



SDI Review Form 1.6

Journal Name:	Asian Journal of Chemical Sciences
Manuscript Number:	Ms_AJOCS_32244
Title of the Manuscript:	In Vitro Antioxidant Properties of Novel Schiff Base Complexes
Type of the Article	Original Research Article

General guideline for Peer Review process:

This journal's peer review policy states that **NO** manuscript should be rejected only on the basis of '**lack of Novelty**', provided the manuscript is scientifically robust and technically sound.

To know the complete guideline for Peer Review process, reviewers are requested to visit this link:

(<http://www.sciencedomain.org/page.php?id=sdi-general-editorial-policy#Peer-Review-Guideline>)



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PART 1: Review Comments

	Reviewer's comment	Author's comment (if agreed with reviewer, correct the manuscript and highlight that part in the manuscript. It is mandatory that authors should write his/her feedback here)
Compulsory REVISION comments	- Grammatical errors and wrong tenses through out the paper– need to be corrected by a professional English Editor. 'The ligand having moderate and all of the metal complexes were existed encouraging results which indicate the importance of Schiff base ligand a metal complexes'(Line 23/24) "Together, these results indicate a strong Schiff base ligand and their metal complexes ability to act against different forms of RS.' (Line 281) makes no sense. - The article ^{d2+oes} not describe any new synthesis but rather the anti-oxidant studies of a known Schiff base and its metal complexes. Although known, some characterisation data of the synthesised complexes should be presented in order to determine successful formation, eg: CHNS elemental analysis, melting points, IR/UV-Vis spectroscopic data etc. - There is no correlation to data of similar complexes. All of the complexes had data that was not as good as the standards used. This needs to be explained more scientifically. As there was an upward trend when concentrations increased, why weren't the experiments continued until no	Grammatical and tenses are corrected throughout the paper. The sentences are revised and the different RS are added i.e DPPH, Fe^{+2}, Fe^{+3}, Mo^{+6} and OH. In this article we discuss only the antioxidant properties. The characterization of these complexes were given in the second already published article part of these research. Reference is given here. The difference are given in the discussion section of this article. that Cu^{2+} complex shows relative high antioxidant potential from rest of metal complexes due to different of their Reduction potential. You are right concentrations increased are not held continuous but we start from very low



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	upward trend was observed?	concentration. So the high concentration is not obtained.
<u>Minor</u> REVISION comments		
<u>Optional/General</u> comments	1. Please check if the references are complete. Some have missing page numbers.	Reference are standardize according to the journal format.