# Composition and Frequency dependent Dielectric properties of Cr-Co Nano ferrites

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#### ABSTRACT

6 Chromium substituted Co-ferrite nanoparticles with general formula  $Cr_xCoFe_{2-x}O_4$ 7  $(0 \le x \le 0.5)$  have been synthesized by using sol-gel technique. The samples were sintered at 8 900°C for 3hours. The effect of chromium substitution on dielectric properties of Co-ferrites is reported in this paper. The X-ray diffraction patterns indicate the formation of single phase 9 spinel structure for all the samples. The particle size of all samples was calculated using Debye-10 Scherer's formula and it is in the range of 56-75 nm. The investigation on dielectric constant ( $\epsilon$ '), 11 12 dissipation factor(D) and AC conductivity( $\sigma_{ac}$ ) was carried out at a fixed frequency 1kHz and in the frequency range of 100Hz to 1MHz at room temperature using LCR meter. The frequency 13 dependence of the dielectric constant ( $\epsilon$ ) and AC conductivity ( $\sigma_{ac}$ ) show a normal dielectric 14 behaviour. The frequency dependence of dissipation factor (D) of the chromium substituted 15 16 cobalt ferrite samples displays a relaxation peak at certain frequency. This peak suggests the presence of relaxing dipoles in the Cr-Co nano ferrite samples. It is also observed that the 17 broadening of the relaxation peak with chromium-content(x) in dissipation factor versus 18 frequency curve is due to the strengthening of dipole-dipole interactions. The dielectric 19 behaviour of the Cr-doped cobalt nano ferrites is explained based on Koop's two layer model 20 and Maxwell-Wagner interfacial polarization theory and Debye-relaxation theory. 21

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Keywords: Co-Cr ferrites, nanoparticles, dielectric properties, dipole - dipole interaction.

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## 25 **1. INTRODUCTION**

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Ferrites open a new zone in the field of material science, because of their scientific aspect andversatile applications. The spinel ferrite nano particles are most interesting magnetic materials in

the field of science and technology due to their twin property as a magnetic conductor and an 29 electrical insulator. In general ferrites are used in high frequency circuits, high speed digital 30 31 tapes, transformer cores and antennas due to their high dielectric constant, high resistivity, high permeability, low dielectric loss, mechanical strength and chemical stability. The importance of 32 these materials can be explained by their low price and their reasonable magnetic performances, 33 so the range of applications is extremely wide [1, 2]. Besides the magnetic properties, 34 polycrystalline ferrites are very good dielectric materials and have many applications ranging 35 from microwave to radio frequencies. Hence, it is important to study their dielectric properties at 36 different frequencies. Detailed knowledge of the effect of dopant on the dielectric properties of 37 ferrites may be useful for synthesis of materials with tunable dielectric properties, which is very 38 helpful in practical applications. 39

 $CoFe_2O_4$  possess an inverse spinel stricture in which  $Co^{2+}$  ions present in the octahedral (B) sites 40 and Fe<sup>3+</sup> ions distributed almost equally between the tetrahedral (A) and octahedral (B) sites [3]. 41 Cobalt ferrite is a familiar hard magnetic material with great physical and chemical stabilities, 42 high coercivity, moderate magnetization and good magnetostrictive properties among all the 43 44 ferrite family. Due to these properties Co-ferrites have a great importance from the application point of view, where these materials are widely used in many ferrite devices and production of 45 magnetic and electronic components, high frequency devices, memory cores, high density 46 information storage devices and also in biomedical field [4-5]. 47

48 It is a well-known fact that the intrinsic properties of ferrite materials are intensively influenced by the composition and the microstructure of ferrite particles, which are highly sensitive to the 49 50 experimental conditions and preparation technique which is used for their synthesis. The wet chemical methods appear to offer an alternative way to control the composition and 51 52 microstructures of the materials. There are generally various wet chemical methods for fabrication of ferrite nanoparticles such as co-precipitation method [6, 7], hydrothermal method 53 [8,9], sol-gel [10-12], spray-spin-heating-coating method [13] and microwave sintering method 54 [14]. Out of all these, sol-gel technique is promising method for synthesis of nano ferrites in 55 bulk scale due to the production of homogeneous particles. sol-gel method is the most 56 convenient to synthesize nano particles because of its simplicity, inexpensive precursors, low 57 sintering temperature, short preparation time, better control over crystallite size and other 58 properties of the materials [15]. 59

Previously, we have studied the effect of  $Cr^{3+}$  ion on the structural and magnetic properties of the 60 Co-ferrite nanoparticles [16]. The structural and magnetic properties of the Cr substituted Co-61 62 ferrite samples have been thoroughly studied by using XRD,SEM,TEM-EDAX,FTIR spectroscopy and Vibrating Sample Magnetometer (VSM) measurements. In this study, we have 63 observed that the chromium ion greatly affected the intrinsic properties of the final product. 64 Therefore, as an extension of our previous research work, the aim of the present work is to 65 investigate the influence of composition and frequency on dielectric behaviour of Cr-substituted 66 Co-ferrite nanoparticles synthesized by sol-gel technique. The variations of dielectric constant, 67 dissipation factor, ac conductivity, have been analyzed as a function of frequency at room 68 69 temperature.

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## 72 **2. EXPERIMENTAL**

## 74 2.1 Synthesis Technique

76 Nanocrystalline  $Cr_x CoFe_{2-x}O_4$  (0.0  $\leq x \leq 0.5$ ) samples were prepared by sol-gel route using high  $Co(NO_3)_2$  $.6H_2O$ ,  $Fe(NO_3)_3.9H_2O$ , 77 purity (Analytical grade)  $Cr(NO_3)_3.9H_2O$ and 78 C<sub>6</sub>H<sub>8</sub>O<sub>7</sub>.H<sub>2</sub>O(Hydrated citric acid) respectively in stoichiometric proportions. The sol was prepared by taking 1:3 ratio of citric acid and metal nitrates and the pH was adjusted to 7 by 79 80 adding small quantity of ammonia to the sol. Then the sol was heated on a hot plate at 100°C under constant and intensive stirring to transform it into a gel. The gel was then heated at 150°C 81 82 for auto-combustion to take place. Then the dried gel burnt in a self-propagating combustion lead 83 to form loose powder .The powder is crushed in an agate mortar. After that the powder was presintered at 900°C for 3 hrs and cooled slowly to room temperature to obtain the nano ferrite 84 particles. The step by step procedure of synthesis of Cr-Co ferrite nanoparticles is shown in the 85 Fig. 1. 86

The pre-sintered ferrite samples were mixed with an organic binder (small quantity of polyvinyl alcohol (PVA)) for granulation. Then the granulated powder was dried and compressed into disc shaped pellet of diameter 10 mm and thickness about 2 mm by applying the pressure of 3 tons. The samples were finally sintered at 950°C for 5hrs and slowly allowed to cool to room temperature. For dielectric measurements both sides of the pellets were coated with a thin layer of silver paste and air dried for making a good electrical contact.

**2.2 Dielectric measurements** 93 94 95 The dielectric parameters of the Co-Cr ferrite samples were measured at room temperature in the frequency range of 100Hz to 1MHz by using LCR Meter (Waynekerr Model: 43100). The 96 values of capacitance (C) and dissipation factor (D) were noted directly by this LCR meter at 97 different frequencies. 98 99 The dielectric constant ( $\epsilon$ ) was calculated by using the following formula [17] 100  $\varepsilon = C t / \varepsilon_0 A$ .....(1) 101 where C is the capacitance of the sample, A is the surface area, t is the thickness of the pellet, 102 and  $\varepsilon_0$  is the permittivity of free space. 103 104 From dielectric constant and dissipation factor (D), the ac conductivity ( $\sigma_{ac}$ ) of the ferrite 105 samples can be calculated by using the following equation [18] 106 107  $\sigma_{ac} = 2\Pi f \epsilon' \epsilon_0 D$ .....(2) Where  $\omega = 2\Pi f$  is the angular frequency. 108 **3. RESULTS AND DISCUSSION** 109 **3.1. Dielectric Properties** 110

111 The dielectric properties of Cr-Co ferrites have been investigated as a function of composition 112 and frequency. Dielectric studies of these samples may be useful for widening its range of 113 applications.

#### 114 **3.1.1. Dielectric constant** (ε')

The polarization phenomenon in ferrites is quite consistent with the conduction process. The transferring of electrons between the Fe<sup>2+</sup> and Fe<sup>3+</sup>ions in octahedral sites results in the local displacement of electrons in the direction of the external applied electric field and these electrons induce the polarization in the ferrite materials. The frequency dependence of the dielectric constant ( $\epsilon'$ ) for Cr<sub>x</sub>CoFe<sub>2-x</sub>O<sub>4</sub> (0.0  $\leq x \leq$  0.5) spinel ferrite system was studied at room temperature in the frequency range of 100Hz to 1MHz. Fig.2 shows the variation of the dielectric

121 constant of Cr-Co nano ferrite samples as a function of frequency at room temperature. It can be observed that all the compositions exhibit dielectric dispersion where the dielectric constant 122 123 decreases rapidly with increasing frequency in low frequency region and the rate of decrease is slow in the high frequency region, the dielectric behaviour approaches almost frequency 124 independent in high frequency region. The dielectric dispersion with increasing frequency is a 125 normal behaviour observed in most of the ferrimagnetic materials [19, 20]. The dielectric 126 127 dispersion in Cr-Co ferrites is due to the Maxwell-Wagner type interfacial polarization which is in agreement with Koop's theory [21, 22]. According to this theory, a ferrite material has been 128 assumed to be composed of fairly well conducting grains which are separated by highly resistive 129 grain boundaries. It has been noticed that in ferrite materials the relative permittivity or dielectric 130 constant is directly proportional to the square root of electrical conductivity. At lower 131 frequencies the grain boundaries are more predominant in electrical conduction hence they have 132 high values of relative permittivity, while the grains are highly effective at high frequency region 133 thus they have smaller values of relative permittivity. However, in case of nano ferrite material, 134 both grains and grain boundaries are large in number as compared with the bulk ferrite material 135 case, which makes the dielectric polarization mechanism more complex. In case of nano ferrite 136 materials, there is a chance of having a high relative permittivity because of large surface 137 polarization due to the high surface area of individual grains. In the low frequency region, 138 surface polarization plays a predominant role than ionic or electronic polarization in determining 139 140 the dielectric behaviour of nano-ferrite materials [22]. The higher value of dielectric constant ( $\epsilon$ ') at lower frequency is due to the inhomogeneous dielectric structure in ferrite material. The 141 inhomogeneities present in the material can be oxygen vacancies, grain boundary defects, 142 interfacial dislocation pile ups, high concentration of  $Fe^{2+}$  ions and porosity, etc., while the 143 144 decrease in dielectric constant value with increase in frequency is due the fact that the lagging of the species which are contributing to the polarization behind the applied ac electric field at very 145 high frequencies [23, 24]. The independent behaviour of the dielectric constant ( $\epsilon$ ) on frequency 146 in the high frequency region (5.1KHz-1MHz) can be understood on the basis of Debye relaxation 147 theory [25]. This independent behaviour of the dielectric constant ( $\varepsilon$ ) is due to the impact of a 148 large relaxation time; since the space charge carriers in a dielectric material require a finite time 149 to align their axes in the direction of an applied ac field. If the frequency of the field reversal 150 increases, at some stage, the space charge carriers cannot line up with the applied field and 151

unable to follow the field variation, as a result dielectric constant of the material decreases andbecomes frequency independent.

154 To investigate the influence of composition on the dielectric properties of Cr-Co ferrite system, the dielectric parameters of Cr-Co ferrite system were measured at a fixed frequency of 1 kHz. 155 156 The values are recorded in Table 1. In the present investigation the dielectric behaviour of 157 CoFe<sub>2</sub>O<sub>4</sub> was drastically affected by Cr doping. From Table 1 one can conclude that, the 158 dielectric constant of  $CoFe_2O_4$  linearly increases up to x=0.3, suddenly decreases at x=0.4 and again it increases for x=0.5. The variation of dielectric constant as a function composition is 159 160 shown in Fig.5. The increase in dielectric constant with increasing Cr-content may be attributed to the formation of  $Fe^{2+}$  ions in octahedral site. The formation of  $Fe^{2+}$  ions in octahedral site 161 increases the electron exchange between  $Fe^{2+}$  and  $Fe^{3+}$  ions which in turn enhances the 162 polarization. However, the decrement in dielectric constant value for  $Cr_0 {}_4CoFe_1 {}_6O_4$  sample may 163 be due to the migration of Fe<sup>3+</sup> ions from octahedral site to tetrahedral site which decreases the 164 hopping and dielectric polarization. 165

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#### 167 **3.1.2. Dissipation factor (D)**

The dielectric loss in ferrites mainly originates due to the exchange of electrons between ferrous 168 ions (Fe<sup>2+</sup>) and ferric ions (Fe<sup>3+</sup>) on the octahedral site and defect dipoles. The variation of 169 dissipation factor (D) against frequency at room temperature for  $Cr_xCoFe_{2-x}O_4$  (where x = 0.0 to 170 x = 0.5) ferrites is depicted in Fig.3. It can be observed that all the Cr doped cobalt nano ferrite 171 172 samples exhibit the loss peak nearly at 5kHz frequency except  $Cr_{0.0}CoFe_2O_4$  (x=0.0) and  $Cr_{0.5}CoFe_{1.5}O_4$  (x=0.5) ferrite samples. In the case of  $Cr_{0.0}CoFe_2O_4$  and  $Cr_{0.5}CoFe_{1.5}O_4$  ferrite 173 174 samples, the dissipation factor (D) exhibits an inverse dependence on frequency which is a 175 normal behaviour of any ferrite materials. The appearance of the relaxation peak in Co-Cr mixed ferrites  $(Cr_{0.1}CoFe_{1.9}O_4 Cr_{0.2}CoFe_{1.8}O_4 Cr_{0.3}CoFe_{1.7}O_4 and Cr_{0.4}CoFe_{1.6}O_4)$  can be explained 176 according to Debye relaxation theory [25]. The loss peak in dissipation factor versus frequency 177 curves can be found when the hopping frequency of electrons is approximately matching with 178 the frequency of the applied electric field [26, 27]. The condition for observing a maximum in 179 dielectric loss of dielectric material is given by 180

181  $\omega \tau = 1$  .....(3)

182 Where  $\omega = 2\Pi f$ , f is the frequency of the applied electric field and  $\tau$  is the relaxation time.

From Fig. 3, it is important to notice that the relaxation peaks become broad, and the height of the peak also increases with increasing chromium concentration(*x*) up to *x*=0.3 after that decreases for *x*=0.4 composition. The increase in peak height with the increase of  $Cr^{3+}$  ion substitution is due to the increase of conductivity of the sample arising due to the increase of the Fe<sup>3+</sup>/Fe<sup>2+</sup> ions available for the conduction process. The increase in the peak width is due to strengthening of the dipole-dipole interactions which causes hindrance to the dipole rotation. The strength and frequency of the relaxation depend on characteristic property of dipolar relaxation.

190 The spinel ferrites have high resistivity and the relation between dissipation factor (D) and 191 resistivity ( $\rho$ ) is shown in the following equation.

192  $D = 1/\omega \epsilon' \epsilon_0 \rho \qquad \dots (4)$ 

193 where  $\omega$  is the angular frequency of applied electric field. From above relation, it can be 194 concluded that the dissipation factor (D) is approximately inversely proportional to the frequency 195 of applied electric field. Thus the dielectric loss of ferrites is expected to decrease with increase 196 in frequency.

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#### **198 3.1.3. AC conductivity**

199 Fig.4 shows the frequency dependence of ac conductivity for the Cr-Co ferrite system sintered at 950°C. All the curves exhibit the significant dispersion with frequency, which is an important 200 behaviour of ferrites. It can be seen that the ac conductivity ( $\sigma_{ac}$ ) seems to be nearly frequency 201 independent for all the compositions upto 0.1 MHz but above this frequency range,  $\sigma_{ac}$  rapidly 202 203 increases until it reaches a maximum value. The electrical conductivity in ferrites is mainly due to the hopping of electrons between the ions of the same element presented in more than one 204 205 valence state and distributed randomly over crystallography equivalent lattice sites [28]. On the application of ac electrical field, the hopping frequency of electrons gets boosted up. The 206 207 increase in the frequency of the applied electric field increases the hopping phenomenon which in turn enhances the ac conductivity of the ferrite ceramics. From Fig. 4 and table 1 it can also 208 noticed that Cr-doped Co-ferrites have higher conductivity than undoped ones. The number of 209 ferrous (Fe<sup> $^{2+}$ </sup>) ions at octahedral (B) sites plays a dominant role in the process of conduction in 210

these ferrite materials. The highest conductivity for composition x=0.3, may be attributed due to the presence of a large number of Fe<sup>2+</sup> ions in the Cr<sub>0.3</sub>CoFe<sub>1.7</sub>O<sub>4</sub> ferrite sample.

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#### 214 **4. CONCLUSIONS**

We have successfully synthesized  $Cr_x CoFe_{2-x}O_4(0.0 \le x \le 0.5)$  nano ferrites with single phase 215 spinel structure by sol-gel method. The dielectric properties have been investigated as a function 216 of composition and frequency. The room temperature dielectric constant decreases rapidly with 217 218 increasing frequency indicates the normal dielectric behaviour for all the samples, and this has been elucidated on the basis of interfacial polarization phenomena as discussed in Maxwell-219 Wagner theory. All the Cr<sup>3+</sup>substituted samples have high dielectric constant than the basic 220 cobalt composition without chromium. The ac conductivity was found to increase with an 221 222 increase in frequency and Cr concentration. Relaxation peaks were observed for the Crsubstituted Cobalt ferrite samples in dissipation versus frequency curves. It is also found that the 223 224 height and width of the relaxation peak increased for the Cr-Co ferrite system when Cr-content is between x=0.0 and x=0.3 and after that, decreased with the further addition of chromium and 225 this has been explained based on the Debye relaxation theory. The substitution of  $Cr^{3+}$  for  $Fe^{3+}$ 226 ions results in a significant impact on the dielectric behaviour of the Cr-Co ferrite system. The 227 low-temperature sintered Cr-Co ferrites exhibit good dielectric properties, as well as fine-228 grained nanostructures, making them become good materials for high-frequency applications 229 230 with high-performance and low-cost.

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**Fig.1:** Flow chart for the synthesis of  $Cr_x CoFe_{2-x}O_4$  nanoparticles by Sol-Gel method



**Fig. 2:** Variation of dielectric constant with frequency for  $Cr_xCoFe_{2-x}O_4$  (x = 0.0 to 0.5)







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**Fig. 4:** Variation of Ac conductivity with frequency for  $Cr_x CoFe_{2-x}O_4$  (0.0  $\le x \le 0.5$ )





- 292  $(0.0 \le x \le 0.5)$  samples at 1 kHz frequency
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**Table 1:** Dielectric parameters of  $Cr_x CoFe_{2-x}O_4$  ( $0.0 \le x \le 0.5$ ) at 1kHz

$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	294	S.No	Ferrite	Dielectric	Dissipation	ac
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	295		Composition	<b>constant</b> (ε')	factor(D)	$\begin{array}{ } \text{conductivity} \\ (\sigma_{ac} \times 10^{-6}) \end{array}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	296	1	Co Fe <sub>2</sub> O <sub>4</sub>	145.76	9.28	1.32
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		2	Cr <sub>0.1</sub> Co Fe <sub>1.9</sub> O <sub>4</sub>	465.36	20.85	9.85
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	297	3	Cr <sub>0.2</sub> Co Fe <sub>1.8</sub> O <sub>4</sub>	467.08	24.14	11.63
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		4	Cr <sub>0.3</sub> Co Fe <sub>1.7</sub> O <sub>4</sub>	729.56	20.33	15.04
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	298	5	Cr <sub>0.4</sub> CoFe <sub>1.6</sub> O <sub>4</sub>	337.60	24.49	8.55
000	299	6	Cr <sub>0.5</sub> Co Fe <sub>1.5</sub> O <sub>4</sub>	409.45	19.53	8.07