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# Impact of automobile repair activities on physicochemical and microbial properties of soils in selected automobile repair sites in Abuja, Central Nigeria

#### 4 5 ABSTRACT

6 7 This research is aimed at evaluating the impact of activities carried out in automobile repair sites on 8 quality of soils in the area. To achieve this target, soil samples were collected from five sampling points in 9 each of the three selected sites (Apo, Kugbo and Zuba) to a depth range of 0 -15 cm using a stainless 10 hand dug auger. Results of analyses of physicochemical properties pH, % porosity, electrical conductivity, particle size distribution, sulphate, chloride, nitrate and microbial contents of the soil samples indicate that 11 most of the values exceeded those of control values. Levels of heavy metals in soils were determined 12 13 using Automated Atomic Absorption Spectrophotometer (AAS). The results of the analysis revealed a 14 decreasing trend in heavy metal contents (mg/kg) in soil in the three studied automobile repair sites as follows: Apo site: Cu (7668) > Zn (5360) > Cr (1174) > Fe (467) > Pb (333) > Ni > (196) > Cd (10.6); 15 Kugbo site: Zn (1587) > Cu (1043) > Cr (783) > Ni (234) > Fe (217) > Pb (170) > Cd (9.47) and Zuba 16 site: Zn (1190) > Cr (767) > Cu (512) > Fe (279) > Pb (250) > Ni (127) > Cd (10.4). Comparative analysis 17 reveals that values of the studied heavy metals have exceeded those of control values and background 18 values of some international regulatory bodies. Pearson's correlation analysis reveals that some of the 19 20 heavy metals had very strong correlations with one another and with some of the physicochemical 21 properties of the soil. This indicates that the studied heavy metals have the same origin, mutual 22 dependence and identical behaviors. 23

Keywords: AAS; automobile repair sites; heavy metals; soil; physicochemical properties; statistical
 analysis.

# 1. INTRODUCTION

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29 Heavy metal contamination refers to the excessive deposition or discharge of toxic metal(s) in soil, 30 sludge, sediments or water as a result of anthropogenic activities [1]. Soil contamination associated with 31 heavy metal has become a major environmental problem in most developing and developed countries in 32 the world especially the potential health and ecological risk associated with such contamination [2-4]. 33 Heavy metals are one of the most serious pollutants in natural environment because of their toxicity, 34 persistence, wide spread sources, non-biodegradable, bioaccumulation properties and other negative 35 effects they have on soil quality, biota and ecosystem at large [5-7]. Heavy metals are natural 36 components of the earth crust which cannot be degraded nor destroyed completely [8-9]. Examples of 37 heavy metals include: Zinc, Manganese, Cadmium, Lead, Copper, Nickel, Antimony, Arsenic, Cobalt, Tin, 38 Vanadium, Platinum etc. Due to rapid industrialization and economic development, heavy metals have 39 been increasingly introduced in the environment through various pathways which include application of 40 pesticides, herbicides, fertilizers, untreated sludge and sewages on farm lands. Also, irrigation, river run off, atmospheric deposition and industrial activities like: metal mining, smelting of metals, combustion of 41 42 coal, leaded gasoline, spillage of petroleum products, paints, electroplating, refining refinishing of by-43 products and automobile repairs [4, 10-13].

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45 In Nigeria, "automobile repair sites" are places where various automobile repairs are carried out such as: 46 welding and fabrication, soldering, car battery recharging, scrapping, spraying and painting of vehicle 47 parts, gear box recycling, panel beating of scratched vehicles, discharge of condemned petroleum 48 products (oils, greases, hydraulics fluids) etc [14 -15]. These activities tend to release various heavy metal 49 containing wastes into the environment vis-a-viz when discharged indiscriminately in soil. Heavy metal 50 contamination in soil does not only persist in soil but also have wide range of distribution and strong 51 latency [16-17]. It has been reported that absorption and bioaccumulation of heavy metals in humans can lead to the following health issues; liver and kidney damage, neurotoxic effects in children, bone effects 52

and fractures, damages of circulatory and nerve tissues, etc [18-24]. Heavy metal contamination in and around automobile repair sites have been extensively studied [25-32].

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In this study the impact of automobile repair activities on the quality of soils in and around some selected automobile repair sites in Abuja were assessed. Physicochemical properties like pH, electrical conductivity, sulphate, chloride, nitrates and microbial properties of soil samples from these sites as well as levels of heavy metal contents were all evaluated. Pearson's correlation coefficient matrix was also conducted to determine the origin of the various heavy metals in soil. The study was conducted in November, 2015.

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## 63 2. MATERIALS AND METHOD

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# 65 2.1 Description of Study Area

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The study area Abuja is situated in the North Central part of Nigeria. The City was made the Federal Capital Territory in 1991. Geographically, Abuja lies on the coordinates of latitude 9°40'N and 9°29'E and falls within the Guinea forest – Savannah mosaic zone in the West Africa sub-region. The automobile repair sites chosen for the study were each drawn from three major districts in Abuja Municipal Area Council namely: Apo in Gudu district, Kugbo in Kugo district and Zuba in Madalla district. The geological





73 74 **Figure 1:** Geological Map of Study Site

#### 75 76 **2.1 Soil Sampling**

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Soil samples were randomly collected with a stainless hand dug auger up to a debt range of 0 -15 cm with five sampling points in each of the three automobile repair sites investigated. A controlled sample was also collected from a distance approximately 100 km where neither industrial nor commercial activities take place. The sampled soils were enclosed in separate dry new polyethylene nylon bags and taken to the laboratory for analysis.

# 84 2.2 Quality Control

All laboratory glass wares used during the analysis were of high quality and Pyrex. Also, they were 86 87 thoroughly washed and air dried prior to their various uses. The reagents used were all of analytical 88 grade. Working standard solutions for the heavy metals were prepared from their stock solutions of 100 ppm. The respective absorbencies of all the standard solutions of each investigated heavy metal were 89 90 determined using Automated Atomic Absorption Spectrophotometer (AAS) with model Unicam 969 Solar 91 according to the method described by (AOAC 1990). The standard calibration curves were obtained for 92 concentration against absorbance for each sample. Triplicate samples were also run to ensure high 93 precision of results.

## 94 **2.3 Sample Preparation and Digestion**

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Soil samples were first dried in an open air after which stones and debris present were removed through handpicking. The respective samples were further crushed in an acid pre washed mortar and pestle, sieved to an aperture size of 338 µm with a stainless laboratory sieve with make Endecott's Limited London England serial number 489494. Soil digestions were done in accordance with the methods by [25, 33].

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Parameter	Method	Reference
Heavy metal	Atomic Absorption Spectrophotometer (AAS)	[34]
рН		[35]
% Porosity		[36]
Particle size distribution	Hydrometer Method	[37]
Total coliform count		[38]
Electrical conductivity		[35]
Chloride		[39]
Sulphate	Precipitation Method	[39]
Nitrate		[39]

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#### 104 2.4 Statistical Analyses

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Statistical analysis was done using IBM SPSS statistics 16.0software. Descriptive statistics was carried to determine the mean, range and standard deviation while Karl Person correlation coefficient was used in determination of correlation between metals and with the physicochemical properties of the soil.

# 110 3. EXPERIMENTAL RESULTS AND DISCUSSION

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## **3.1 Physicochemical Contents of Studied Soil**

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114 The accumulation of certain heavy metals in sediments had been reported to be directly or indirectly controlled by redox conditions either through a change in the redox state or speciation [40]. The result of 115 the study revealed that highest pH value of 7.88 was recorded in a point in Kugbo automobile repair site 116 while the least value of 7.10 was also recorded in Zuba automobile repair site. A decreasing trend in the 117 118 mean values of pH in the investigated automobile repair site were observed to follow the sequence of Kugbo site (7.548) > Apo site (7.26) > Zuba site (7.20). This depicts that soil samples from all the sites are 119 120 slightly basic which could also be attributed to anthropogenic activities like indiscriminate discharge of used electrolytes on the soil. The results of the pH as recorded were also found to be higher than those 121 122 reported by [10, 41]. Importantly, pH plays significant role in solute concentration and in sorption and 123 desorption of contaminants in soil [42].

Results of percentage porosity of soil as shown in Table 2 reveal that all the values of percentage porosity in the investigated soils were above average with least and highest values of 59.8% and 66.4% recorded in Zuba and Kugbo automobile repair sites respectively. A decreasing order of mean values of 127 % porosity in all the sites can be written as Zuba site (61.9%) > Apo site (60.5%) > Kugbo site (59.4%). 128 High % porosity in soil could be traceable to some automobile repair activities like welding and fabrication, panel beating of automobile parts, indiscriminate discharge of metal scraps, lubricants, 129 130 hydraulics, battery electrolytes and petroleum products. Electrical conductivity recorded mean values of 131 281 µs/cm, 383 µs/cm and 384 µs/cm in Kugbo, Zuba and Apo automobile repair sites respectively. 132 These mean values also exceeded that of control site (206 µs/cm) which possibly indicates anthropogenic 133 influence on the quality of the soil. High values of electrical conductivity could be traced to deposit of 134 heavy metals which are also good electrical conductors. In addition, results of the study showed that 135 values of particle size distribution in all the sites ranges from (349 - 596) µm as shown in Table 2. Mean values of particle size distribution in the investigated sites were observed to follow a decreasing order of 136 Zuba site (576  $\mu$ m) > Apo site (563  $\mu$ m) > Kugbo site (428  $\mu$ m) respectively which also exceeded that of 137 control value and thus depicts anthropogenic influence. High particle size distribution could also be linked 138 to some automobile repair activities like scrapping and refurbishment of vehicles, spraving and painting 139 140 etc.

#### 141 **3.2 Results of Anionic Contents of Studied Soil**

Results of anionic contents in investigated soil as shown in Table 2 reveal that values of sulphate 142 fluctuated between 0.51 - 0.68 mg/g, 0.18 - 0.57 mg/g and 0.45 - 0.65 mg/g in Apo, Kugbo and Zuba 143 144 automobile repair sites respectively. High sulphate content is soil could be attributed to automobile repair sites activities like indiscriminate discharge of lubricants, electrolytes, oil sludge and used petroleum 145 146 products. Chloride contents in investigated soils recorded a decreasing mean values in the order of Apo 147 site (0.110 mg/g) > Zuba site (0.097 mg/g) > Kugbo site (0.033 mg/g). Some automobile repair activities that could have added to chloride content in soil include: Changing and repair of automobile air condition 148 149 gases, radiator coolants etc. Nitrates contents in investigated soil fluctuated between 0.09 - 0.25 mg/g in 150 Apo site, 0.09 - 0.35 mg/g in Kugbo site and 0.02 - 0.11 mg/g in Zuba site. Total coliform count unit 151 (cfu/g) as shown in Table 2 recorded some values that exceeded those of control 0.016 cfu/g and 152 standard acceptable count of 0.01 cfu/g. comparatively the values of total coliform count unit in all the 153 sites fluctuate between 0.011 - 0.023 cfu/g in Apo site, 0.009 - 0.059 cfu/g in Kugbo site and 0.09 -0.025 cfu/g in Zuba site respectively. These values also indicate various levels microbial contamination in 154 the investigated automobile repair sites. 155

157 **Table 2:** Physiochemical Properties of Soil samples from Investigated Automobile Repair Sites

Sample	pН	Percentage	Electrical	Particle	Sulphate	Chloride	Nitrate	Total
points		Porosity	Conductivity	Size	(mg/g)	(mg/g)	(mg/g)	Coliform
		(%)	(µs/cm)	Distribution				count (cfu/g)
				(µm)				
A <sub>1</sub>	7.20	59.6	388	511	0.60	0.05	0.15	0.023
A <sub>2</sub>	7.19	62.1	391	568	0.51	0.11	0.25	0.021
A <sub>3</sub>	7.22	60.1	386	561	0.63	0.13	0.15	0.014
A <sub>4</sub>	7.39	60.9	369	576	0.68	0.12	0.09	0.011
A <sub>5</sub>	7.31	59.7	388	596	0.68	0.08	0.12	0.012
$\overline{X} \pm SD$	7.26±0.09	60.5±1.04	384 <u>+</u> 8.79	562±31.7	0.51±0.26	0.10 ±0.34	0.15 <u>+</u> 0.58	0.016±0.006
Range	7.19-7.39	59.6-60.9	369-391	511-596	0.51-0.68	0.05-0.11	0.09-0.25	0.011-0.023
K <sub>1</sub>	7.49	59.8	219	459	0.57	0.06	n.d	0.03
$K_2$	7.88	63.4	230	349	0.29	n.d	n.d	0.059
K <sub>3</sub>	7.77	66.4	310	458	0.40	0.11	0.07	0.045
$K_4$	7.41	59.8	289	474	0.18	0.09	0.11	0.018
K <sub>5</sub>	7.19	60.2	355	400	0.34	n.d	n.d	0.009
$\overline{X}\pm SD$	7.55±0.28	61.9 <u>+</u> 2.92	281±56.6	428±525	0.35±0.14	0.87±0.25	0.09±0.35	0.033±0.201
Range	7.19 -7.88	59.8-66.4	219-355	349-474	0.18-0.57	0.06-0.11	0.07-0.11	0.009-0.059
Z <sub>1</sub>	7.10	59.3	389	527	0.45	0.11	0.02	0.014
<b>Z</b> <sub>2</sub>	7.19	56.8	388	556	0.58	0.15	0.09	0.025
Z <sub>3</sub>	7.19	60.8	367	587	0.65	0.13	n.d	0.008
$Z_4$	7.24	59.7	391	600	0.62	0.09	n.d	0.008
<b>Z</b> 5	7.33	60.4	380	610	0.59	0.11	0.11	0.009
$\overline{X}\pm SD$	7.20±0.09	159±1.57	383 <u>+</u> 9.87	576±34.2	0.58±0.77	0.12 ±0.23	0.07 ±0.49	0.013 <u>+</u> 0.007

Range	7.19-7.33	56.8-60.8	367-291	527-610	0.45-0.65	0.09-0.15	0.02-0.11	0.009-0.025
Control	7.29	56.6	206	366	0.16	n.d	n.d	0.016
A: Apo a	utomobile rep	<mark>air sites; K: Kเ</mark>	<mark>ugbo automob</mark>	oile repair sites;	Z: Zuba autom	nobile repair s	ites; n.d: not c	letermined

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### 3.3 Heavy Metal Contents of Studied Soil

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161 Results of heavy metal distribution in soil in the investigated automobile repair sites are shown in Table 3. From the results, copper is the most abundant heavy metal with its values ranging from (217-22000) 162 mg/kg. Copper recorded mean values of 7668 mg/kg, 1043 mg/kg and 512 mg/kg in Apo, Kugbo and 163 164 Zuba automobile repair sites respectively. These values were found to be higher than those reported by [43-46]. These values also exceeded that of control 37.3 mg/kg, background value of 36 mg/kg by (DPR, 165 166 2002) and background values of some international regulatory bodies listed in table 3. Although copper is 167 an essential mineral, high content of it could lead to serious health problem. Values of zinc were in the 168 range of (410-8421) mg/kg in all the sites. Mean values of zinc in the investigated automobile repair sites 169 decreases in the order of Apo site (5360 mg/kg) > Kugbo site (1587 mg/kg) > Zuba site (1190 mg/kg). 170 These values were observed to be very high especially when compared with values from control, DPR 171 and those of some international regulatory bodies (Table 3). Also the recorded values were also seen to 172 have exceeded those reported by [15, 47-48]. This possibly suggest anthropogenic influence which could 173 be from activities of auto mechanics like scrapping and painting of vehicles, attrition of vehicle tires, 174 indiscriminate discharge of lubricating oil containing zinc additives like zinc dithiophosphates etc.

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176 More so, nickel recorded a decreasing mean values in the order of Kugbo site (234 mg/kg) > Apo site 177 (196 mg/kg) > Zuba site (127 mg/kg). These values are higher than those from control (108 mg/kg), 178 background values of DPR (35 mg/kg), South Africa (91 mg/kg), France (50 mg/kg), China (50 mg/kg), 179 EU guidelines (75 mg/kg) and FAO/WHO guidelines (50 mg/kg). They are also higher than those reported by some researchers [15, 49-52]. Nickels entering the natural environment are mainly through human 180 181 activities like discharge of used batteries, diesel, grease, lubricating oils, tanks storing petroleum products 182 etc. High concentration of nickel in the body can displace vital elements from the enzymes in humans 183 system which could result in the breakage of metabolism route and subsequently result to heart and liver 184 disease [53] Cadmium contents in all the investigated sites were also observed to be in the range of (1.23) - 19.2) mg/kg. Cadmium also recorded mean values of 10.5 mg/kg, 10.4 mg/kg and 9.47 mg/kg in Apo, 185 Zuba and Kugbo automobile repair sites respectively. Comparatively, these values were higher than DPR 186 background values of 0.80 mg/kg, international regulatory bodies (Table 3) and those reported by [54-56]. 187 Cadmium in soil could be from condemned batteries, pigments, paints, etc. Some health problems 188 189 associated with cadmium poising include: chronic renal, anemia, cancer, lung infection, cardiovascular 190 diseases, respiratory system disorders, skin and tooth decay among others [22].

Sample points	Fe	Zn	Cu	Ni	Pb	Cr	Cd
A <sub>1</sub>	561	8200	1677	238	96.4	1117	12.5
A <sub>2</sub>	426	5288	22000	212	357	1173	11.5
A <sub>3</sub>	423	8421	12830	402	967	1916	10.6
A4	411	847	219	48.6	194	814	8.90
A <sub>5</sub>	512	4045	1616	80.5	51.7	848	8.90
⊼±SD	467 <u>+</u> 66.4	5360±3144	7668 <u>+</u> 9488	196±141	333±373	1174 <u>+</u> 444	10.5±1.59
Range	411-561	847-8421	219-22000	48.6±402	51.7-967	813- 915	8.94±12.5
K <sub>1</sub>	203	2869	3144	195	89.6	911	1.20
К2	320	719	407	370	201	288	10.2
K₃	259	2016	340	110	316	726	15.2
K <sub>4</sub>	145	1441	1017	178	15.7	915	1.50
K <sub>5</sub>	157	890	306	318	225	1074	19.2
<u>X</u> ±SD	217 <u>+</u> 73.3	1587 <u>+</u> 879	1043±1210	$234 \pm 107$	170±118	783±303	9.47±8.07
Range	145-320	719-2869	340-3144	110-370	15.7-316	288-1074	1.23-19.2
$Z_1$	302	410	686	187	199	830	10.2
Z <sub>2</sub>	331	976	351	148	58.3	764	12.5
$\overline{Z_3}$	195	1010	956	127	249	1120	9.50
Ž <sub>4</sub>	306	1710	352	126	443	630	8.80

Z <sub>5</sub>	260	1845	217	48.0	298	491	11.1
<b>⊼</b> ±SD	279 <u>+</u> 53.4	1190±589	512±303	127 <u>+</u> 50.7	250±140	767 <u>+</u> 236	10.4±1.41
Range	195-331	410-1845	217-956	48.0-187	58.3-443	491-1120	8.84-12.5
C⊤	2.45	73.4	37.3	108	102	1108	n.d
BT	5000	140	36.0	35.0	85.0	100	0.800
Iv	n.l	720	190	210	530	380	17.0

A: Apo automobile repair sites; K: Kugbo automobile repair sites; Z: Zuba automobile repair sites;  $C_T$ : control sample; n.d: not determined; n.l: no limit;  $B_T$ : background values of DPR (2002);  $I_V$ : Intervention value of DPR (2002)

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194 Furthermore, lead was observed to have mean concentrations of 333 mg/kg, 250 mg/kg and 170 mg/kg in 195 Apo, Zuba and Kugbo automobile repair sites. When compared with values established by some 196 regulatory bodies (Table 3) the recorded values were observed to higher than those reported by [57-60]. 197 Lead enters the soil through some processes like; welding and soldering, gases from vehicle exhaust, car 198 paints, dry cell batteries, leaded gasoline etc. Lead in human blood can replace calcium in the bones and is capable to create blood, bone, enzyme and nerve disorders. It can lead to general weakness, muscle 199 200 relaxation, neurotic disorders, anemia, insomnia and skin discoloration [22]. Chromium fluctuated 201 between (288 – 1174) mg/kg in all the sites. A decreasing trend in mean concentration of chromium in the 202 three automobile repair sites is seen to follow the order of Apo site (1174 mg/kg) > Kugbo site (788.6 203 mg/kg) > Zuba site (766.8 mg/kg). These values were also higher than the acceptable values of some 204 regulatory bodies as shown in Table 3 and those reported by [52, 59, 61]. Chromium can enter the soil 205 through any of the following processes: discharge of oils and greases, scrapping of vehicle parts, spraying of paints, pigments containing chromium, air conditioning coolants, brake emission, petroleum 206 products, etc. Although chromium is essential to the body, high content of it especially in form of 207 chromium (VI) is toxic to human system. Mean values of iron were seen to follow a decreasing order of 208 Apo site (467 mg/kg) > Zuba site (279 mg/kg) > Kugbo site (217 mg/kg) with a general value range of 209 (145 - 561) mg/kg. These values were observed to be lower than those reported by [33, 62-63]. 210

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_	lable 4: Background	values	of Heav	y Meta	als of so	ome Int	ernatio	onal Regulatory Bodies
_	Countries	Zn	Cu	Ni	Pb	Cr	Cd	References
	Tanzania	150	200	100	200	100	4	[C 4]

Tanzania	150	200	100	200	100	1	[64]
South Africa	240	16	91	20	6.5	7.5	[65]
France	n.a	100	50	100	n.a	2	[66]
China	250	100	50	80	200	0.5	[67]
Sweden	n.a	40	30	40	60	0.4	[15]
EU Guidelines	300	140	75	300	150	3	[68]
FAO/WHO Guidelines	300	100	50	100	100	3	[69]

<mark>n.a: not available</mark>

#### 213 **3.4 Karl Pearson's Correlation Analysis**

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Correlation analysis which is statistical tool that help to measure and analyze the degree of relationship between two of more variables. This enables us to have an idea about the degree and direction of the relationship between the variables. Correlation coefficient data is also a vital which can be used to deduce the possible source(s) of heavy metals in soil. Mathematically, Karl Pearson's correlation coefficient can be stated as:

$$r = \frac{N\sum XY - \sum X\sum Y}{\sqrt{N\sum X^2 - \left(\sum X\right)^2} \sqrt{N\sum Y^2 - \left(\sum Y\right)^2}}$$
(1)

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where N = number of samples; X,Y are the single samples indexed;

The correlation coefficient matrix for heavy metals present in soil samples from the various automobile repair sites investigated are shown in Tables 5, 6 and 7 below. Pearson correlation coefficients were implored for all the sites. The results shown in Table 5 indicate that strong positive correlation exist between the following metals like Pb/Cr (r = 0.94) evidencing that in 94 % of cases, the correlation of both 225 heavy metals increases simultaneously. Other strong positive correlation were seen among Zn/Ni (r = 226 0.88), Cr/Ni (r = 0.96), Cr/Zn (r = 0.75), Cd/Zn (r = 0.76), Ni/Cd (r =0.60) and Cu/Pb (r = 0.55) 227 respectively. This indicates that the studied heavy metals have identical behavior, are mutually 228 dependence and are also from the same source(s). Strong negative correlations also exist between some 229 physicochemical properties of the soil samples and some heavy metals as follows: Cd/pH (r = -0.88), 230  $SO_4^{2-}/NO_3^{-1}$  (r = -0.96), Zn/pH (r = -0.87), Cd/PSD (r = -0.86), Cu/SO $_4^{2-}$  (r = -0.82), Cd/SO $_4^{2-}$  (r = -0.77), Ni/pH (r = -0.76), Fe/%P (r = -0.69), Cu/pH (r = -0.65) and Cr/pH (r = -0.59). This strong negative 231 correlation indicates that the sources of the metal were from different origin. 232

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234 Table 5: Pearson's Correlation Coefficient Matrix of Heavy Metals in Apo Automobile Repair Sites (n = 5). Fe Zn Cu Ni Pb Cr Cd EC %P PSD SO42 Cľ NO<sub>3</sub> pН Fe 1.00 0.40 1.00 Zn Cu -0.49 0.33 1.00 -0.07 0.88\* Ni 0.53 1.00 Pb 0.48 0.83 -0.57 0.55 1.00 0.96\*\* 0.94\* Cr -0.28 0.75 0.52 1.00 0.38 0.76 Cd 0.41 0.60 0.12 0.37 1.00 pН -0.27 -0.87 -0.65 -0.76 -0.35 -0.59 -0.88\* 1.00 EC 0.44 0.72 0.52 0.49 0.11 0.34 0.59 -0.85 1.00 %P -0.69 -0.36 0.71 -0.10 0.11 -0.07 -0.08 -0.07 0.06 1.00 PSD -0.51 -0.64 0.07 -0.46 0.01 -0.24 -0.86 0.56 -0.21 0.26 1.00 SO42-0.08 -0.41 -0.82 -0.41 -0.16 -0.26 -0.77 0.80 -0.59 -0.64 0.38 1.00 -0.61 0.30 0.62 0.68 0.63 0.48 Cľ 0.69 -0.44 -0.03 0.59 -0.33 -0.62 1.00 NO<sub>3</sub><sup>-</sup> -0.14 0.41 0.91\* 0.43 0.24 0.31 0.64 -0.80 0.70 -0.16 -0.96\*\* 0.53 0.64 1.00

EC: Electrical Conductivity; %P: Percentage Porosity; PSD: Particle Size Distribution; Significant /r/\*(p < 0.05); \*\* (p < 0.01)

235 Results of correlation coefficient matrix of Kugbo site shown in Table 6 reveals that strong positive 236 correlations exist between heavy metals in the sampled soils as Pb/Cd (r = 0.87), Zn/Cu (r = 0.81) and 237 Pb/Zn (r = 0.53). Also among metals and physiochemical properties like Fe/pH (r = 0.93), Pb/%P (r = 0.80), Fe/%P (r = 0.73), Zn/SO<sub>4</sub><sup>2-</sup> (r = 0.79), PSD/Zn (r = 0.74), Cu/SO<sub>4</sub><sup>2-</sup> (r = 0.70) and Zn/Cl<sup>-</sup> (r = 0.61). 238 239 Strong negative correlation also occurred between heavy metals like Fe/Cr (r = -0.93), Ni/Zn (r = -0.72), 240 Cd/Cu (r = -0.74) and Pb/Cu (r = -0.58) and with physiochemical properties of soil like Cr/pH (r = -0.91), Ni/PSD (r = -0.87), PSD/Cd (r = -0.69), Ni/NO<sub>3</sub> (r = -0.65) and Cu/EC (r = -0.63). Strong positive 241 correlations were also seen among some heavy metals in Zuba site as follows: Cu/Cr (r = 0.95), Pb/Zn (r 242 = 0.65), Cr/Ni (r = 0.53) and with some physiochemical properties like Fe/EC (r = 0.96), Zn/pH (r = 0.94), 243 Zn/PSD (r = 0.94), Pb/%P (r = 0.69), Pb/PSD (0.65), Zn/ SO<sub>4</sub><sup>2-</sup> (r = 0.66). Major strong negative 244 245 correlation among heavy metals like Ni/Zn (r = -0.87), Cd/Pb (r = -0.84), Cr/Zn (r = -0.65), Cu/Zn (r = -246 0.64) and Cu/Fe (r = -0.63). Also between heavy metals and some physiochemical properties like Ni/pH (r 247 = -0.94), Pb/Cl<sup>-</sup> (r = -0.89), Ni/PSD (r = -0.88), Fe/%P (r = -0.80), Cr/pH (r = -0.75), Cu/pH (r = -0.75), Cd/%P (r = -0.74), Cu/NO<sub>3</sub> (r = -0.65) and Cu/EC (r = -0.61) respectively. The correlation coefficients 248 between concentrations of various heavy metals and those of physiochemical properties of the soil 249 250 samples shows strong linear relationship between the variables, which probably indicate their common 251 origin or their common sink in the soils. Presence of heavy metals in these soils could also be attributed 252 to indiscriminate discharge of heavy metal containing wastes generated from various automobile activities 253 in soils in and around the investigated automobile repair sites.

255	Table (	6: Pearson's	s Correlatior	Coefficient	Matrix	of Heavy	y Metal	s in	Kugbo	Automobi	le Repair	Sites	
256	<mark>(n=5).</mark>												
	Fe	Zn	Cu Ni	Pb	Cr	Cd	рН	EC	%P	PSD	SO42-	Cl	NO <sub>3</sub> <sup>-</sup>
Fe	1.00												
Zn	-0.14	1.00											

Cu	-0.22	0.81	1.00											
Ni	-0.27	-0.72	-0.28	1.00										
Pb	-0.53	-0.20	-0.58	0.05	1.00									
Cr	-0.93*	0.33	0.28	-0.37	-0.27	1.00								
Cd	0.18	-0.51	-0.74	0.28	0.87	0.00	1.00							
pН	0.93	0.00	-0.17	-0.02	-0.38	-0.91*	-0.04	1.00						
EC	-0.50	-0.38	-0.63	-0.10	0.39	0.56	0.71	-0.55	1.00					
%P	0.73	-0.06	-0.51	- 0.24	0.80	-0.58	0.47	0.77	0.06	1.00				
PSD	-0.13	0.74	0.49	-0.87	-0.40	-0.11	-0.69	0.21	-0.35	0.08	1.00			
SO4 <sup>2-</sup>	0.08	0.79	0.70	-0.27	0.17	0.23	-0.09	-0.02	-0.31	-0.01	0.18	1.00		
Cl	-0.20	0.61	0.19	-0.97**	-0.14	0.23	-0.38	0.13	0.03	0.27	0.92*	0.06	1.00	
NO <sub>3</sub> <sup>-</sup>	-0.36	0.06	-0.19*	-0.65	-0.33	0.18	-0.32	-0.02	0.24	0.08	0.66	-0.54**	0.79	1.00

EC: Electrical Conductivity; %P: Percentage Porosity; PSD: Particle Size Distribution; Significant /r/\*(p < 0.05);\*\* (p < 0.01)

257	Table 7:	Pearson	's Correl	ation Coe	efficient M	latrix of	Heavy N	letals ir	n Zuba <i>I</i>	<mark>Automo</mark>	bile Re	oair Site	<mark>s (n=5</mark> )	) <mark>.</mark>
	Fe	Zn	Cu	Ni	Pb	Cr	Cd	рН	EC	%P	PSD	SO4 <sup>2-</sup>	Cl	NO <sub>3</sub> <sup>-</sup>
Fe	1.00													
Zn	-010	1.00												
Cu	-0.63	-0.64	1.00											
Ni	0.37	-0.87	0.59	1.00										
Pb	-0.22	0.65	-0.14	-0.39	1.00									
Cr	-0.53	-0.65	0.95*	0.53	-0.32	1.00								
Cd	0.41	-0.18	-0.40	-0.05	-0.84	-0.21	1.00							
pН	-0.04	0.94*	-0.75	-0.94*	0.43	-0.75	0.10	1.00						
EC	0.96*	-0.03	-0.61	0.35	0.03	-0.60	0.16	-0.02	1.00					
%P	-0.80	0.34	0.36	-0.44	0.69	0.12	-0.73	0.22	-0.59	1.00				
PSD	-0.42	0.94*	-0.35	-0.88*	0.65	-0.37	-0.30	0.85	-0.36	0.53	1.00			
SO4 <sup>2-</sup>	-0.48	0.66	-0.03	-0.58	0.34	0.09	-0.20	0.52	-0.53	0.66	0.82	1.00		
Cl⁻	-0.02	-0.45	0.24	0.24	-0.89*	0.49	0.72	-0.32	-0.29	-0.59	-0.36	0.11	1.00	
NO <sub>3</sub> <sup>-</sup>	0.26	0.32	-0.69	-0.55	-0.43	-0.59	0.83	0.60	0.10	-0.38	0.17	0.00	0.33	1.00

EC: Electrical Conductivity; %P: Percentage Porosity; PSD: Particle Size Distribution; Significant /r/\*(p < 0.05);\*\* (p < 0.01)

258 259

#### 3.5 Variation in Level of Heavy Metal in the Study Area

260

261 In order to have a comparative knowledge about the level of heavy metal contamination in soil in and around the studied mechanic villages, data obtained from these sites were compared with background 262 values established by DPR 2002 and other standard regulatory bodies as shown in Table 4. The 263 264 background value of an element is the maximum level of the element in an environment beyond which the 265 environment is said to be polluted by the element [70]. All the investigated heavy metals but iron had 266 values greater than the maximum acceptable limit of these bodies. This implies that the auto mechanic 267 sites had various degrees of contamination which could be traceable to anthropogenic activities. A trend 268 of variation of heavy metal contents in soils in three automobile repair sites can be summarized as: Apo 269 site: Cu > Zn > Cr > Fe > Pb > Ni > Cd; Kugbo site: Zn > Cu > Cr > Ni > Fe > Pb > Cd; Zuba: Zn > Cr > Cu > Fe > Pb > Ni > Cd. The result of the study also reveals that Cu, Zn and Cr had very high variation 270 and standard deviation. Pb, Ni and Fe showed moderate variation while Cd showed the least variation. 271 Large variations imply great heterogeneity of metals in soil while low variations show more or less 272 273 homogeneous distribution of heavy metals in soil. This could be traced to different levels of contamination 274 caused by varying degrees of automobile wastes discharge in soils [71].

### 276 4. CONCLUSION

277 The results obtained from the study supply valuable information on various levels of heavy metal contents 278 in soils in and around the three major automobile repair sites in Abuja, Nigeria. The results also showed 279 the distribution pattern of the studied heavy metals whose values in all the sites with the exception of iron 280 were found to have exceeded the background or pre-industrial reference value(s) provided by some world 281 regulatory bodies. The high values recorded could be attributed to anthropogenic activities like indiscriminate discharge of heavy metal containing wastes generated from various auto-mechanic 282 283 practices. A trend of variation of heavy metal contents in soils in three automobile repair sites can be summarized as: Apo site: Cu > Zn > Cr > Fe > Pb > Ni > Cd; Kugbo site: Zn > Cu > Cr > Ni > Fe > Pb > 284 Cd and Zuba: Zn > Cr > Cu > Fe > Pb > Ni > Cd. Statistical analysis conducted using Pearson's 285 286 correlation coefficient on the variables revealed that these heavy metals had strong correlation with each other and with some of the physicochemical properties of the soil. They also showed a high 287 approximation to perfect correlation indicating a strong linear relationship between the measured 288 289 variables.

290

#### 291 **RECOMMENDATION**

292 Based on the findings of this research work, it is therefore suggested that systematic investigation should 293 be conducted in order to check the rate of heavy metal loading and change in the quality of soil in and 294 around these automobile repair sites. Indiscriminate discharge of heavy metal containing wastes 295 generated from auto mechanic repairs on soil in particular and environment at large should be totally 296 stopped. Better still, these wastes should be collected, recycled and properly disposed in order to save 297 our environment from harmful pollutants. Also, adequate sensitization on the damages of indiscriminate 298 discharge of waste in the environment should be made by relevant authorities and a more environment 299 friendly automobile mechanic village concept and proper waste management encouraged.

#### 300 **COMPETING INTEREST**

301 Authors have strongly declared that no competing interest exists.

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