains bioactive agents. These agents are flavonoids, 237 glycosides, saponins, benzoic acid derivatives, carotenoids, phytosterols and allicins; with 238 flavonoids being the most abundant and allicins the least. Some of the main compounds of 239 the plant have been shown in various studies (Guardia et al., 2001; Hou et al., 2005; Giuliani 240 et al., 2008; Mackraj et al., 2008; Nogueira et al., 2011; Ather et al; 2007;). Further statistical 241 analysis show that the most occurring phytochemical compounds present are quercetin. 242 kaempferol, quercitrin, viola-xanthin, apigenin, rutin and digoxin. These compounds may be credited for the medicinal properties of the aqueous root extract of Manniphyton fulvum. 243

244 The roots of *manniphyton fulfum* has high Flavonoids, such as quercetins. Flavoniods had 245 previously been reported to have multiple biological properties including anti-microbial, 246 cytotoxicity, anti-inflammatory as well as anti-tumour activities. But the best described 247 properties of almost all the group of flavonoid is ability to act as a powerful anti-oxidants 248 which can protect the body against free radicals reactive oxygen species. (Tapas et al., 2008; 249 Atmani et al., 2009) .For instance querctins are antioxidants (Bando et al., 2010), they 250 scavenge damaging particles in the body known as free radicals, which damage cell 251 membranes, tamper with DNA, and even cause cell death. Antioxidants can neutralize free 252 radicals and may reduce or even help prevent some of the damage they cause (Boota et al., 253 2008).

Quercetin acts like an antihistamine and an anti-inflammatory (Guardia *et al.*, 2001), and helps to protect against heart disease (Egert *et al.*, 2009) and cancer. Quercetin can also help to stabilize the cells that release histamine in the body and thereby may help reduce symptoms of allergies and anti-inflammatory effect. Studies show that quercetin

supplementation reduces blood pressure in people who have hypertension (Mackraj et al., 258 259 2008). Kaempferol is known for it strong anti-oxidant and anti-inflammatory properties (Lau, 260 2008). It also has antibacterial, anti-cancer, antifugal, cardioproctective, hepatoprotective, hypocholesterolemic, hypoglycemic, hypotensive 261 and immunomodulatory activities (Ahmad et al., 1993; song et al. 2003, Desousa et al 2004) 262 263 This results suggest that the root extracts has both anti-inflammatory and anti-microbial 264 activities

The roots of *Manniophytum fulfum* has moderate glycoside, prominent which is digoxin. Digoxin is a structurally related and more lipophilic cardiac glycoside that largely replaced Quabain for therapy because of its superior oral bioavailability. The most common indications for digoxin are atrial fibrillation and atrial flutter with rapid ventricular response; though beta blockers and/or calcium channel blockers are a better first choice (Hallberg *et al.*, 2007). Digoxin is used widely in the treatment of various arrhythmias.

The leaves have moderate saponin, low phytosterol and benzoic acid but a very insignicant amount of allicin. Saponins are reported to have broad range of pharmcological properties (Soetan 2008). Allicin is reported to have an anti-inflammatory, antimicrobial, anti oxidation, anti-thrombotic, ` anti-ulcer, cardioprotective, hypolipidemic, hypotenisve and insecticidal properties (Elilat et al,1995; Elkayam et al 2003). Phytosterol reduce cholesterol levels by competing with cholesterol absorption in the gut of humans (Tilvis and Miethiner 1986)

Carotenoid is another bioactive compound found in manniphytum fulfum In humans, three
carotenoids (beta-carotene, alpha-carotene, and beta-cryptoxanthin) have vitamin A activity
(meaning they can be converted to retinal), and these and other carotenoids can also act as
antioxidants. In the eye, certain other carotenoids (lutein, astaxanthin and zeaxanthin)

apparently act directly to absorb damaging blue and near-ultraviolet light, in order to protect

the macula of the retina, the part of the eye with the sharpest vision (Kidd *et al.*, 2011).

283 CONCLUSION

These results suggest strong nutraceutical potential of this plant and suggest further research in it therapeutic uses in the management and prevention of disease as a result of its rich phytochemical composition.

It is a potential pharmaceutical which will help to alleviate some certain kind of diseases and infections such as cancer, cardiovascular diseases, type 2 diabetics, cough, hypertension, piles, asthma, malaria etc.

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