

**Accumulation of Heavy Metals in Soil and Maize Plant (*Zea mays*) in the Vicinity of
Two Government Approved Dumpsites in Benin City, Nigeria**

Abstract

Soil contamination by heavy metals is of great concern with respect to human health risks, groundwater contamination, phytotoxicity to plants, adverse effects on microbial activity and diversity, long-term effects on soil fertility and depreciation of land. Soil samples were obtained with the aid of soil Auger within a depth of 0 – 20 cm from the vicinity of the two selected dumpsites in Benin City, Edo State Nigeria. The soil samples were assessed for some physico-chemical properties using standard methods. Maize plants found growing in the dumpsite and control areas were also sampled, partitioned into leaves, stems, and roots prior to analysis in the laboratory for heavy metals determination using atomic absorption spectrophotometer (AAS). The soil sample showed Zn, Cu, Cd, Pb and Cr levels ranging from 42.66-243.81, 2.16-21.41, 0.35-2.59, 1.11-7.76 and 2.99-10.99 mg/kg respectively. Pollution indices such as contamination factor, contamination degree and pollution load index of the heavy metals analyzed were calculated and their values varied among the heavy metals and between the dumpsites. Concentrations of the metals in the dumpsite soil and plant were found to be higher compared to those of the control sites. Significant differences of heavy metals accumulations were observed per plant parts, roots having the highest concentrations. The translocation factor, biological concentration factor and biological accumulation coefficient values of the plant species varied for all the metals. These results imply that the dumpsites have associated human health and ecological risks.

Keywords: Dumpsites, Heavy Metals, translocation factor, Soil Maize (*Zea mays*).

Introduction

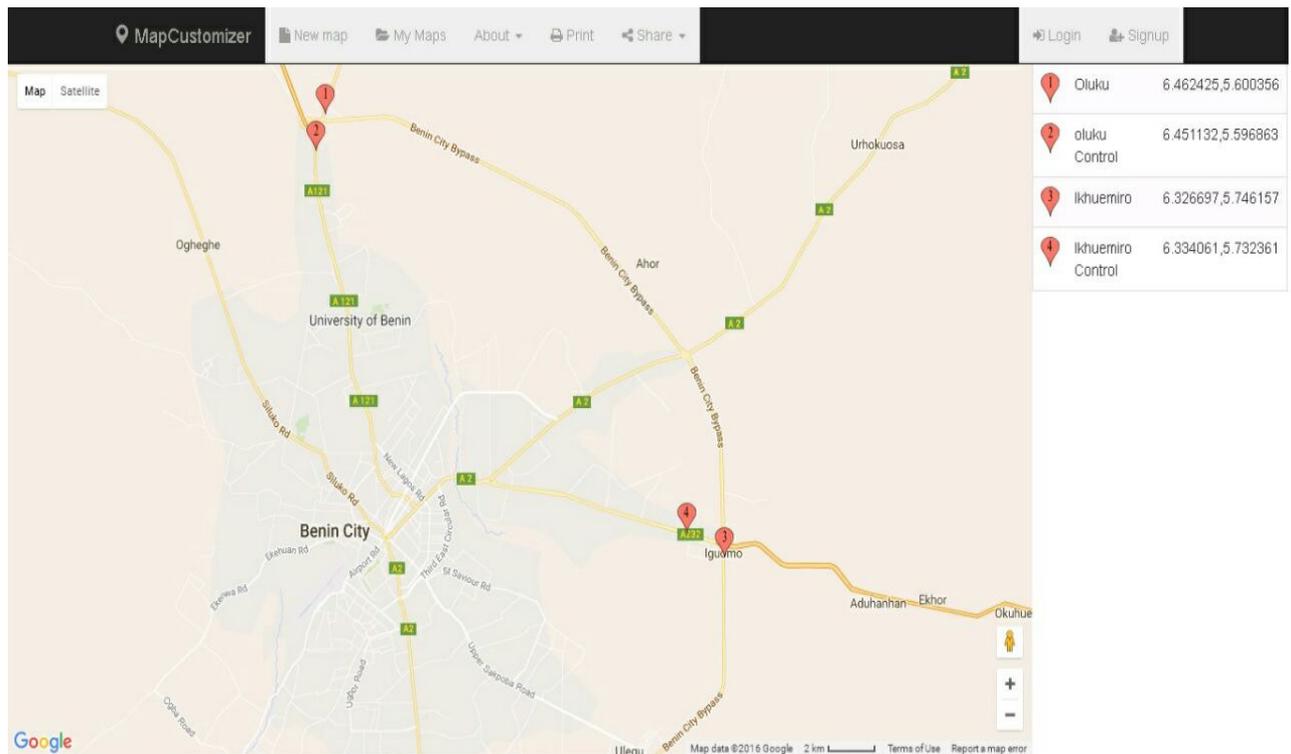
Dumpsites exist throughout developing countries. Most of these dumping sites are uncontrolled and years old, having grown over time from small dumps to large, unmanaged waste sites (Joan *et al.*, 2016). This constitutes serious health and environmental concerns because of the effects on the host soils, crops, animal and human health. Many cities in Nigeria have developed without proper planning and it has led to the presence of open dumps within built-up areas inhabited by millions of people (Amadi and Nwankwoala, 2013). Consequently, such waste dumps become point source for soil pollution as they serve as host for leachate from dumpsites. Composition of solid wastes in major cities in Nigeria comprises domestic garbage, wood, agricultural waste, industrial waste, hospital waste, polythene bags, plastics, broken glasses, abandoned automobiles, demolition waste, ash, dust, human and

35 animal waste (Awomeso *et al.*, 2010). The proper wastes disposal has been a serious problem
36 in Benin City and most cities in Nigeria. Solid and fluid wastes generation and their poor
37 disposal mechanism in the urban areas of most developing countries have become a threat to
38 the environment (Amadi *et al.*, 2010). The contamination of soil with heavy metals is an
39 environmental concern because accumulated metals may have adverse effects on soil
40 ecology, agricultural production, animal and human health as well as groundwater quality
41 (Okiemen *et al.*, 2011). While many heavy metals are essential elements at low levels of
42 concentration, they can exert toxic effects at concentrations higher than permitted in the
43 environment (Anegbe *et al.*, 2014; Mmolawa *et al.*, 2011). They may be volatilized to the
44 atmosphere, especially during dry seasons (Okuo and Okolo, 2011). In Benin City
45 metropolis, most of the dumpsites are used as fertile soils for the cultivation of some fruits,
46 food crops and vegetables due to the high cost of fertilizer. Some farmers collect the
47 decomposed parts of the dumpsites and apply to their farms as manure. These cultivated
48 plants take up these heavy metals either as mobile ion in the soil solution through their roots
49 or through their leaves thereby making it unfit for human consumption (Ganesh *et al.*, 2010).
50 Recent studies have also reviewed that waste dumpsite can transfer significant levels of these
51 toxic and persistent metals into the soil environment. Eventually these metals are taken up by
52 plant part and transfer same into the food chain (Anhwange and Asemave, 2013).
53 Consequently, higher soil heavy metals concentration can result in higher levels of uptake by
54 plants. Although, the rate of metal uptake by crop plants could be influenced by factors such
55 as metal species, plants species, plant age, plant part soil composition, geographic and
56 atmospheric conditions (Dulama *et al.*, 2012). Transfer of heavy metal from soils to plants
57 has been proved as an efficient way for removal of these heavy metals through harvestable
58 plant parts such as roots, stems and leaves (Malik *et al.*, 2010). However, the metal
59 availability and toxicity to plant can be determined by the soluble and exchangeable fraction
60 of metals in particular (OK *et al.*, 2011). Intake of heavy metals via the crop-soil system has
61 been regarded as the predominant pathway of human exposure to toxic metals (Liu *et al.*,
62 2007), and is normally chronic. This is because these metals are non-biodegradable and can
63 undergo global ecological circles (Opaluwa *et al.*, 2012). Thus, it become necessary to assess
64 the uptake of heavy metals by maize plant (*zea mays*) in two government approved dumpsites
65 in Benin City, Nigeria in order to determine their potential hazards to human beings and
66 animals. The objective of this study was to determine and compare the content of heavy
67 metals in various parts of the maize plant, namely the leaves, stems and roots compared to the
68 levels in the soil around Oluku and Ikhuenirolu dumpsite zones.

69 **Materials and Methods**

70 **Study Area**

71 Ikhuenirol dumpsite is the largest and the major open dumpsite site in Benin metropolis. It has
72 been in operation as a disposal facility; permitted to receive commercial and municipal solid
73 waste. However, the absence of waste management and sorting systems lead to the dumping
74 of industrial waste into the dumpsite (Ighodaro *et al.*, 2015). **Ikhuenirol and Oluku dumpsites**
75 **comprise** household materials, hospitals disposables, metals scraps, polyethylene bags and
76 papers, plants materials and debris among other substances. They also consist of scavengers
77 that are involved in sorting some of these materials for re-us



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79 **Fig 1: Map of Benin City Showing the Sampled Sites**

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81 **Collection of Soil Samples**

82 Soil and plant samples were collected from two government approved dumpsites and their
83 control sites within Benin City, namely Oluku dumpsites (A), Oluku control site (B),
84 Ikhuenirol dumpsites (C) and Ikhuenirol control site (D). In this research, Soil samples at 0- 20
85 cm depth from rhizosphere of the maize plant were taken from each site from where plant
86 sample was rooted. **At each site, three different points were chosen using cluster random**

87 **sampling technique to collect the sample.** The soil samples were obtained by the use of soil
88 auger, and then blended (mixed) to obtain a representative sample. The samples were air
89 dried and ground to pass through a 2 mm sieve and used for both physico-chemical analysis
90 (Anege and Okuo, 2013). Prior to the analysis of plant material, leaf, stem and roots of
91 plants were separated and carefully washed with tap and deionized water in order to remove
92 soil or dust deposits. Then the plant samples were oven-dried at 70°C to constant weight,
93 pulverized, passed through 2 mm steel sieve and weighed in order to determine the heavy
94 metals concentrations by atomic absorption spectrometry (Tsvetomil *et al.*, 2013). The
95 control soil and plant samples were taken at about 1500 m away from each of the dumpsites.

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97 **Analysis of the Soil and Plant Samples**

98 The pH and the CEC were determined as described by Anege and Okuo (2013). The
99 hydrometer method described by Ugbune and Okuo (2011) were used in evaluating the
100 particle size. The method described by Anege *et al.* (2017) were used to determine the
101 organic carbon content, while the total heavy metals determination was carried out using the
102 Tessier's method described by Okuo *et al.* (2016). **According to the method, 5 ml of aqua**
103 **regia (BDH, England) and 1 ml of perchloric acid (BDH, England) were added to 1 g of soil**
104 **sample in a 150 ml digestion tube and digested on a heating digester until white fumes of**
105 **perchloric acid appeared. The tube was cooled and the sides rinsed with distilled water and**
106 **then filtered through a Whatman 1 filter paper into a 100 ml volumetric flask. The volume**
107 **was made up with distilled water** All glasswares used were soaked and washed with chromic
108 acid and rinsed with distilled water. Bulk scientific standard solution was used to calibrate the
109 Atomic Absorption Spectrometer (Pg A500 model, USA). Procedural blank samples were
110 subjected to similar extraction method using the same amount of reagents.

111 **Results and Discussion**

112 The physico-chemical properties of the soil samples at various sites are shown in Table 1.
113 Soil pH plays a major function in the sorption of heavy metals as it directly controls the
114 solubility and hydrolysis of metal hydroxides, carbonates and phosphates (Tokalioglu *et al.*,
115 2006). The pH of the studied areas ranged from neutral to moderately alkaline. Most mineral

116 and nutrients are more soluble or available in acidic soils than in neutral or slightly alkaline
 117 soils. Soils tend to become acidic as a result of rain water leaching away basic ions (Ca^{2+} ,
 118 Mg^{2+} , K^+ and Na^+) (Bickelhaupt, 2015).

119 **Table 1: Physico-chemical Properties and Total Heavy Metal Concentration of the Soil**
 120 **Samples.**

Parameters	Units	Dumpsite soil (Oluku)	Control soil (Oluku)	Dumpsite soil (Ikhueni)	Control soil (Ikhueni)
pH		8.10±0.01	7.14±0.02	7.40±0.00	7.05±0.02
CEC	meq/100g	12.82±1.22	5.22±0.15	11.21±0.30	5.20±0.22
TOC	%	1.90±0.26	1.56±0.14	1.74±0.15	1.23±0.30
OM	%	3.28±0.26	2.69 ±0.14	3.00±0.15	2.12±0.30
Clay	%	10.71±0.12	16.71±0.22	28.24±0.40	31.72±0.30
Silt	%	2.09±0.11	2.09±0.10	2.56±0.15	2.49±0.17
Sand	%	87.20 ±0.35	81.20±0.50	69.20±0.31	65.79±0.25
Zn	mg/kg	156.78±2.35	58.93±2.07	243.81±5.00	42.66±2.00
Cu	mg/kg	20.17±1.44	2.30±0.24	21.41±1.71	2.16±0.22
Cd	mg/kg	1.40±0.19	0.35±0.08	2.59±0.42	0.73±0.10
Pb	mg/kg	6.36±0.25	1.32±0.56	7.76±0.55	1.11±0.05
Cr	mg/kg	7.92±1.12	4.68±0.50	10.99±0.64	2.99±0.23

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 122 The CEC parameter particularly measures the ability of soils to allow for easy exchange of
 123 cations between its surface and solution. The soil samples from the two dumpsites and their
 124 controls shows a low CEC which indicates that they are more likely to develop deficiencies
 125 in potassium (K^+), magnesium (Mg^{2+}) and other cations, while high CEC soils are less
 126 susceptible to leaching of these cations (CUCE, 2007). The low values of the CEC were
 127 attributed to high sandy nature of the soil samples. Textural analysis showed the
 128 preponderance of sand fraction, followed by clay then silt, thus classifying the parent soil as
 129 loamy sand. Sandy soils are known to have a poor retention capacity for both water and
 130 metals (Wuana *et al.*, 2010). Low organic matter (2.12 – 3.28 %) was observed in all the soil
 131 samples. The low organic matter content of the dumpsite and the control soil samples is an

132 indication that these soils will have low adsorption strength and an increased metal mobility
133 and bioavailability (Udeigwe, 2010).

134 The soil samples showed the presence of Zn, Cu, Cd, Pb and Cr in all the sites analysed This
135 could be attributed to the availability of wastes containing those metals at the dumpsites
136 which are eventually leached into the underlying soils. The concentrations of heavy metals
137 are higher in each of the dumpsites than their respective control sites. Liu *et al.* (2007),
138 observed that heavy metal concentration in soils is usually high near the sources, and decline
139 with both distance and depth due to physical dilution and increasing limits in mobility. In
140 overall terms, the results of the present study suggested that the five metals decline in the
141 following order: Zn>Cu>Cr>Pb>Cd for each of the two sites. The results of heavy metals
142 obtained from the analysis also indicate that the concentrations of the heavy metals were
143 found to be higher in Ikhueniro dumpsite when compared with Oluku dumpsite. The higher
144 concentration could be as a result of waste carrying more concentrations of these metals in
145 Ikhueniro dumpsite compared to that of Oluku dumpsite.

146 **Assessment of Metal Contamination**

147 **Contamination Factor (CF)**

148 The level of contamination of soil by metal is expressed in terms of a contamination factor
149 (CF) calculated as:

$$150 \quad CF = \frac{C_m \text{ Sample}}{C_m \text{ Background}} \quad (1)$$

151 $C_m \text{ Sample}$ = metal concentration in Sample

152 $C_m \text{ Background}$ = metal concentration in background or control Sample. (Lin *et al.*, 2009)

153 Where the contamination factor $CF < 1$ refers to low contamination; $1 \leq CF < 3$ means
154 moderate contamination; $3 \leq CF \leq 6$ indicates considerable contamination and $CF > 6$
155 indicates very high contamination.

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161 **Table 2: Contamination Factor of Each Metal in the Two Dumpsites**

Metals	Oluku Dumpsite	Oluku Classes	Ikhueniro Dumpsite	Ikhueniro Classes
Zn	2.66	Moderate contamination	5.72	Considerable contamination
Cu	8.77	Very high contamination	9.91	Very high contamination
Cd	4.00	Considerable contamination	3.55	Considerable contamination
Pb	4.82	Considerable contamination	6.99	Very high contamination
Cr	1.69	Moderate contamination	3.68	Considerable contamination

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163 From the results of the contamination factors, the soil samples may be classified as
 164 moderately contaminated with respect to Zn and Cr, considerably contaminated with respect
 165 to Cd and Pb, and very highly contaminated with respect to Cu in the vicinity of Oluku
 166 dumpsite. The soil in the vicinity of Ikhueniro dumpsite may be classified as considerably
 167 contaminated with respect to Zn, Cd and Cr, and very highly contaminated with respect to Cu
 168 and Pb. It was also observed that the contamination factor of each metal in the vicinity of
 169 Ikhueniro dumpsite is greater than the contamination factor of the same metal in Oluku
 170 dumpsite except for cadmium which proved otherwise. This is because contamination factor
 171 is directly proportional to the concentration of each metal in the sediment, and all the metals
 172 in the vicinity of Ikhueniro dumpsite except cadmium have higher sediment concentrations
 173 than their counterpart in Oluku dumpsite.

174 **Degree of Contamination**

175 The sum of contamination factors for all elements examined represents the contamination
 176 degree (Cdeg) of the environment. Using the sum of contamination factors obtained in table 2
 177 for all elements in each dumpsite.

178 The contamination degree (Cdeg) values were 21.94 and 29.84 for Oluku and Ikhueniro
 179 dumpsites respectively. Therefore, the soils in the vicinity of the two dumpsites where the
 180 plant samples were gotten can be classified as having considerable degree of contamination

181 according to Dasaram *et al.* (2011). However, it is worthy of note that the degree of
 182 contamination of soil in the vicinity of Ikhueniro dumpsite is greater than that of Oluku
 183 dumpsite; this might be attributed to the higher concentration, and hence higher
 184 contamination factor of each metal at the vicinity of Ikhueniro dumpsite than that of Oluku
 185 dumpsite.

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187 **The Pollution Load Index (PLI)**

188 Generally, pollution load index (PLI) as reported by Harikumar *et al.* (2009), is as follows:

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$$PLI = \sqrt[n]{Cf1 \times Cf2 \times Cf3 \times Cf4 \dots \dots \dots Cf n} \quad (2)$$

190 Where, CF = contamination factor, n = number of metals

191 The PLI value of > 1 is polluted, whereas < 1 indicates no pollution (Harikumar *et al.*,2009).

192 The PLI values as calculated for both sites were 3.77 and 5.53 for Oluku and Ikhueniro
 193 dumpsites respectively. This showed a strong sign of pollution deterioration by the five
 194 measured metals in both sites since their PLI values are greater than 1. The PLI value was
 195 higher in the vicinity of Ikhueniro dumpsite than that of Oluku dumpsite. Hence, the vicinity
 196 of Ikhueniro dumpsite may cause more pollution to the environment than Oluku dumpsite.

197 **Uptake of Heavy Metals by Plants in the Vicinity of the Open Dumpsites in Benin City**

198 Although certain trace elements are essential in plant physiology, plants growing in a polluted
 199 environment can bioaccumulate trace elements at very high concentration in their edible parts
 200 and can present public health concerns (Quierolo, *et al.*, 2000).

201 **Table 3: Uptake of Zn, Cu, Cd, Pb and Cr (mg/kg) by Different Parts of the Maize**
 202 **Plants Collected from Oluku Dumpsite and its Control**

Metals	Roots	Stems	Leaves	Shoot
Zn	62.30(17.64)	25.14(7.36)	7.37(3.11)	32.51(10.47)
Cu	9.50(0.78)	3.21(0.36)	1.98(<0.05)	5.19(0.36)
Cd	0.62(0.10)	0.21(<0.05)	0.26(0.09)	0.47(0.09)

Pb	2.05(0.32)	0.62(<0.05)	0.35(<0.05)	0.97(0.00)
Cr	3.08(0.90)	2.15(0.72)	1.07(0.54)	3.22(1.26)

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204 **Table 4: Uptake of Zn, Cu, Cd, Pb and Cr (mg/kg) by Different Parts of the Maize**
 205 **Plants Collected from Ikhueniro Dumpsite and its Control.**

Metals	Roots	Stems	Leaves	Shoot
Zn	89.55(13.83)	40.94(5.79)	22.87(3.88)	63.81(9.67)
Cu	10.30(0.44)	4.98(0.26)	1.08(<0.05)	6.06(0.26)
Cd	0.96(0.19)	0.54(0.09)	0.60(0.15)	1.14(0.24)
Pb	2.62(0.21)	1.05(0.10)	0.26(<0.05)	1.31(0.10)
Cr	5.07(0.72)	2.02(0.29)	1.77(<0.05)	3.79(0.29)

206 **Key:** Shoot = Stem + Leaf. **Note:** the values in bracket are for control.

207 Among all plant parts, roots accumulated the highest metal contents in all the sites (Table 4).
 208 Highest uptake of metals in roots compared to other parts was also observed in similar studies
 209 (Sekara *et al.*, 2005). This could be due to roots direct exposure to the contaminated soil. It
 210 was also found that plant parts from the dumpsites have higher metal concentrations than
 211 control sites. This might be attributed to higher concentrations of metals in the dumpsite than
 212 the control site (Amusan *et al.*, 2005; Li *et al.*, 2007). The results also indicated that the levels
 213 of metals in plants are dependent upon their concentrations in their habitual soil environment
 214 (Ayari *et al.*, 2010; Malik *et al.*, 2010).

215 **Transferability of Metals.**

216 Biological Accumulation Coefficient (BAC) = $\frac{[\text{Metals}]_{\text{shoot}}}{[\text{Metals}]_{\text{soil}}}$ (3) (Li *et al.*, 2007).

217 Biological Concentration Factor (BCF) = $\frac{[\text{Metals}]_{\text{root}}}{[\text{Metals}]_{\text{soil}}}$ (4) (Yoon *et al.*, 2006).

218 Translocation Factor (TF) = $\frac{[\text{Metals}]_{\text{shoot}}}{[\text{Metals}]_{\text{root}}}$ (5) (Cui *et al.*, 2007).

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220 **Table 5: BAC, BCF and TF for Zn, Cu, Cd, Pb and Cr in the Maize Plant at Oluku**

221 **Dumpsite and its Control.**

Metals	Biological Accumulation Coefficient (BAC)	Biological Concentration Factor (BCF)	Translocation Factor (TF)
Zn	0.21(0.18)	0.40(0.30)	0.52(0.59)
Cu	0.26(0.16)	0.47(0.34)	0.55(0.46)
Cd	0.34(0.26)	0.44(0.29)	0.76(0.90)
Pb	0.15(0.00)	0.32(0.24)	0.47(0.00)
Cr	0.41(0.27)	0.39(0.19)	1.05(1.40)

222 **Note:** the values in bracket are for control

223 For Oluku dumpsite, the biological accumulation coefficient (BAC) decreased in the
 224 following order Cr>Cd>Cu>Zn>Pb, the biological concentration factor (BCF) decreased as
 225 follows Cu>Cd> Zn>Cr>Pb while the translocation factor (TF) followed the order of
 226 Cr>Cd>Cu>Zn>Pb. For Oluku control site, the BAC decreased in the following order
 227 Cr>Cd>Zn>Cu>Pb, the BCF decreased as follows Cu > Zn >Cd >Pb>Cr while the TF
 228 followed the order of Cr>Cd>Zn>Cu>Pb. Pb was not detected in the various plant parts but
 229 occurred in low concentrations in the roots of the maize plant. The TF of the maize plant is
 230 greater than 1 with respect to chromium shows the special ability of the maize plant to absorb
 231 chromium from soils and transport and store it in its above-ground part (Wei *et al.*, 2002).
 232 BAC was categorised as: < 1 excluder, 1-10 accumulator and > 10 hyperaccumulator (Ma
 233 *et al.*, 2001). Hence, using the results obtained for BAC in table above, it could be
 234 suggested that the maize plant is an excluder with respect to all the heavy metals analysed in
 235 Oluku dumpsite and its control site because all the BAC values were less than 1. TF > 1
 236 signifies that the plant effectively translocates heavy metals from roots to the shoots (Baker
 237 and Brooks, 1989). Hence it could be observed from Tables 5 that the maize plant effectively
 238 translocate chromium from roots to the shoots since the TF of the maize with respect to
 239 chromium is greater than 1 both in Oluku dumpsite and its control site.

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245 **Table 6: BAC, BCF and TF for Zn, Cu, Cd, Pb and Cr in the Maize Plant at Ikhueniro**
 246 **Dumpsite and its Control.**

Metals	Biological Accumulation Coefficient (BAC)	Biological Concentration Factor (BCF)	Translocation Factor (TF)
Zn	0.26(0.23)	0.37(0.32)	0.71(0.70)
Cu	0.28(0.12)	0.48(0.20)	0.59(0.59)
Cd	0.44(0.33)	0.37(0.26)	1.19(1.26)
Pb	0.17(0.09)	0.34(0.19)	0.50(0.48)
Cr	0.34(0.10)	0.46(0.24)	0.75(0.40)

247 **Note:** the values in bracket are for control.

248 For Ikhueniro dumpsite, the biological accumulation coefficient (BAC) decreased in the
 249 following order Cd> Cr>Cu>Zn>Pb, the biological concentration factor (BCF) decreased as
 250 follows Cu> Cr>Cd= Zn> Pb while the translocation factor (TF) followed the order of
 251 Cd>Cr> Zn>Cu> Pb. For Ikhueniro control site, the BAC decreased in the following order
 252 Cd>Zn>Cu> Cr>Pb, the BCF decreased as follows Zn >Cd > Cr >Cu>Pb while the TF
 253 followed the order of Cd>Zn>Cu>Pb >Cr. Using the results obtained for BAC in table 6, it
 254 could be suggested that the maize plant is an excluder with respect to all the heavy metals
 255 analysed in Ikhueniro dumpsite and its control site (Ma *et al.*, 2001). Also considering the TF
 256 values in table 6, it could be observed that the maize plant effectively translocate cadmium
 257 from roots to the shoots since the TF of the maize with respect to cadmium is greater than 1
 258 (Baker and Brooks, 1989). High accumulation of heavy metals in roots and low translocation
 259 in shoots may indicate appropriateness of a plant species for phytostabilisation (Archer and
 260 Caiwell, 2004; Malik *et al.*, 2010). Phyto-stabilization process depends on roots' ability to
 261 limit the heavy metals' mobility and bioavailability in the soils and these occurs through
 262 sorption, precipitation, complexation or metal valance reduction (Ghosh and Singh, 2005).
 263 High root to shoot translocation of metals indicate that the plants have vital characteristics to
 264 be used in phytoextraction of the metals (Malik et al., 2010).

265 **Conclusion**

266 The two dumpsites studied in Benin City had revealed that indiscriminate disposal of wastes
 267 such as municipal wastes, industrial wastes, agricultural wastes, etc are major sources of soil
 268 contamination and pollution by heavy metals. A knowledge of the total concentration of these
 269 heavy metals through soil analysis (as indicator) could be considered as a starting point for
 270 evaluating the degree of pollution as investigated in this study. All of the heavy metals

271 studied were found to accumulate mainly in the roots of the maize plant. It can be concluded
272 that Ikhueniro dumpsite area is more polluted than Oluku dumpsite area and that the
273 indiscriminate disposal of wastes on these land had contributed to the increment in
274 concentrations of these metals in the land. All of the heavy metals studied were found to
275 accumulate mainly in the roots of the maize plant. Linearity dependence was found between
276 the total heavy metal content in the soil and in the plant for all the elements studied. This may
277 suggest that, plant absorption is controlled by the content of heavy metals in the soil solution
278 and also by the content that is bioavailable in the soil.

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