1	The static (hyper)polarizabilities of push-pull polyenes: A
2	theoretic study
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30	Abstract
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33	We have studied the static longitudinal polarizability, and first and second hyperpolarizabilities of
34	push-pull conjugated polyenes (NH_2 -($CH=CH$) _N - NO_2) within finite field scheme. Ab initio methods and
35	density functional theory (DFT) are used to evaluate electron correlation effects. B3LYP and CAM-B3LYP
36	are compared with second-order Møller-Plesset perturbation theory (MP2) and highly reliable coupled
37	cluster theory with single and double excitations (CCSD). Our study shows CAM-B3LYP fails to predict
38	correct correlation contributions, and MP2 is also unsatisfactory. CCSD is the only reliable computational
39	tool to this day.
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41	Keywords: polarizability; first and second hyperpolarizabilities; electron correlation effects; MP2, CCSD,
42	CAM-B3LYP
43	1. Introduction
лл	Organic conjugated systems are intensely investigated both experimentally and theoretically due
44	Organic conjugated systems are mensery investigated both experimentary and medicidentary due
45	to its bright future as new materials. Common conjugated structure, which results in delocalized electronic
46	structure, plays a key role in understanding the physical and chemical properties of organic conjugated
47	systems. It is early known electron correlation is essential to describe correctly the electronic structures of
48	organic conjugated systems [1-3].
49	Ab initio methods, such as many-body perturbation theory and coupled clusters method, are often
50	used to access the electron correlation effects [4, 5]. But the huge computational costs seriously restrict the
51	use of ab initio methods. DFT becomes more and more popular due to its simple picture and competitive

efficiency [6, 7]. Unfortunately, researchers found traditional functionals confront difficulties in presenting 53 a correct description for organic conjugated systems [3, 8-9]. Many efforts are devoted to developing new 54 functionals to fix this problem. CAM-B3LYP was shown to perform much better than traditional functionals such as B3LYP in studying conjugated systems [10-14]. 55

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56 Response properties of organic conjugated systems were intensely explored in past decades [15-20]. 57 Many significant developments are reported, but some important issues are still unclear. Yang and coworkers reported their studies on the response properties of push-pull polyenes with newly-developed 58

59 OEP and/or OEP-EXX exchange functionals, and they found OEP and/or OEP-EXX exchange functionals 60 perform much better than conventional functionals [15]. Champagne and coworkers studied electron 61 correlation effects on the first hyperpolarizability of push-pull pi-conjugated systems. They found 6-31G(d) 62 basis set is an optimal choice for polyene linkers [16]. Mohammed and coworkers discussed finite field 63 method as an effective tool for calculating the nonlinear optical properties [17-18]. We studied previously 64 electron correlation effect on static response properties of pristine polyenes [2-3, 19-20]. And we report 65 here our recent study on response properties of a typical donor-acceptor derivatives of all-trans polyenes, 66 also known as push-pull polyenes, NH₂-(CH=CH)_N-NO₂.

67 **2.** Computational Details

The molecular geometries have an important impact on the properties. We here focus on the performances of different electronic structure methods in the same molecular geometry. The molecular geometries of NH_2 -(CH=CH)_N-NO₂ are optimized by Hartree-Fock method (HF) with 6-31G basis set. Basis sets of medium size 6-31G(d) and cc-pVDZ are used to compute molecular energies in this study. The 6-31G(d) basis set were reported to be the optimal choice for study of nonlinear optical properties of push-pull polyenes [16]. Correlated electronic structure methods such as MP2, CCSD, B3LYP and CAM-B3LYP are used to evaluate electron correlation effects.

Finite field scheme is chosen to calculate static response properties of NH_2 -(CH=CH)_N-NO₂ [21-25]. First, molecular energies are computed in zero or given electric fields applied along the direction of chains, and the response properties such as longitudinal polarizability (α), first hyperpolarizability (β) and second hyperpolarizability (γ) are obtained with numerical derivatives of energy with respect to electric field as

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$$\alpha = -\left(\frac{\partial^2 E}{\partial h^2}\right)\Big|_{h=0} = -\frac{E(h) + E(-h) - 2E(0)}{h^2}$$
(1)

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$$\beta = -\left(\frac{\partial^3 E}{\partial h^3}\right)\Big|_{h=0} = -\frac{E(2h) - 2E(h) + 2E(-h) - E(-2h)}{2h^3}$$
(2)

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$$\gamma = -\left(\frac{\partial^4 E}{\partial h^4}\right)\Big|_{h=0} = -\frac{E(2h) - 4E(h) + 6E(0) - 4E(-h) + E(-2h)}{h^4}.$$
 (3)

The electric field used in this study reads $h = 8 \times 10^4$ a.u., and E(h) is the molecular energy in electric field *h*, and so on. Optimized molecular geometries and molecular energies are obtained with Gaussian 09 software package [26]. An extrapolation scheme [3] is used to give the trend as the chain approaches
infinite. The properties are obtained with Hartree-Fock method (HF) and correlated electronic structure
methods respectively. One can get contributions from electron correlation simply by comparing
correlated properties with HF results.

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89 3. Results and Discussion

90 The 6-31G(d) and cc-pVDZ basis sets are found to make much less difference in this study, and only 91 results obtained by 6-31G(d) are discussed for most cases. The polarizability obtained as second 92 derivatives of the energy with respect to electric field is plotted in Fig. 1. B3LYP is well known to fail to 93 present proper treatment for electron correlation effect. From Table I and Fig. 1, one can see that CCSD 94 and CCSD(T) predict negative electron correlation contributions for polarizability, except for N = 2. 95 B3LYP, CAM-B3LYP, and MP2 predict positive electron correlation contributions. CAM-B3LYP performs 96 better than B3LYP, comparing to CCSD. It is worth mentioning MP2 give wrong predictions for 97 correlation contributions too. The trend predicted by MP2 is obviously much better than B3LYP and CAM-B3LYP, comparing to CCSD. CCSD results are very close to CCSD(T) ones for N = 2, 3, 4, and 98 99 are used as a reference thereafter. The extrapolation result is plotted in Fig. 2, from which damping effect 100 of electron correlation effect on polarizability [3] can be found.

From Table II and Fig. 3, one can see both ab initio methods and DFT predict positive electron correlation contributions for β . CAM-B3LYP performs better than B3LYP, but both B3LYP and CAM-B3LYP give a trend of being steeper than CCSD. MP2 predicts a better trend, but overestimates the electron correlation contributions. As is known, the first hyperpolarizabilities of all-trans polyenes are zero for symmetric structure. The push-pull structure breaks the symmetry and leads to a nonzero β . The damping effect of electron correlation effect on first hyperpolarizability can be found from Fig. 4.

107 One can easily see from Table III and Fig. 5 that the second hyperpolarizabilities obtained by MP2 for 108 small systems (N = 2, 3, 4) are close to those obtained by CCSD and CCSD(T), but MP2 predicts a steeper

109 trend. B3LYP and CAM-B3LYP predict a worse trend as the chain length increases. A damping effect of

110 electron correlation for γ can be seen from Fig. 6.

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112 **4.** Conclusions

Electron correlation effect is essential in studying polarizability and first and second hyperpolarizabilities of organic conjugated systems. Though CAM-B3LYP performs better than B3LYP, it is still far from satisfactory. Searching for proper exchange and correlation functionals describing organic conjugated systems are still much valuable and challenging. MP2 is also questionable, especially for β and γ , and is therefore an inappropriate benchmark for developing new functionals for conjugated systems, at least for medium or large conjugated systems. CCSD with appropriate basis set are still the only reliable choice for studying the response properties of conjugated systems.

120 The damping effect of electron correlation on the second hyperpolarizability of pristine polyenes [3] is 121 also found in push-pull polyenes. And the damping effect of electron correlation on first 122 hyperpolarizability is found for push-pull polyenes.

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N	2	3	4	5	6
6-31G(d)					
HF	70.32	78.67	87.14	94.61	100.91
MP2	76.16	83.44	90.58	96.56	101.32
CCSD	72.06	76.23	79.99	82.81	84.91
CCSD(T)	72.63	76.98	81.31		
B3LYP	78.64	91.83	107.21	123.07	138.62
CAM-B3LYP	75.74	86.14	97.29	107.68	116.77
cc-pVDZ					
HF	70.42	79.06	87.86	95.66	102.28
MP2	75.75	83.58	91.42	98.18	103.72
CCSD	71.68	76.75	81.50	85.30	88.24
CCSD(T)	72.21	77.50	82.86		
B3LYP	78.48	91.71	107.11	122.98	138.54
CAM-B3LYP	75.59	86.12	97.44	108.00	117.31

168 TABLE I Static longitudinal polarizability per unit cell α_L/N of NH₂-(CH=CH)_N-NO₂ (in a.u.).

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170 TABLE II Static longitudinal first-order hyperpolarizability per unit cell β_L/N of NH₂-(CH=CH)_N-NO₂ (in 10² a.u.).

		JIII	J P P P P P P P P P P	2 ()	N - 2 (
N	2	3	4	5	6
6-31G(d)					
HF	6.80	11.88	17.64	23.10	27.55
MP2	15.15	26.56	39.24	51.10	60.70
CCSD	14.60	23.82	31.84	37.37	40.25
CCSD(T)	15.49	24.83	33.08		
B3LYP	8.35	16.54	29.38	47.89	72.78
CAM-B3LYP	9.14	17.29	28.28	41.12	54.41
cc-pVDZ					
HF	6.35	11.22	16.83	22.18	26.70
MP2	13.92	24.67	36.92	48.75	58.68
CCSD	12.94	21.61	29.77	35.90	39.68
CCSD(T)	13.61	22.42	30.95		
B3LYP	7.91	15.86	28.28	46.15	70.13
CAM-B3LYP	8.60	16.39	26.93	39.28	52.13

173 TABLE III Longitudinal second-order hyperpolarizability per cell γ_L/N of NH2-(CH=CH)_N-NO2 (in 10⁴ a.u.)

N	2	3	4	5	6	
6-31G(d)						
HF	3.1518	9.4760	21.5609	39.3065	60.9501	
MP2	7.1885	23.5264	54.7753	100.3947	155.4416	
CCSD	9.4954	27.3940	54.6792	86.3269	116.1628	
CCSD(T)	11.2739	31.2911	61.1759			
B3LYP	1.4093	6.6420	22.4183	60.6237	140.0925	
CAM-B3LYP	2.8426	11.2753	31.9916	71.1198	131.6961	
cc-pVDZ	cc-pVDZ					
HF	3.0089	9.0068	20.5871	36.8609	59.1967	
MP2	6.6044	21.6309	51.0172	95.0994	149.8380	
CCSD	8.2236	23.9669	51.1046	82.0634	114.8293	
CCSD(T)	9.6284	27.1768	57.9478			
B3LYP	1.5666	6.9540	22.8413	60.8804	139.2329	
CAM-B3LYP	2.8670	11.0342	31.0094	68.6723	127.0823	



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177 Fig. 1. Static longitudinal polarizability α_L/N of NH₂-(CH=CH)_N-NO₂ (in a.u.) obtained by different methods with 6-31G(d)

178 basis set.



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180 Fig. 2. The ratio α_{CCSD} : α_{HF} as a function of the inverse chain length for NH₂-(CH=CH)_N-NO₂. Values at $N \rightarrow \infty$ were

¹⁸¹ obtained by a quadratic extrapolation.



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183 Fig. 3. Static longitudinal first-order hyperpolarizability β_L/N of NH₂-(CH=CH)_N-NO₂ (in 10² a.u.) obtained by different

184 methods with 6-31G(d) basis set.





186 Fig. 4. The ratio β_{CCSD} : β_{HF} as a function of the inverse chain length for NH₂-(CH=CH)_N-NO₂. Values at $N \rightarrow \infty$ were obtained





189 Fig. 5 Static longitudinal second-order hyperpolarizability γ_L/N of NH₂-(CH=CH)_N-NO₂ (in 10⁴ a.u.) obtained by different







192 Fig. 6. The ratio γ_{CCSD} : γ_{HF} as a function of the inverse chain length for NH₂-(CH=CH)_N-NO₂. Values at $N \rightarrow \infty$ were obtained

193 by a quadratic extrapolation.