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SDI FINAL EVALUATION FORM 1.1

PART 1:

Journal Name:	Chemical Science International Journal
Manuscript Number:	Ms_CSIJ_41304
Title of the Manuscript:	CORROSION INHIBITION AND ADSORPTION CHARACTERISTICS OF MYRIANTHUS arboreus LEAVES EXTRACT ON COPPER IN SULPHURIC ACID SOLUTION.
Type of Article:	

PART 2:

PART 2:		
FINAL EVALUATOR'S comments on revised	Authors' response to final evaluator's comments	
paper (if any) The phrase in response 2 "because there		
alwaysinhibitor" does not make sense.		
Please use the section number from 1.	All comments have been corrected.	
I. 16. 1. Introduction		
between ls. 71 and 72. 2. *******		
I. 242. 3.5 Adsorption I. 314. 3.6 Phytochemical		
I. 335. 4. Conclusions		
Please denote the value of W _i .		
Please consider the number of significant figures		
in Tables and text.		
Solid lines in Figure 1 were depicted by the linear approximation of the data points $AW = W$	(M, M) = MM It's log(MM) assignst time (t). Therefore	
approximation of the data points. $\Delta W = W_i - W_f = at + b$. So, $W_f = -at + W_i - b$.	$(W_i-W_f) = \Delta W$. It's log(ΔW) against time (t). Thanks for that observation.	
But it contradicts the discussion of Figure 3.		
$\log(W_i - \Delta W) = \log(W_f) = \log(-at + W_i - b)$		
Not straight line.		
It is difficult to each linear regression for Figs. 4		
It is difficult to apply linear regression for Figs. 4 and 5.		
The concentration in adsorption isotherm part was		
not revised.		
Discount of comparison that is		
Please unify the format of concentration list in Tables 1-4. Table 3 seems to be better.		
I. 176 Figure 5 \rightarrow Figure 3		
I. 192. equation (7) \rightarrow equation (6)		
I. 193. equation (7) \rightarrow equation (6)		
I. 196. Figure $6 \rightarrow$ Figure 4	All commonte corrected	
I. 213. equation (8) \rightarrow equation (7) I. 214. equation (8) \rightarrow equation (7)	All comments corrected.	
1. 217. Figure 7 \rightarrow Figure 5		
I. 250. 9 \rightarrow 8, 10 \rightarrow 9, 11 \rightarrow 10		
$1.252.9 \rightarrow 8$		
$1.261.10 \rightarrow 9$		
I. 263. 11 → 10 I. 265. 8 → 6, 9 → 7		
$1.\ 200.\ 0 \rightarrow 0,\ 3 \rightarrow 1$		



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I. 268. Table 4a \rightarrow Table 5 second Fig. 5 \rightarrow Fig. 6 Fig. $6 \rightarrow$ Fig. 7 I2. 291, 295 and 296. $K_{ads} \rightarrow K_{ad}$ I. 294. 12 → 11 I. 295. 12 → 11 Is. 294 and 300. Table 5 \rightarrow Table 6 minor I. 37. [8] reported \rightarrow Hart et al. [8] reported Is. 42, 43, 48, 55 and 56. al., \rightarrow al. no comma I. 103. surface coverage (θ) and rate constant (k) \rightarrow and surface coverage (θ) I. 103. Equations, 2,3,4 and $5 \rightarrow$ Equations, 2, 3 and 4 I. 105. "100" → "× 100" I. 106. $W_B \rightarrow \Delta W_B$, $W_{inh} \rightarrow \Delta W_{inh}$ I. 136. °K \rightarrow K I. 137. M → g/l I. 150. by $[24] \rightarrow$ by Nwangbo and James [24]I. 166. $3.5 \rightarrow 3.3$ In Figures 3, 4, 5 and second 5. Log $\rightarrow \log$ Is. 173, 193, 199, 214 and 220.Log $\rightarrow \log$ All minor comments have been corrected. I. 180. Equations $5 \rightarrow$ Equation 5 I. 189. 3.6 \rightarrow 3.4