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SDI FINAL EVALUATION FORM 1.1

PART 1:

Journal Name:	**Asian Journal of Chemical Sciences**
Manuscript Number:	**Ms_AJOCS_31789**
Title of the Manuscript:	**A study of Polarographic Characteristics, Kinetic parameter and Stability Constant of [Zn(II)-antibiotics-vitamin-BX] complexes**
Type of Article:	**Original research paper**

PART 2:

FINAL EVALUATOR'S comments on revised paper (if any)	**Authors' response to final evaluator's comments**
Still some points are being described: 1) How the authors decided or determined or found or understand (by which methods or studies) that ligand structures are 1:1:1, 1:1:2 and 1:2:1 complexes? I understand that these are ligands but the origin of this values are not given. Which methods were used for this results or what values indicate that ligand structures are 1:1:1, 1:1:2 and 1:2:1 complexes. As a result, who says these are 1:1:1, 1:1:2 and 1:2:1 complexes. 2) In Figure 1: the curves straight line and increasing potentially should be concluded. I ask; what is the meaning of the curves straight line or increasing potentially. Tabulation is not a good expression because tabulation is also need an explanation. So, an explanation should be insert to the text. Curves straight line means what? increasing potentially means what? 3) Authors say that "The value of rate constant (k) showed that the electrode process were quasireversible". Authors say that in the previous sentence: "It is obvious from the value of α that the values varied from [Zn(II)-neomycin-vitamin-B_X] 0.42 to 0.52 (about 0.50), and value of α for other systems were also about 0.50 confirmed that 'transition state' lies midway between dropping mercury electrode and solution interface". But no explanation exists for quasireversiblity as same as given for transition state. What values or results indicates quasireversiblity. It should be given for the readers.	

Reviewer Details:

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