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Comparative Phytochemical Screening and Antioxidant Activity of the Fruit and Stem bark of *Tetrapleura tetraptera*

3 ABSTRACT

4 Tetrapleura tetraptera belongs to the family Mimosaceae. The present investigation was carried out to evaluate the antioxidant and phytochemical activities of ethanol extracts of 5 *T.tetraptera* stem bark and fruit. DPPH radical scavenging assay and Fe^{+3} reducing assay 6 7 were carried out to determine the antioxidant activities of the extracts. The extracts exhibited 8 marked antioxidant activity by scavenging DPPH free radical in a concentration dependent manner. In Fe⁺³ reducing assay, increase in the absorbance revealed the reducing power of 9 10 the extracts. For the stem bark, the value ranged from 0.393 to 1.641 mg Ascorbic acid 11 equivalent/ml of extract and fruit (0.342 to 1.325 mg Ascorbic acid equivalent/ml of extract. 12 The DPPH for the stem bark ranged from 28.74% to 85.26% while that of fruit ranged from -13 10.56% to 66.01%. The Preliminary phytochemical analysis of ethanol extract of the stem 14 bark showed the presence of tannins, saponins, flavonoids, glycosides, and anthraquinones 15 while for the fruit extract glycosides and anthraquinones were absent.

Key words: *Tetrapleura tetraptera*, % DPPH radical scavenging assay, % Fe⁺³reducing assay
and phytochemical

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19 **1.0 INTRODUCTION**

Obtaining adequate nutrients from various foods plays a vital role in maintaining normal function of human body. With recent advances in medical and nutrition sciences, natural products and health-promoting foods have received extensive attention from both health professionals and the common population. New concepts have appeared with this trend, such as nutraceuticals, nutritional therapy, phytonutrients and phytotherapy ^[1,2, 3]. These functional or medicinal foods are maintaining well-being, enhancing health and modulating immune

function to prevent specific diseases. They also hold great promise in clinical therapy due to 26 their potential radiotherapy and significant advantages in reducing the health care cost^[4]. The 27 28 history of plants being used for medicinal purpose is probably as old as the history of 29 mankind. Extraction and characterization of several active phyto-compounds from these green factories have given birth to some high activity profile drugs. The potential natural 30 anticancer drugs like vincristine, vinblastine and taxol can be the best examples^[5]. Free 31 32 radicals are found to be a product of normal phytonutrients or phytomedicines play positive 33 roles in metabolism. Although oxygen is essential for aerobic forms of life, oxygen 34 metabolites are highly toxic. As a consequence, reactive oxygen species (ROS) are known to 35 be implicated in many cell disorders and in the development of many diseases including cardiovascular diseases, atherosclerosis, chronic inflammation etc^[6;7]. Although organisms 36 have endogenous antioxidant defences produced during normal cell aerobic respiration 37 against ROS, other antioxidants are taken both from natural and synthetic origin^[8]. 38 39 Antioxidants that can inhibit or delay the oxidation of an oxidizable substrate in a chain reaction, therefore, appear to be very important ^[9]. Synthetic antioxidants are widely used but 40 41 their use is being restricted nowadays because of their toxic and carcinogenic effects. Thus, 42 interest in finding natural antioxidants, without any undesirable effect, has increased greatly^[8]. The objective of the present investigation was to evaluate the phytochemical and 43 44 antioxidant potential of different concentrations of the stem bark and fruit of *T.tetraptera*.

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46 **2.0 MATERIALS AND METHODS**

2.1 Collection of Plant Material: Fresh stem bark and fruit of *Tetrapleura tetraptera* were
bought in a local market in Ondo, Nigeria. The plant parts were washed thoroughly 2-3 times
with running water and once with distilled water and then air-dried on sterile blotter under
shade.

51 2.2 Solvent Extraction: Thoroughly washed plant parts were dried in shade for five days and 52 then powdered with the help of blender. The powdered plant parts were extracted 53 successively with ethanol solvent in Soxhlet extractor for 48h. The solvent extracts were 54 concentrated under reduced pressure and preserved at 5°C in airtight bottle until further use.

2.3 Phytochemical Analysis: The ethanol extracts were subjected to preliminary
 phytochemical screening to screen the presence of various secondary metabolites^[10;11].

57 Reducing property: The reducing property was determined by assessing the ability of extracts to reduce FeCl₃ solution as described by Pulido, et al ^[12]. Briefly, extracts (0-250 µL of stock) 58 59 were mixed with 250 µL 200mM Sodium phosphate buffer (pH 6.6) and 250 µL of 1% 60 potassium ferrocyanide, the mixture was incubated at 50°C for 20mins, thereafter 250 μ L of 61 10% trichloroacetic acid was added, and subsequently centrifuged at 650rpm for 10mins, 62 1000 μ L of the supernatant was mixed with equal volume of water and 100 μ L of 63 0.1g/100mL ferric chloride, the absorbance was later measured at 700nm. A higher 64 absorbance indicates a higher reducing power.1,1-diphenyl-2 picrylhydrazyl free radical 65 scavenging ability: The free radical scavenging ability of the extracts against DPPH (1,1diphenyl-2- picrylhydrazyl) free radical was evaluated as described by Halliwell, et al.^[9]. 66 67 Briefly, appropriate dilution of the extracts (1mL) was mixed with 1mL of 0.4mM methanol 68 solution containing DPPH free radicals, the mixture was left in the dark for 30mins and the 69 absorbance was measured at 516nm. The DPPH free radical scavenging ability was 70 subsequently calculated.

71 Scavenging activity (%) = $A - B / A \ge 100$

72 Where A is absorbance of DPPH and B is absorbance of DPPH and extract combination.

73 **3.0 RESULTS**

74 Table 1: Phytochemical analysis

Sample	Alkaloids	Flavonoids	Saponins	Tannins	Glycosides	Anthraquinones
<i>Stem</i> bark						
	-	+	+	+	+	+
Fruit						
	+	+	+	+	-	-

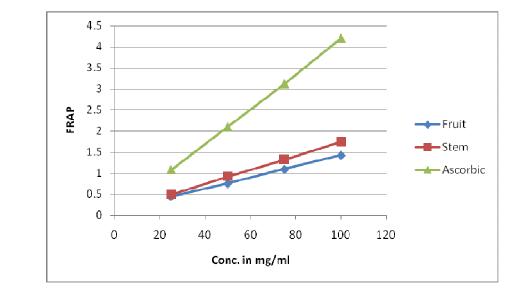
75 Key: +ve = present; -ve = absent.

76 Preliminary phytochemical analysis of ethanol extract revealed the presence of tannins,

saponins, glycosides, flavonoids and anthraquinones in the stem bark while only alkaloids

was absent. The fruit extract contains all except glycosides and anthraquinones. (Table 1)

79 Antioxidant activities



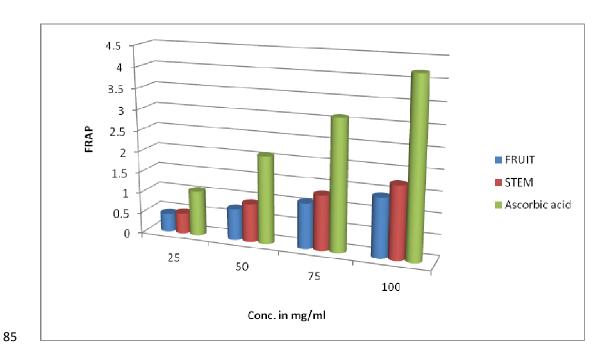
80 Figure 1: Ferric Reducing Antioxidant Property

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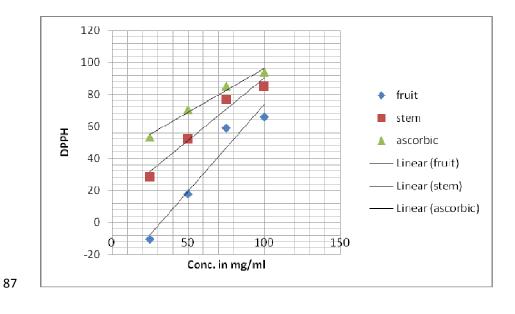
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84 Figure 2: Ferric reducing antioxidant property



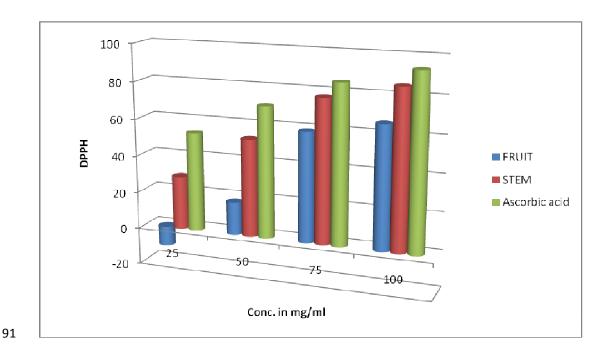
86 Figure 3: DPPH free radical scavenging ability (%)



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90 Figure 4: DPPH free radical scavenging ability (%)



92 4.0 DISCUSSION

93 Plants produce a diverse range of bioactive molecules, making them rich source of different 94 types of medicines. Higher plants, as sources of medicinal compounds, have continued to 95 play a dominant role in the maintenance of human health since ancient times. Over 50% of all 96 modern clinical drugs are of natural plant origin and natural products play an important role in drug development programs in the pharmaceutical industry ^[13]. The medicinal value of 97 plants lies in some chemical substances that produce a definite physiological action on the 98 99 human body. The most important of these bioactive constituents of plants are alkaloids, tannins, flavonoids and phenolic compounds.^[14]. Phytomedicines derived from plants have 100 101 shown great promise in the treatment of various diseases including viral infections. Single 102 and poly herbal preparations have been used throughout history for the treatment of various types of illness^[15]. Plant derived natural products have received considerable attention in 103 recent years due to their diverse pharmacological activities.^[16] 104

Free radicals are chemical species containing one or more unpaired electrons that makes them
highly unstable and cause damage to other molecules by extracting electrons from them in

107 order to attain stability. In recent years much attention has been devoted to natural antioxidant and their association with health benefits ^[17]. Free radicals contribute to more 108 109 than one hundred disorders in humans including atherosclerosis, arthritis, ischemia and reperfusion injury of many tissues, central nervous system injury, gastritis, cancer and AIDS 110 ^[18;19]. There are several methods available to assess antioxidant activity of compounds. DPPH 111 112 free radical scavenging assay is an easy, rapid and sensitive method for the antioxidant 113 screening of plant extracts. In presence of an antioxidant, DPPH radical obtains one more electron and the absorbance decreases.^[20] The antioxidant activities have been reported to be 114 the concomitant development of reducing power.^[21]. 115

116 The result of antioxidant activity of different concentrations of ethanol extracts and standard 117 (Ascorbic acid) is shown in Figures 3&4. The extracts exhibited marked antioxidant activity 118 by scavenging DPPH* (free radical) and converting into DPPHH. The extracts have shown 119 concentration dependent radical scavenging activity, conversely, at 25 mg/ml, the extract of 120 the fruit of *Tetrapleura tetraptera* is pro-oxidant. The scavenging activity of the standard was 121 higher than those of ethanol extracts. Though the DPPH radical scavenging abilities of the 122 extract were less than those of Ascorbic acid, the study showed that the extracts have the 123 proton-donating ability and could serve as free radical inhibitors or scavenger, acting possibly as primary antioxidant. Fe⁺³ reducing power assay was carried out for the measurements of 124 125 reductive ability of different concentrations of ethanol extracts and standard, ascorbic acid. 126 An increase in the absorbance revealed the reducing power of extracts. In this study, the 127 reducing power of ethanol extracts was found to increase with the dose (Figures 1&2). The 128 reducing capacity of a compound may serve as significant indicator of its potential antioxidant activity.^[22]. 129

130 CONCLUSION

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- 131 The efficacy of the fruit and stem bark extracts of *T.tetraptera*, may be attributed to the 132 phytochemicals present in the solvent extract. Further studies on isolation of active
- 133 constituents and their biological activities are to be carried out.

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